Introduction

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Organometallics. 3rd Ed. C. Elschenbroich Copyright © 2006 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim ISBN: 3-527-29390-6

1 Milestones in Organometallic Chemistry

1760 The cradle of organometallic chemistry is a Paris military pharmacy. It is there that Cadet works on invisible inks based on cobalt salt solutions. For their preparation, he uses cobalt minerals that contain arsenic.

 $\begin{array}{l} \operatorname{As_2O_3} + 4 \operatorname{CH_3COOK} \longrightarrow ``Cadet's \ fuming \ liquid'' \\ & \ contains \ cacodyloxide \ [(CH_3)_2As]_2O \\ & (\kappa \alpha \kappa \omega \delta \eta \varsigma = malodorous) \\ & \ first \ organometallic \ compound \end{array}$

- 1827 Zeise's salt, Na[PtCl₃C₂H₄]: the first olefin complex
- 1840 *R. W. Bunsen* continues the study of cacodyl compounds, which he names "alkarsines". The weakness of the As–As bond in molecules of the type R₂As–AsR₂ leads to a profusion of derivatives such as (CH₃)₂AsCN, whose taste (!) is checked by *Bunsen*.
- 1849 *E. Frankland*, a student of *Bunsen* at Marburg, attempts the preparation of an "ethyl radical" (cacodyl was also taken to be a radical).

 $3C_{2}H_{5}I + 3Zn \longrightarrow (C_{2}H_{5})_{2}Zn \text{ (a pyrophoric liquid)} + C_{2}H_{5}ZnI \text{ (solid)} + ZnI_{2}$

Frankland is admirably skilled in the manipulation of air-sensitive compounds. He uses hydrogen gas as a protective atmosphere!

1852 Frankland prepares the important alkylmercury halides:

 $CH_3I + Hg \xrightarrow{sunlight} CH_3HgI$

additionally: (C₂H₅)₄Sn, (CH₃)₃B (1860).

In the following years, **alkyl-transfer reactions** with R_2Hg and R_2Zn serve in the synthesis of numerous main-group organometallic compounds.

Frankland also introduced the concept of valency ("combining power") and the term "organometallic".

- 1852 C. J. Löwig and M. E. Schweizer in Zürich first prepare $(C_2H_5)_4Pb$ from ethyl iodide and Na/Pb alloy. In a similar manner, they also obtain $(C_2H_5)_3Sb$ and $(C_2H_5)_3Bi$.
- 1859 W. Hallwachs and A. Schafarik generate alkylaluminum iodides:

 $2AI + 3RI \longrightarrow R_2AII + RAII_2$

1863 C. Friedel and J. M. Crafts prepare organochlorosilanes:

 $SiCl_4 + m/2 ZnR_2 \longrightarrow R_m SiCl_{4-m} + m/2 ZnCl_2$

1866 *J. A. Wanklyn* develops a method for the synthesis of halide-free alkylmagnesium compounds:

 $(C_2H_5)_2Hg + Mg \longrightarrow (C_2H_5)_2Mg + Hg$

- 1868 M. P. Schützenberger obtains [Pt(CO)Cl₂]₂, the first metal-carbonyl complex.
- 1871 D. I. Mendeleev uses organometallic compounds as test cases for his periodic table. *Example*:

Known:	Predicted:	Found:
$Si(C_2H_5)_4$	Eka-Si(C ₂ H ₅) ₄	Ge(C ₂ H ₅) ₄ (C.Winkler, 1887)
	d = 0.96	d = 0.99
$Sn(C_2H_5)_4$	bp: 160 °C	bp: 163.5 °C

- 1890 L. Mond: Ni(CO)₄, first binary metal carbonyl, used in a commercial process for refining nickel. Mond is the founder of the English company ICI (Imperial Chemical Industries) as well as a renowned collector and patron of the arts.
- 1899 P. Barbier replaces Zn by Mg in reactions with alkyl iodides:

$$\begin{array}{c} H_{3}C \\ H_{3}C \\ H_{3}C \end{array} \xrightarrow{C=CH-CH_{2}-C} CH_{3} \xrightarrow{1.CH_{3}I, Mg} \\ CH_{3} \xrightarrow{H_{2}O} H_{3}C \\ H_{3}C \\ C=CH-CH_{2}-C \\ H_{3}C \\ CH_{3} \\ H_{3}C \\ C=CH-CH_{2}-C \\ H_{3}C \\ CH_{3} \\ H_{3}C \\ C=CH-CH_{2}-C \\ H_{3}C \\ H_{3}C \\ CH_{3} \\ H_{3}C \\ C=CH-CH_{2}-C \\ H_{3}C \\ H_{3}C \\ CH_{3} \\ H_{3}C \\ C=CH-CH_{2}-C \\ H_{3}C \\ H_{3}C \\ CH_{3} \\ H_{3}C \\ C=CH-CH_{2}-C \\ H_{3}C \\ H_{3}C \\ CH_{3} \\ H_{3}C \\ C=CH-CH_{2}-C \\ H_{3} \\ H_{3}C \\ C=CH-CH_{2}-C \\ H_{3} \\ H_{3}C \\ H_{3}C \\ C=CH-CH_{2}-C \\ H_{3} \\ H_{3$$

The reaction is explored in more detail by *Barbier*'s student *V. Grignard* (Nobel Prize 1912 shared with *P. Sabatier*). Although less sensitive than ZnR₂, RMgX is a more potent alkyl-group-transfer reagent.

1901 L. F. S. Kipping prepares $(C_6H_5)_2$ SiO, suspects its high molecularity, yet calls the material **diphenylsilicone**.

- 1909 *W. J. Pope*: formation of (CH₃)₃PtI, the **first σ-organotransition-metal compound.**
- 1909 *P. Ehrlich* (developer of chemotherapy, Nobel Prize 1908) introduces Salvarsan for the treatment of syphilis.
- 1917 W. Schlenk: alkyllithium reagents through transalkylation.

 $\begin{array}{l} 2\,\text{Li} + \text{R}_2\text{Hg} \longrightarrow 2\,\text{LiR} + \text{Hg} \\ 2\,\text{EtLi} + \text{Me}_2\text{Hg} \longrightarrow 2\,\text{MeLi} + \text{Et}_2\text{Hg} \end{array}$

- 1919 *F. Hein* synthesizes polyphenylchromium compounds, now known to be sandwich complexes, from CrCl₃ and PhMgBr.
- 1922 *T. Midgley* and *T. A. Boyd* introduce $Pb(C_2H_5)_4$ as an antiknock additive in gasoline.
- 1927 A. Job and A. Cassal prepare Cr(CO)₆.
- 1928 *W. Hieber* inaugurates his systematic study of metal carbonyls:

 $\begin{array}{l} Fe(CO)_5 + H_2NCH_2CH_2NH_2 \longrightarrow (H_2NCH_2CH_2NH_2)Fe(CO)_3 + 2 CO \\ Fe(CO)_5 + X_2 \longrightarrow Fe(CO)_4X_2 + CO \end{array}$

- 1929 *F. A. Paneth* generates alkyl radicals through pyrolysis of PbR₄; radical identification by means of their ability to cause the transport of a metallic mirror. *Paneth* thus reaches the goal set by *Frankland* in 1849.
- 1930 *K. Ziegler* encourages more extensive use of organolithium compounds in synthesis by developing a more simple preparation:

PhCH₂OMe + 2 Li \longrightarrow PhCH₂Li + MeOLi (ether cleavage) H. Gilman: RX + 2 Li \longrightarrow RLi + LiX (procedure used today)

- 1931 *W. Hieber* prepares Fe(CO)₄H₂, the first transition-metal-hydride complex.
- 1935 L. Pauling provides a valence-bond description of the bonding in Ni(CO)₄.
- 1938 O. Roelen discovers hydroformylation (the oxo process).
- 1939 *W. Reppe* starts work on the transition-metal catalyzed reactions of acetylenes.
- E. G. Rochow: 2 CH₃Cl + Si Cu cat., 300 °C (CH₃)₂SiCl₂ + ...
 This "direct synthesis" triggers the large-scale production and use of silicones. Preliminary work by *R. Müller* (Radebeul, near Dresden) was interrupted by World War II.

- 1 Milestones in Organometallic Chemistry
 - 1951 *M. J. S. Dewar* proposes a bond theory for complexes of alkenes with transition metals (elaborated on by *J. Chatt* and *L. A. Duncanson*, 1953).
 - 1951 *P. Pauson* (UK) and S. A. Miller (USA) obtain ferrocene, $(C_5H_5)_2Fe$, the first sandwich complex.
 - 1952 *H. Gilman* prepares LiCu(CH₃)₂, thereby establishing a now synthetically important class of compounds, the **organocuprates**.
 - 1953 *G. Wittig* develops a new synthesis of olefins from phosphonium ylides and carbonyl compounds (Nobel Prize 1979).
 - 1955 E. O. Fischer: rational synthesis of **bis(benzene)chromium**, $(C_6H_6)_2Cr$.
 - 1955 *K. Ziegler, G. Natta*: polyolefins from ethylene or propylene in a lowpressure process employing mixed metal (transition-metal halide/AlR₃) catalysts (Nobel Prize 1963).
 - 1956 H. C. Brown: hydroboration (Nobel Prize 1979).
 - 1959 *J. Smidt, W. Hafner:* preparation of $[(C_3H_5)PdCl]_2$, installation of the field of π -allyl-transition-metal complexes.
 - 1959 R. Criegee: stabilization of cyclobutadiene by complexation in [(C₄Me₄)NiCl₂]₂, thereby verifying a prediction by H. C. Longuet-Higgins and L. Orgel (1956).
 - 1960 *M. F. Hawthorne* prepares the icosahedral *closo*-borane dianion $[B_{12}H_{12}]^{2-}$, predicted by *H. C. Longuet-Higgins* (1955).
 - 1961 D. Crowfood Hodgkins: Based on X-ray crystal-structure analysis, vitamin B₁₂ coenzyme contains a Co–C bond (Nobel Prize 1964).
 - 1963 USA: Reports of the dicarba-*closo*-borane C₂B₁₀H₁₂ are issued by several industrial laboratories.
 - 1963 L. Vaska: trans-(PPh₃)₂Ir(CO)Cl reversibly binds O₂.
 - 1964 E. O. Fischer: (CO)₅WC(OMe)Me, the first carbene complex.
 - 1965 G. Wilkinson, R. S. Coffey: (PPh₃)₃RhCl acts as a homogeneous catalyst in the hydrogenation of alkenes.
 - 1965 *R. Petit*: Synthesis of (C₄H₄)Fe(CO)₃, stabilization of the antiaromatic cyclobutadiene through complexation.

- 1965 J. Tsuji discovers the first Pd-mediated C-C coupling.
- 1967 *G. Wilkinson* stabilizes the highly reactive carbon monosulfide in the rhodium complex (Ph₃P)₂Rh(Cl)CS.
- 1968 A. Streitwieser: preparation of uranocene, $U(C_8H_8)_2$.
- 1969 *P. L. Timms*: synthesis of organotransition-metal complexes by means of metal-atom–ligand-vapor cocondensation.
- 1969 *A. E. Shilov* discovers the Pt^{II}-catalyzed H/D exchange of alkenes with solvent protons in homogeneous solution, thereby laying the foundation for the now flourishing field of **C–H activation**.
- 1970 *G. Wilkinson*: kinetically inert transition-metal alkyl compounds by blocking β -elimination.
- 1972 *R. F. Heck* discovers the palladium-catalyzed substitution of vinylic H atoms with aryl, benzyl, and styryl halides which he subsequently develops into one of the most important named reactions in organometallic chemistry.
- 1972 *H. Werner*: $[(C_5H_5)_3Ni_2]^+$, the first triple-decker sandwich complex.
- 1973 E. O. Fischer: I(CO)₄Cr(CR), the first carbyne complex.
- 1973 Nobel Prize to E. O. Fischer and G. Wilkinson.
- 1976 Nobel Prize to *W. N. Lipscomb*: theoretical and experimental clarification of the structure and bonding in boranes.
- 1976 *M. F. Lappert* opens the field of main-group-element **dimetallenes** with the synthesis of [(Me₃Si)₂CH]₂Sn=Sn[CH(SiMe₃)₂]₂.
- 1979 *H. Köpf* and *P. Köpf-Maier* discover the cancerostatic action of titanocene dichloride, (C₅H₅)₂TiCl₂.
- 1980 *H. Bock*: synthesis and studies of silabenzene C₅H₅SiH in the gas phase (matrix isolation: *G. Maier*, 1982).
- 1981 *R. West*: (Mes)₂Si=Si(Mes)₂, the first stable compound with a **silicon-silicon double bond**.
- 1981 Nobel Prize to *R. Hoffmann* and *K. Fukui*: semiempirical MO concepts in a unified discussion of the structure and reactivity of inorganic, organic, and organometallic molecules (**isolobal analogy**).

- 1 Milestones in Organometallic Chemistry
 - *G. Becker* synthesizes $tBu-C\equiv P$, the first compound with a carbon–phosphorus triple bond.
 - *R. G. Bergman*: intermolecular reactions of organotransition-metal compounds with alkanes (**C–H activation**).
 - *W. Kaminsky* and *H. Brintzinger* introduce the "chiral zirconocene dichloride/methyl alumoxane (MAO)" as a new generation of catalysts for the isotactic polymerization of propene.
 - *R. Noyori* develops the catalytic, enantioselective addition of organozinc reagents ZnR₂ to carbonyl compounds.
 - *P. Jutzi*: preparation of decamethylsilicocene, Cp*₂Si.
 - *H. Schnöckel* synthesizes AlCl(solv), which he uses in the development of the organometallic chemistry of monovalent aluminum, for example, $Cp*_4Al_4$ (1991).
 - *W. Uhl*: synthesis of anionic $[i-Bu_{12}Al_{12}]^{2-}$, an icosahedral *closo*-alane.
 - 1993 D. Milstein reports the insertion of Rh into a C-C bond (C-C activation).
 - 1994 S. Harder prepares the lightest metallocene, the lithocene anion $[\text{Li}(\text{C}_5\text{H}_5)_2]^-$.
 - *A. H. Zewail* studies M–M and M–CO bond cleavage in Mn_2CO_{10} in a molecular beam on the femtosecond timescale $(10^{-15} s)$ by means of a pulsed laser (Nobel Prize 1999).
 - *G. Kubas* synthesizes the first σ -complex of a silane and studies the tautomerism with the hydridosilyl form:

$$\begin{array}{c} \mathbf{H} \\ \mathbf{L}_{n}\mathbf{M}\cdots| \\ \mathbf{SiMe_{3}} \end{array} \rightleftharpoons \mathbf{L}_{n}\mathbf{M} \\ \begin{array}{c} \mathbf{H} \\ \mathbf{SiMe_{3}} \end{array}$$

This observation contributes to the understanding of the mechanism of C–H activation.

- *P. P. Power* prepares the first germyne complex with a molybdenum–germanium triple bond.
- 1997 C. C. Cummins: the C atom as the ultimate ligand in an organometallic compound: [(R₂N)₃MoC]⁻, a "carbon complex".

- 1997 *G. M. Robinson* synthesizes the salt Na₂[ArGaGaAr] and postulates a gallium–gallium triple bond for the diaryldigallyne anion. (Extreme example of the steric protection of a labile structural element!)
- 1999 *W. Ho* monitors the dehydrogenation of single ethylene molecules on a Ni(110) surface by means of scanning tunneling microscopy (STM) and inelastic electron tunneling spectroscopy (IETS).
- 2001 Nobel Prize to *K. B Sharpless, W. S. Knowles*, and *R. Noyori* for pioneering work in the field of enantioselective catalysis.
- 2004 *E. Carmona* reports on decamethyldizincocene Cp*Zn–ZnCp*, the first molecule with an unsupported Zn^I–Zn^I bond.
- 2005 *A. Sekiguchi* fully characterizes R–Si≡Si–R, the first compound with a **silicon–silicon triple bond.**
- 2005 Nobel Prize to Y. Chauvin, R. R. Schrock, and R. H. Grubbs for mechanistic and applications-oriented studies on catalysts active in **olefin metathesis**.