

Contents

Preface *xi*

- 1 Application of Mössbauer Spectroscopy to Energy Materials** *1*
Pierre-Emmanuel Lippens, Jean-Claude Jumas, and Josette Olivier-Fourcade
- 1.1 Introduction *1*
- 1.2 Mössbauer Spectroscopy for Li-ion and Na-ion Batteries *2*
- 1.2.1 Characterization of Electrode Materials and Electrochemical Reactions *2*
- 1.2.2 Tin-Based Negative Electrode Materials for Li-ion Batteries *3*
- 1.2.2.1 Electrochemical Reactions of Lithium with Tin *3*
- 1.2.2.2 Tin Oxides *7*
- 1.2.2.3 Tin Borophosphates *10*
- 1.2.2.4 Tin-Based Intermetallics *13*
- 1.2.3 Iron-Based Electrode Materials *17*
- 1.2.3.1 LiFePO₄ as Positive Electrode Material for Li-ion Batteries *17*
- 1.2.3.2 Fe_{1.19}PO₄(OH)_{0.57}(H₂O)_{0.43}/C as Positive Electrode Material for Li-ion Batteries *18*
- 1.2.3.3 Na_{1.5}Fe_{0.5}Ti_{1.5}(PO₄)₃/C as Electrode Material for Na-ion Batteries *19*
- 1.3 Mössbauer Spectroscopy of Tin-Based Catalysts *21*
- 1.3.1 Reforming Catalysis *21*
- 1.3.2 Redox Properties of Pt-Sn Based Catalysts *22*
- 1.3.3 Trimetallic Pt-Sn-In Based Catalysts *24*
- 1.4 Conclusion *26*
- Acknowledgments *27*
- References *27*
- 2 Mössbauer Spectral Studies of Iron Phosphate Containing Minerals and Compounds** *33*
Gary J. Long and Fernande Grandjean
- 2.1 Introduction *33*
- 2.2 Thermodynamic Properties of Iron Phosphate Containing Compounds *34*

2.3	Room Temperature Mössbauer Spectra of Iron Phosphate Containing Minerals	37
2.4	Analysis of Magnetically Ordered Mössbauer Spectra	50
2.5	Structural and Thermodynamic Properties of the Polymorphs of FePO_4	53
2.5.1	Polymorphs of FePO_4	53
2.6	Mössbauer Spectra of $\alpha\text{-FePO}_4$	55
2.7	Magnetic Structure of $\alpha\text{-FePO}_4$, Obtained by Mössbauer Spectroscopy	57
2.7.1	Magnetic Structure of $\alpha\text{-FePO}_4$	57
2.8	Temperature Dependence of the $\alpha\text{-FePO}_4$ Structure Tilt Angle	60
2.9	Mössbauer Spectral Studies on Metastable Polymorphs of FePO_4	62
2.9.1	Crystallographic Structures of Two Polymorphs of $\text{FePO}_4 \cdot 2\text{H}_2\text{O}$	62
2.9.2	Preparation and Crystallographic Structures of the Two Polymorphs, $\gamma\text{-FePO}_4$ and $\zeta\text{-FePO}_4$	62
2.9.3	Mössbauer Spectral Studies of FePO_4 Metastable Polymorphs	64
2.9.4	Preparation and Mössbauer Spectra of Synthetic Heterosite, $(\text{Fe,Mn})\text{PO}_4$	67
2.9.5	Fits of the Magnetic Mössbauer Spectra of $\eta\text{-Fe}_{0.9}\text{Mn}_{0.1}\text{PO}_4$	68
2.10	Mössbauer Spectral Studies of Various Iron Phosphate Compounds	73
2.10.1	Mössbauer Spectral Properties of $\alpha\text{-Fe}_2(\text{PO}_4)\text{O}$	74
2.10.2	Mössbauer Spectral Properties of $\text{Fe}_3(\text{PO}_4)\text{O}_3$	79
2.10.3	Preparation and Structural Properties of $\text{Fe}_9(\text{PO}_4)\text{O}_8$	80
2.10.4	Mössbauer Spectral Properties of $\text{Fe}_9(\text{PO}_4)\text{O}_8$	81
	Acknowledgments	85
	References and Notes	85
3	Mössbauer Spectroscopic Investigation of Fe-Based Silicides	93
	<i>Xiao Chen, Junhu Wang, and Changhai Liang</i>	
3.1	Introduction	93
3.2	Mössbauer Spectroscopic Investigation of Iron Silicides Prepared By Mechanical Alloying and Heat Treatment	95
3.3	Mössbauer Spectra of Iron Silicide on Silica Prepared by Pyrolysis of Ferrocene-Polydimethylsilane Composites	99
3.4	Synthesis and Mössbauer Spectra of Iron Silicides by Temperature-Programmed Silicification	102
3.5	Mössbauer Spectroscopic Investigation of Doped Iron Silicides	104
3.6	Conclusion and Perspective	107
	References	108
4	Mössbauer Spectroscopy of Catalysts	113
	<i>Károly Lázár</i>	
4.1	Introduction	113
4.2	Principles of the Mössbauer Effect and Outlook of Its Application for Catalyst Studies	116

4.2.1	Brief Overview of the Basics of Mössbauer Spectroscopy	116
4.2.2	Mössbauer Spectroscopy from the Point of View of Catalyst Studies – Particular Features	117
4.2.3	The Probability of the Mössbauer Effect – f-Factor and Size Effects	118
4.2.4	Variants of the Technique	120
4.2.4.1	⁵⁷ Co Emission Spectroscopy	120
4.2.4.2	Synchrotron-Based NFS (Nuclear Forward Scattering)	122
4.2.4.3	Conversion Electron Mössbauer Spectroscopy	122
4.2.5	Technical Implementations – Experimental Conditions	123
4.3	Heterogeneous Catalysts	124
4.3.1	Sites on Supported Particles with Different Participation in Catalytic Processes	124
4.3.2	Collective Effects in Particles (Magnetism)	125
4.3.3	Case Studies	126
4.3.3.1	Metals and Alloys	126
4.3.3.2	Oxide Catalysts	130
4.3.3.3	Catalysts with Fe–N, Fe–C, and Fe–N–C Centers	133
4.4	Biocatalysts – Enzymes	135
4.5	Homogeneous Catalysts – Frozen Solutions	135
4.5.1	Studies on Reaction Intermediates – Time-Resolved Freeze-Quenched Spectra	136
4.6	Conclusions	137
	Acknowledgment	137
	References	138
5	Application of Mössbauer Spectroscopy in Studying Catalysts for CO Oxidation and Preferential Oxidation of CO in H₂	145
	<i>Kuo Liu, Junhu Wang, and Tao Zhang</i>	
5.1	Introduction	145
5.2	Application of Mössbauer Spectroscopy in CO Oxidation	147
5.2.1	⁵⁷ Fe Mössbauer Spectroscopy	147
5.2.2	¹¹⁹ Sn Mössbauer Spectroscopy	150
5.2.3	¹⁹⁷ Au Mössbauer Spectroscopy	151
5.2.4	¹⁹³ Ir Mössbauer spectroscopy	152
5.3	Application of Mössbauer Spectroscopy in PROX	153
5.3.1	PtFe-Containing Catalysts	153
5.3.2	Au-Based Catalysts	155
5.3.3	IrFe-Containing Catalysts	158
5.3.3.1	Porous Carbon Supported IrFe Catalysts	158
5.3.3.2	SiO ₂ and Al ₂ O ₃ Supported IrFe Catalysts	159
5.3.4	CuO/CeO ₂ with Fe ₂ O ₃ Additive	165
5.4	Concluding Remarks	165
	Acknowledgments	166
	References	166

6	Application of ^{57}Fe Mössbauer Spectroscopy in Studying Fe–N–C Catalysts for Oxygen Reduction Reaction in Proton Exchange Membrane Fuel Cells 171
	<i>Xinlong Xu, Junhu Wang, Suli Wang, and Gongquan Sun</i>
6.1	Introduction 171
6.2	Advanced ^{57}Fe Mössbauer Spectroscopy Technique 173
6.2.1	Room Temperature ^{57}Fe Mössbauer Spectroscopy 173
6.2.2	Low Temperature and Computational ^{57}Fe Mössbauer Spectroscopy 174
6.2.3	In Situ Electrochemical ^{57}Fe Mössbauer Spectroscopy 175
6.3	Characterization of Fe–N–C Using ^{57}Fe Mössbauer Spectroscopy 177
6.3.1	Identification of Active Sites 177
6.3.2	Investigation of Degradation Mechanism 180
6.3.3	Optimization for Synthesis of Fe–N–C 184
6.3.3.1	Precursor Composition 184
6.3.3.2	Heat Treatment 185
6.4	Summary and Perspective 187
	Acknowledgments 188
	References 188
7	^{197}Au Mössbauer Spectroscopy of Thiolate-protected Gold Clusters 195
	<i>Norimichi Kojima, Yasuhiro Kobaqyashi, and Makoto Seto</i>
7.1	Introduction 195
7.2	Synthesis of Thiolate Protected Gold Clusters 197
7.3	^{197}Au Mössbauer Spectroscopy of Gold Nano-clusters 198
7.3.1	Experimental Procedure of ^{197}Au Mössbauer Spectroscopy 198
7.3.2	^{197}Au Mössbauer Spectra of $\text{Au}_n(\text{SG})_m$ ($n = 10\sim 55$) 198
7.3.3	Molecular Structure and ^{197}Au Mössbauer Spectra of $\text{Au}_{10}(\text{SG})_{10}$ 198
7.3.4	Molecular Structure and ^{197}Au Mössbauer Spectra of $\text{Au}_{25}(\text{SG})_{18}$ 200
7.3.5	Structural Evolution of $\text{Au}_n(\text{SG})_m$ ($n = 10\sim 55$) Based on ^{197}Au Mössbauer Spectroscopy 201
7.3.6	^{197}Au Mössbauer Spectra of $\text{Au}_{24}\text{Pd}_1(\text{SC}_{12}\text{H}_{25})_{18}$ 204
7.3.7	^{197}Au Mössbauer Spectra of $\text{Au}_n(\text{SC}_{12}\text{H}_{25})_m$ 205
7.4	Conclusion 208
	Acknowledgments 208
	References 209
8	^{197}Au Mössbauer Spectroscopy of Gold Mixed-Valence Complexes, $\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{Y}_4](\text{X}, \text{Y} = \text{Cl}, \text{Br}, \text{I})$ and $[\text{NH}_3(\text{CH}_2)_n\text{NH}_3]_2[(\text{Au}^{\text{I}}\text{I}_2)(\text{Au}^{\text{III}}\text{I}_4)(\text{I}_3)_2]$ ($n = 7, 8$) 213
	<i>Norimichi Kojima, Yasuhiro Kobaqyashi, and Makoto Seto</i>
8.1	Introduction 213
8.2	Experimental Procedure 216
8.2.1	Synthesis and Characterization 216

8.2.1.1	$\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{Y}_4]$ (X, Y = Cl, Br, I)	216
8.2.1.2	$[\text{NH}_3(\text{CH}_2)_n\text{NH}_3]_2[(\text{Au}^{\text{I}}\text{I}_2)(\text{Au}^{\text{III}}\text{I}_4)(\text{I}_3)_2]$ (n = 7, 8)	217
8.2.2	^{197}Au Mössbauer Spectroscopy	217
8.3	Crystal Structure of $\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{Y}_4]$ (X, Y = Cl, Br, I)	218
8.4	Chemical Bond of Au–X in $[\text{Au}^{\text{I}}\text{X}_2]^-$ and $[\text{Au}^{\text{III}}\text{X}_4]^-$	221
8.5	Mössbauer Parameters of ^{197}Au in $[\text{Au}^{\text{I}}\text{X}_2]^-$ and $[\text{Au}^{\text{III}}\text{X}_4]^-$	223
8.5.1	Mössbauer Parameters of ^{197}Au in $(\text{C}_4\text{H}_9)_4\text{N}[\text{Au}^{\text{I}}\text{X}_2]$ and $(\text{C}_4\text{H}_9)_4\text{N}[\text{Au}^{\text{III}}\text{X}_4]$	224
8.5.1.1	Isomer Shift	224
8.5.1.2	Quadrupole Splitting	224
8.5.2	Mössbauer Parameters of ^{197}Au in $\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{X}_4]$ (X = Cl, Br, I)	225
8.5.2.1	Isomer Shift	225
8.5.2.2	Quadrupole Splitting	226
8.5.2.3	Analysis of ^{197}Au Mössbauer Parameters for $\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{X}_4]$	226
8.6	Charge Transfer Interaction in $\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{X}_4]$ (X = Cl, Br, I)	227
8.7	^{197}Au Mössbauer Spectra of $\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{Y}_4]$ (X, Y = Cl, Br, I)	228
8.7.1	Isomer Shift of Au^{I} in $\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{Y}_4]$	228
8.7.2	Isomer Shift of Au^{III} in $\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{Y}_4]$	230
8.7.3	Quadrupole Splitting of Au^{I} in $\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{Y}_4]$	230
8.7.4	Quadrupole Splitting of Au^{III} in $\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{Y}_4]$	231
8.8	Single Crystal ^{197}Au Mössbauer Spectra of $\text{Cs}_2[\text{Au}^{\text{I}}\text{I}_2][\text{Au}^{\text{III}}\text{I}_4]$	231
8.8.1	Comparison of ^{197}Au Mössbauer Spectra Between Single Crystal and Powder Crystal	231
8.8.2	Sign of EFG for Au^{I} in $[\text{Au}^{\text{I}}\text{I}_2]^-$ and Au^{III} in $[\text{Au}^{\text{III}}\text{X}_4]^-$	234
8.9	^{197}Au Mössbauer Spectra of $\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{X}_4]$ (X = Cl, I) Under High Pressures	235
8.9.1	Phase Diagram of $\text{Cs}_2[\text{Au}^{\text{I}}\text{X}_2][\text{Au}^{\text{III}}\text{X}_4]$ (X = Cl, Br, I)	235
8.9.2	Origin of Metallic Mixed-Valence State in $\text{Cs}_2[\text{Au}^{\text{I}}\text{Cl}_2][\text{Au}^{\text{III}}\text{Cl}_4]$	236
8.9.3	Au Valence Transition in $\text{Cs}_2[\text{Au}^{\text{I}}\text{I}_2][\text{Au}^{\text{III}}\text{I}_4]$	239
8.10	^{197}Au Mössbauer Spectra of $[\text{NH}_3(\text{CH}_2)_n\text{NH}_3]_2[(\text{Au}^{\text{I}}\text{I}_2)(\text{Au}^{\text{III}}\text{I}_4)(\text{I}_3)_2]$ (n = 7, 8)	241
8.11	Conclusion	243
	Acknowledgments	244
	References	245

9 Temperature- and Photo-Induced Spin-Crossover in Molecule-Based Magnets 251

Hiroko Tokoro, Kenta Imoto, and Shin-ichi Ohkoshi

9.1	Introduction	251
9.2	Spin-Crossover Phenomena in Cesium Iron Hexacyanidochromate Prussian Blue Analog	252
9.3	Light-Induced Spin-Crossover Magnet in Iron Octacyanidoniobate Bimetal Assembly	254

9.4	Chiral Photomagnetism and Light-Controllable Second Harmonic Light in Iron Octacyanidoniobate Bimetal Assembly	258
9.5	Conclusion and Perspective	265
	References	265
10	Developing a Methodology to Obtain New Photoswitchable Fe(II) Spin Crossover Complexes	271
	<i>Varun Kumar and Yann Garcia</i>	
10.1	Introduction and Context	271
10.2	Introduction to a New Photo-responsive Anion: <i>psca</i>	275
10.3	Combining Fe(II) and <i>psca</i> Together in a Single Compound	276
10.4	Fe(II) Mononuclear Complexes with DMPP and <i>psca</i> Ligands	278
10.5	1D Fe(II) Coordination Polymer with <i>psca</i> as Non-Coordinated Anions	281
10.6	Conclusions and Perspectives	284
	References	285
11	⁵⁷Fe Mössbauer Spectroscopy as a Prime Tool to Explore a New Family of Colorimetric Sensors	291
	<i>Li Sun, Weiyang Li, and Yann Garcia</i>	
11.1	Introduction and General Context	291
11.2	Colorimetric Gas Sensors Based on Fe(II) Complexes	292
11.3	Conclusions and Perspectives	306
	References	306
	Index	311