

FIORD errata list

(Typographical and other errors in **Fundamentals of Ionizing Radiation Dosimetry**, Wiley 2017)

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PAGE

- xxiii** - In the list of symbols for ‘**A, a**’ the first entry for A should be replaced by **atomic mass number or nucleon number (dimensionless)**. When an expression includes N_A/A , A is in g mol^{-1} , i.e., it corresponds to the molar mass M .
- xxviii** - In the list of symbols for ‘**M, m**’ the entry for M (3rd item) should be replaced by **molar mass, $M=A \text{ g mol}^{-1}$** .
- xxxv** - In the list of symbols for ‘**Lambda (Λ, λ)**’ a second definition for λ should be added, so that:
 λ decay constant of a radionuclide
photon wavelength
- 9** - End the top sentence with **multiplied by the molar mass constant $M_u (= 1 \text{ g mol}^{-1})$, i.e., the molar mass**.
 In the last sentence, the word ‘non-relativistic’ should be inserted so that it reads “...yield the following **non-relativistic** expressions for the particle kinetic...”
- 18** - In caption to Figure 1.5, the 2nd line, should read “(a) and (b) show the probabilities of a given emission...”.
- 20** - In Table 1.3 the c_0 and c_1 fitting coefficients for the L-shell fluorescence yield in the Z range 26-51 should be

Fluorescence yield	Range of Z	Fitting coefficient				
		c_0	c_1	c_2	c_3	c_4
ω_L	26-51	-9.2521×10^{-2}	8.7531×10^{-3}	-2.8087×10^{-4}	3.4823×10^{-6}	-

- 21** - The legend of Figure 1.6(b) should read “Mean fluorescence x-ray energies, \bar{k}_i (**dashed** lines), in the K, L1 and M1 shells; for comparison, the binding energies, $U_B(i)$ are also shown (**solid** lines)”.
- 28** - The solutions to exercise #3 are correct but the non-relativistic expression given in the *Solutions to Exercises* book (page 2) does not yield the solutions provided. The student can derive E from Eq. (1.27), to arrive at

$$E = m_0 c^2 \left(\frac{1}{\sqrt{1 - \beta^2}} - 1 \right) = m_0 c^2 \tau$$

which is also given as Eq. (2.95) in page 82.

- 28** - The solutions to exercise #5 are correct but the expression for the Birge ratio given in the *Solutions to Exercises* book (page 3) should be

$$R_{\text{Birge}} = \frac{s(\bar{x}_w)_{\text{ext}}}{s(\bar{x}_w)_{\text{int}}} = 2.2$$

- 47 - First line below Eq. (2.37), “atomic mass A ” should be replaced by “**atomic mass number A** ”.
- 64 - Three lines below Eq. (2.75), “ $\text{cm}^2 \text{g}^{-1}$ ” should be replaced by “ **$\text{cm}^2 \text{mol}^{-1}$** ”.
- 79 - Figure 2.22 and its legend should be replaced by

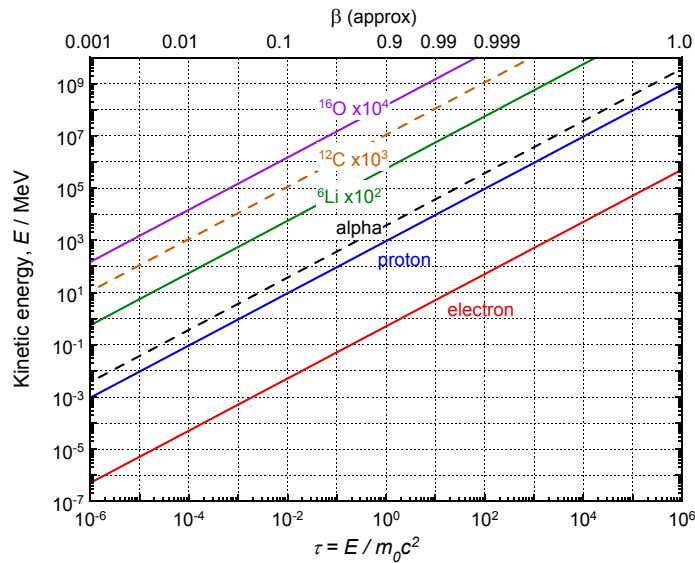


Figure 2.22. Graphical representation of the relation between a particle kinetic energy E (MeV) and its equivalent in terms of the particle rest energy, $\tau = E/m_0c^2$, for electrons and some heavy charged particles and ions. The upper abscissa provides the relation with the relativistic velocity, $\beta = v/c$. **For better visualization, the rest masses of ${}^6\text{Li}$, ${}^{12}\text{C}$ and ${}^{16}\text{O}$ have been multiplied by 10^2 , 10^3 and 10^4 , respectively.**

- 83 - Second paragraph, second line, “scaled by the atomic mass” should be replaced by “**scaled by the atomic mass number**”.
- 91 - In Section 2.4.12, line 7, “trasversed” should be replaced by “**traversed**”.
- 99 - In Figure 2.34, the labeling of the electronic stopping power curves (S_{el}/ρ) is incorrect: C should be the top curve and U the bottom curve, i.e.,

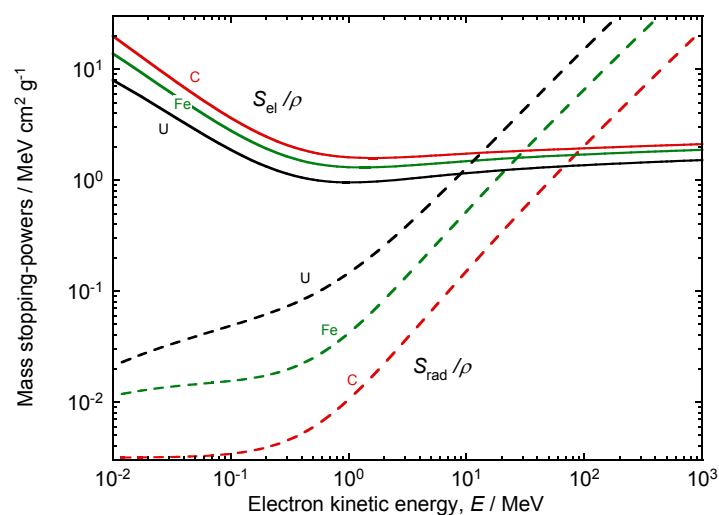


Figure 2.34. Mass radiative (dashed lines) and electronic (solid lines) stopping powers as a function of the kinetic energy of electrons in carbon ($Z = 6$), iron ($Z = 26$), and uranium ($Z = 92$). (Data from ICRU, 1984b).

- 104 - First sentence, “atomic weight” should be replaced by “**atomic mass number**”.

105 - In Figure 2.38, the label of the horizontal axis should just be “Kinetic energy”, as proton data are also included in the plot.

114 - Four lines after Eq. (2.157), “absorber atomic mass” should be replaced by “absorber atomic mass number”.

122 - At the bottom of the page, in the last step of Eq. (2.180), the Thomas-Fermi radius should be squared, i.e.,

$$\frac{4\pi}{q^2 + (1/R_{\text{TF}})^2}$$

136 - The left hand side of Eq. (2.247), should not have $d\mathcal{W}$, i.e., the equation should be $d\sigma = 2\pi b db$.

136 - Two and three lines above Section 2.11.3.3, item (iv) should read “is proportional to \mathcal{W}^{-2} (instead of \mathcal{W}^2)”.

156 - In Eqs. (3.31) – (3.34), the polarization angle ϕ should be replaced by φ , and the last term of Eq. (3.33) should have E_0^2 instead of E_0 , i.e.,

$$B_{\text{out}} = \frac{e}{4\pi\epsilon_0} \frac{\dot{v} \sin \varphi}{c^3 r} = \frac{E_{\text{out}}}{c} \quad (3.31)$$

$$S_{\text{out}} = \frac{E_{\text{out}} B_{\text{out}}}{\mu_0} = \frac{E_{\text{out}}^2}{c \mu_0} = \epsilon_0 c E_{\text{out}}^2 = \epsilon_0 c \left(\frac{e}{4\pi\epsilon_0 c^2} \frac{\dot{v} \sin \varphi}{r} \right)^2 \quad (3.32)$$

$$S_{\text{out}} = \epsilon_0 c \left(\frac{e}{4\pi\epsilon_0 c^2} \frac{\sin \varphi}{r} \right)^2 \left(\frac{e}{m_e} E_0 \sin \omega t \right)^2 = \epsilon_0 c \frac{r_e^2 E_0^2 \sin^2 \omega t \sin^2 \varphi}{r^2} \quad (3.33)$$

$$\bar{S}_{\text{out}} = \epsilon_0 c \overline{E_{\text{out}}^2} = \epsilon_0 c \frac{r_e^2 E_0^2 \overline{\sin^2 \omega t} \overline{\sin^2 \varphi}}{r^2} = \frac{1}{2} \epsilon_0 c r_e^2 E_0^2 \frac{\overline{\sin^2 \varphi}}{r^2} \quad (3.34)$$

160 - In Figure 3.9, the label of the y-axis should have a capital X, i.e., “Atomic form factor, $F(X, Z)$ ”.

167 - In line below Eq. (3.77), the reference should be to figure 3.16a instead of 3.15a, i.e., “and is shown in Figure 3.16a. The cross section...”.

169 - In Figure 3.16(b), the label of the y-axis should be “ $d\sigma_{c,\text{KN}}/d\phi$ / (mb electron⁻¹ rad⁻¹)”.

173 - In Eq. (3.95), $m_e c^2$ on both sides of the equation should be squared, i.e.,

$$E'_{\text{tot}}{}^2 - (m_e c^2)^2 = c^2 (q^2 + 2 \mathbf{q} \cdot \mathbf{p}_e) + E_{\text{tot}}^2 - (m_e c^2)^2 \quad (3.95)$$

180 - In Figure 3.25, the symbols E_+ and E_- should, for consistency with the text, be replaced by E^+ and E^- .

191 - In first line below the section header **3.8.2 Cross Section**, a space is missing between “Using” and “ $\epsilon = k/m_e c^2 \dots$ ”

195 - In first line after Eq. (3.156), “A is the atomic mass” should be replaced by “A is the atomic mass number multiplied by the molar mass constant M_u (1 g mol⁻¹)”.

196 - In Figure 3.33 lower panel (b), the horizontal axis number labels don’t match up with the grid, see upper panel (a).

197 - In Figure 3.34, the vertical axis label should be \bar{f}_{ph} , to be consistent with Eq. (3.161), instead of f_{p} .

213 - The second answer to exercise #15 should be corrected as:

$$\text{Answer: } n_{\text{av}} = \frac{\ln \frac{E_{\text{fin}}}{E_{\text{in}}}}{\ln \left[\frac{M^2 + m_n^2}{(M + m_n)^2} \right]}; n_{\text{min}} = \frac{\ln \frac{E_{\text{min,fin}}}{E_{\text{in}}}}{\ln \left[\frac{(M - m_n)^2}{(M + m_n)^2} \right]}$$

as there was an error in the expression for the minimum neutron kinetic energy. The solution (for the Exercises book) then becomes

$$E_{\text{min},f} = E_i - Q_{\text{max}} = E_i \left[\frac{(M - m_n)^2}{(M + m_n)^2} \right] \sim E_i [\text{m'-ratio}]$$

and for n_{min} interactions

$$E_{\text{min,fin}} = E_{\text{in}} \left[\frac{(M - m_n)^2}{(M + m_n)^2} \right]^{n_{\text{min}}}$$

from where the corrected expression for n_{min} is obtained.

249 - In Table 4.2 the photon attenuation values should be

Primary radiation energy (MeV)	Photon attenuation (%) over maximum secondary electron range	Neutron attenuation (%) over maximum secondary proton range
0.1	0.25	0.05
1.0	2.31	0.04
10	10.2	0.5
30	20.0	1.5

250 - In Figure 4.15, the scale of the vertical axis (4.0×10^{-14} to 1.0×10^{-13}) should be replaced by 4.0×10^{-12} to 1.0×10^{-11} .

258 - The answer to exercise #13 should be:

$$\text{Answer: } K = 7.65 \times 10^5 \text{ erg g}^{-1} = 76.5 \text{ Gy}; K_{\text{el}} = 5.58 \times 10^5 \text{ erg g}^{-1} = 55.8 \text{ Gy.}$$

as there was a typo in the erg-to-MeV conversion. The full solution (for the Exercises book) then becomes

(a) For kerma, $K = \Psi (\mu_{\text{tr}}/\rho)_{\text{Pb}}$, where

$$\Psi = 3.5 \times 10^6 \frac{\text{phot}}{\text{cm}^2 \text{ s}} \times 6.048 \times 10^5 \text{ s} \times \frac{6 \text{ MeV}}{\text{phot}} \times \frac{1.6022 \times 10^{-6} \text{ erg}}{\text{MeV}} = 2.0349 \times 10^7 \frac{\text{erg}}{\text{cm}^2}$$

and from the Data Tables

$$(\mu_{\text{tr}}/\rho)_{\text{Pb},6\text{MeV}} = 0.0376 \text{ cm}^2/\text{g}$$

therefore

$$K = 2.0349 \times 10^7 \frac{\text{erg}}{\text{cm}^2} \times 0.0376 \frac{\text{cm}^2}{\text{g}} = 7.6513 \times 10^5 \frac{\text{erg}}{\text{g}} = 76.51 \text{ Gy}$$

(b) For the electronic kerma, $K_{\text{el}} = \Psi (\mu_{\text{en}}/\rho)_{\text{Pb}}$

$$(\mu_{\text{en}}/\rho)_{\text{Pb},6\text{MeV}} = 0.0274 \text{ cm}^2/\text{g}$$

therefore

$$K_{\text{el}} = 2.0349 \times 10^7 \frac{\text{erg}}{\text{cm}^2} \times 0.0274 \frac{\text{cm}^2}{\text{g}} = 5.5757 \times 10^5 \frac{\text{erg}}{\text{g}} = 55.76 \text{ Gy}$$

- 263 - In line -7 from bottom, “*A* is the atomic weight” should be replaced by “*A* is the atomic mass number multiplied by the molar mass constant M_u (1 g mol^{-1})”.
- 317 - [NEW]
On Eq. (7.4): Note that, despite the use of the symbol K_{air} , the expression is written in terms of the air electronic kerma $(K_{\text{el}})_{\text{air}}$. The use of $(K_{\text{el}})_{\text{air}}$ is strictly correct because radiative interactions of secondary electrons do not contribute to local energy deposition. However, at photon energies below about 200 keV, radiative interactions in air are almost negligible ($1 - g_{\text{air}} > 0.999$), so that K_{air} and $(K_{\text{el}})_{\text{air}}$ are practically identical as are the corresponding photon coefficients $[\mu_{\text{tr}}(E)/\rho]_{\text{air}}$ and $[\mu_{\text{en}}(E)/\rho]_{\text{air}}$, see Eqs. (4.27) and (4.45).
- 323 - [NEW]
In Figure 7.5, the label of the Y-axis should be “Effective energy” instead of “Equivalent energy”. It should be emphasized, however, that both terms are used in the literature.
- 340 - In line above Eq. (7.23), “atomic mass *A*” should be replaced by “atomic mass number *A*”.
- 358 - Citation “Andreo, 1991;” above eq. (8.35) should be “Andreo, 1991a;”.
- 368 - In Figure 8.7 the caption cites Andreo (1981) but it should be “Andreo (1991a)”.
- 372 - At the end of the fourth paragraph the reference should be “Andreo (1988a)”.
- 417 - In Figure 9.11, the label of the x-axis should be “*y* / MV”.
- 524 - Six lines after eq. (12.35), the Boag *et al.* reference should be 1996, instead of 1966.
- 574 - [NEW]
In exercise #2. Add at the end of the statement “and the light path length is 1.0 cm”. The same change applies to p.117 of the Book of Exercises.
- 575 - [NEW]
In exercise #4. The cell length should be changed to 1.0 cm, consistent with p.119 of the Book of Exercises. Note that for a length of 1.5 cm the answer would be $314.3 \text{ J m}^{-2} \text{ s}^{-1}$.
- 575 - [NEW]
In the solution to exercise #8 in p.121 of the Book of Exercises, second line of the fluence, the denominator should be changed to read $[(0.3 \times 10^{-4})^2 \pi]^2$.
- 616 - In Figure 14.20, the right-y axis label should be “ $[S_{\text{el}}(E)/\rho]_{\text{med,w}}$ ”.
- 690 - [NEW]
In exercise #5. The three M_{field} values given in the statement should be changed to $M_{\text{field}}(+250\text{V}) = 8.632$, $M_{\text{field}}(-250\text{V}) = 8.690$, $M_{\text{field}}(+80\text{V}) = 8.569$.
The same changes apply to p.134 of the Book of Exercises.
- 754 - Second paragraph, in 3rd line “atomic mass (g)” should be replaced by “atomic mass number”; in 7th line delete “mass number or”. In the last sentence of footnote no. 4, “atomic mass *A*” should be replaced by “atomic mass m_a ”.
- 774 - In second line “(amu)” should be replaced by “(u)”. Replace “1/12 of” by “1/12 of” (i.e., “of” in roman font, not in italics).
- 842 - In Table A.1, under “Atomic mass constant”, delete the factor $\times 10^8$ for the unit of m_u in MeV, i.e., $m_u = 931.494\,061(21) \text{ MeV}$.
- 883 - Reference Andreo (1981) should be changed to “Andreo (1991a)”.