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The Asymmetric MSPV Reduction of Prochiral Ketones with *i*PrOH Catalyzed by Al Catalysts

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Supporting Information Available: Experimental procedure, quantitative analysis (including chiral GC traces), plots of conversion and ee vs. time for the reduction of α -bromoacetophenone, and the characterization of the active chiral aluminum alkoxide catalyst (9 pages). This material is available on the WWW under at <http://www.angewandte.com> or from the author.

General Information and Materials. Toluene was distilled over sodium/benzophenone. 2-propanol was distilled over Mg(O*i*Pr)₂. Propiophenone, *isovalerophenone*, *isobutyrophenone*, acetophenone, α -methoxyacetophenone, 4-methylacetophenone, 4-fluoroacetophenone, 4-chloroacetophenone, and 4-bromoacetophenone were dried over CaH₂. All solvents were distilled under nitrogen and saturated with nitrogen prior to use. Ligands **1** and **2** were synthesized according to literature procedure.^{1,2} Monomeric chiral (BINOL)AlMe complexes was synthesized following literature precedent.³ Pure samples of 1-(4-fluorophenyl)ethanol, 1-(4-chlorophenyl)ethanol, and 1-(4-bromophenyl)ethanol were prepared by LAH reduction. Pure samples of α -bromomethylbenzyl alcohol, α -methoxymethylbenzyl alcohol, and 1-phenyl-3-methylbutanol were prepared from the corresponding ketones according literature procedures.⁴ All other reagents were purchased from the Aldrich Chemical Company and used without further purification, unless otherwise noted.

GC analyses of reaction mixtures were carried out on a Hewlett Packard 5890A GC equipped with an FID detector and a HP3396A integrator. The column used was a 30-m HP-5 capillary column with 0.32-mm inner diameter and 0.25- μ m film thickness. Flow rate = 1.8 mL/min. GC yields of 1-phenylethanol, 1-phenyl-1-propanol, 2-chloro-1-phenylethanol, and 1-*p*-tolylethanol were determined through integration of the product peak against 1,2,4,5-tetramethylbenzene (internal standard) using pre-established response factors. All other yields of the alcohol were determined through the integration of the product peak against that of the starting material. For verification a scale-up MSPV reduction was carried out for α -bromoacetophenone and its isolated yield found to be consistent with GC data (*vide infra*).

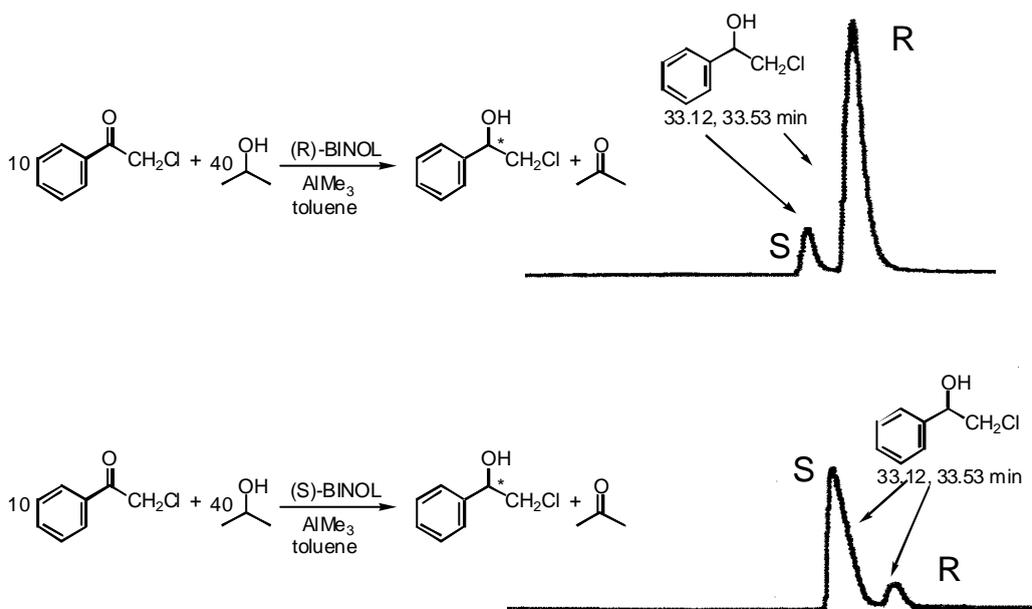
Chiral GC analysis (for Table 1, entries 1,2,5-8 and Table 2, entries 1-5 of manuscript) was performed on a Hewlett Packard 5890A GC equipped with an FID detector and a HP3390A integrator. The chiral column used was a 30-m Supelco β -DexTM 225 fused silica capillary column with 0.25-mm inner diameter and 0.25- μ m film thickness. Flow rate = 1.8 mL/min. Retention times for various components of the reaction mixture were assigned by the injection of a pure sample of each component in the reaction. All other chiral GC analysis (Table 1, entries 3 and 4 of manuscript) was performed on a Hewlett Packard 5890A equipped with a FID detector and an HP3390A integrator. The chiral column used was a 30-m Supelco β -DexTM 120 fused silica capillary column with 0.25-mm inner diameter and 0.25- μ m film thickness. Flow rate = 1.8 mL/min.

General MSPV Reduction Procedure. All reactions were carried out under a dry nitrogen atmosphere unless otherwise noted. Into a 4-mL-vial equipped with a magnetic stir bar was added toluene (500 μ L) and enantiopure BINOL (5.8 mg, 0.02 mmol). AlMe₃ (1.9 ml, 0.02 mmol) was added to the mixture via syringe

and the reaction was allowed to stir for 5 minutes when a white precipitate formed. The carbonyl substrate (10 equiv) and 2-propanol (40 or 150 equiv) were added and the vial was sealed with a teflon-lined silicone septa. The reaction was stirred at room temperature under nitrogen for 16 hours. Aliquots (20 μL) were passed through a plug of neutral aluminum oxide [activated, Brockmann activity 1, ~ 150 mesh] and analyzed on GC to determine selectivity and conversion data. The chiral GC traces of the MSPV reductions are provided on the following pages.

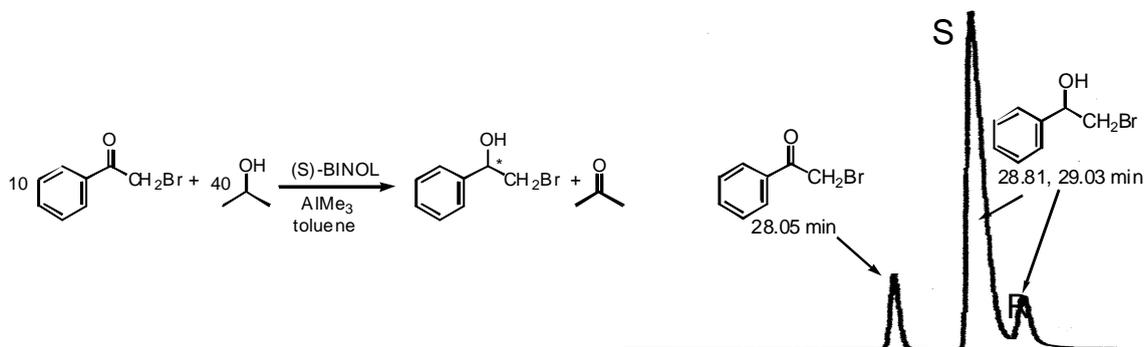
Scale-up Asymmetric MSPV Reduction. In the drybox and into a 20-mL-vial equipped with a magnetic stir bar was added toluene (2 mL) and (*S*)-(-)-BINOL (57.3 mg, 0.2 mmol). AlMe_3 (19.7 μL , 0.2 mmol) was added to the mixture via syringe and the reaction was allowed to stir for 5 minutes when a white precipitate formed. α -Bromoacetophenone (0.40 g, 2.0 mmol) and 2-propanol (610 μL , 8.0 mmol) were added and the vial was sealed with a teflon-lined cap, taken out of the drybox, and stirred at room temperature under nitrogen for 16 hours. Flash column chromatography using 230-400 mesh silica gel (purchased from Merck, column dimensions = 2.5 cm x 12 cm) and CH_2Cl_2 eluent, was done on the reaction and a pure sample of α -(bromomethyl)benzyl alcohol was isolated in 93% yield and 79% ee.

The chiral GC traces for the MSPV reduction of α -chloroacetophenone catalyzed by (*R*)- and (*S*)-BINOL and AlMe_3 (Table 1, entry 1).



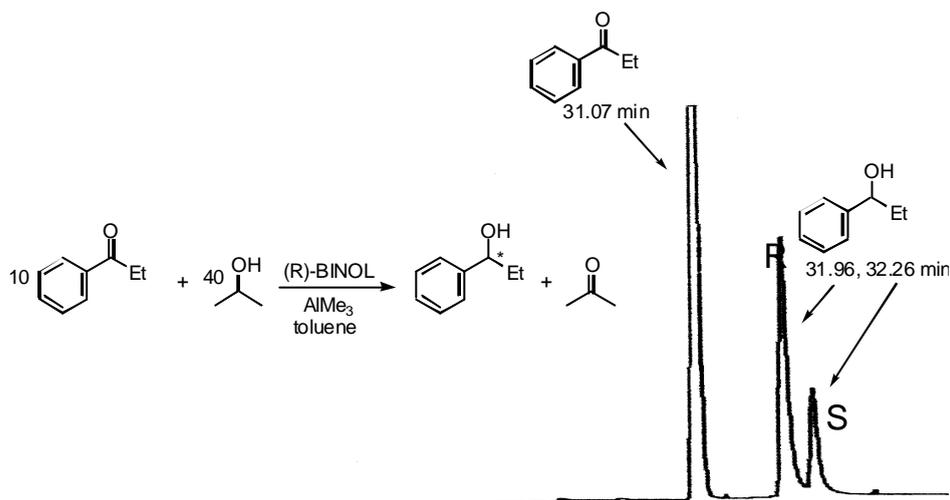
Temp. Program: initial temp. = 80 $^{\circ}\text{C}$, initial time = 0 min., ramp = 2 $^{\circ}\text{C}/\text{min.}$, final temp = 180 $^{\circ}\text{C}$, final time = 10 min.

The chiral GC trace (after 12h) for the MSPV reduction of **a**-bromoacetophenone catalyzed by (*S*)-BINOL and AlMe₃ (Table 1, entry 2).



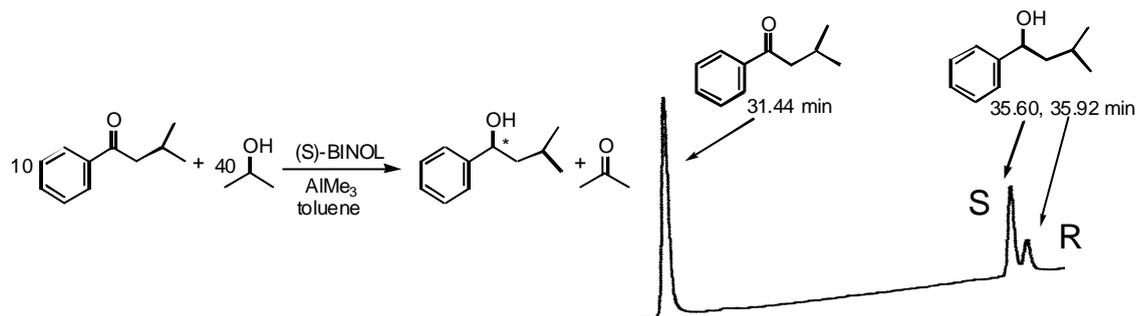
Temp. Program: initial temp. = 100 °C, initial time = 0 min., ramp = 2 °C/min., final temp = 180 °C, final time = 10 min.

The chiral GC trace for the MSPV reduction of propiophenone catalyzed by (*R*)-BINOL and AlMe₃ (Table 1, entry 3).



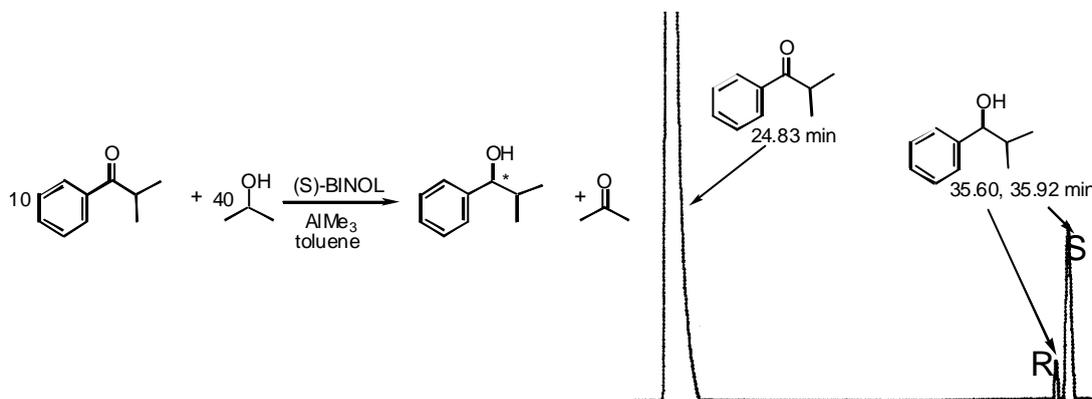
Temp. Program: initial temp. = 100 °C, initial time = 0 min., ramp = 2 °C/min., final temp = 180 °C, final time = 10 min.

The chiral GC trace for the MSPV reduction of *isovalerophenone* catalyzed by (S)-BINOL and AlMe₃ (Table 1, entry 4).



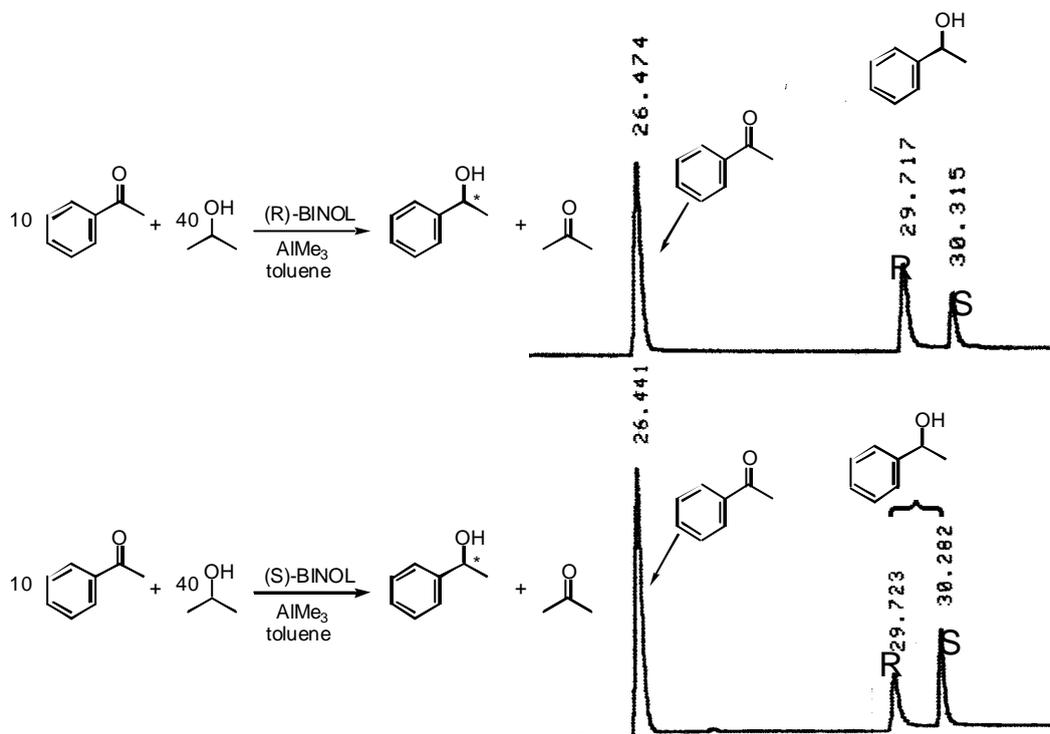
Temp. Program: initial temp. = 100 °C, initial time = 0 min., ramp = 2 °C/min., final temp = 180 °C, final time = 10 min.

The chiral GC trace for the MSPV reduction of *isobutyrophenone* catalyzed by (S)-BINOL and AlMe₃ (Table 1, entry 5).



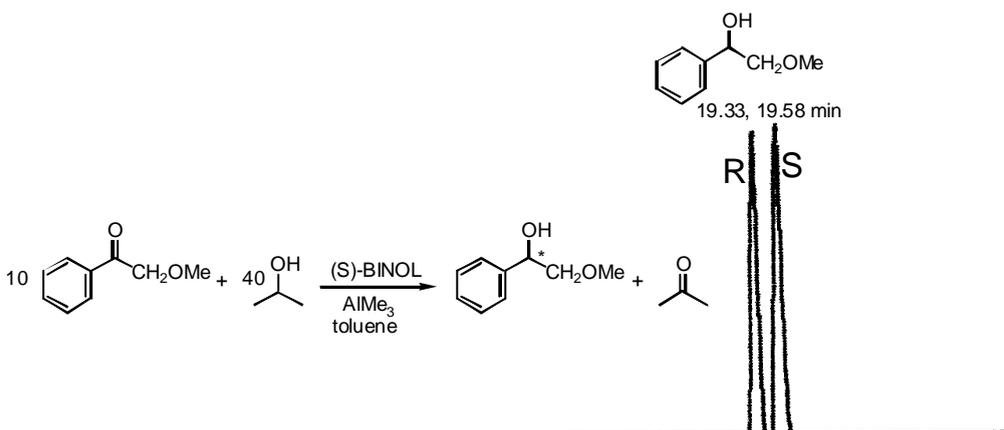
Temp. Program: initial temp. = 100 °C, initial time = 0 min., ramp = 2 °C/min., final temp = 180 °C, final time = 10 min.

The chiral GC trace for the MSPV reduction of acetophenone catalyzed by (R)- and (S)-BINOL and AlMe₃ (Table 1, entry 6).



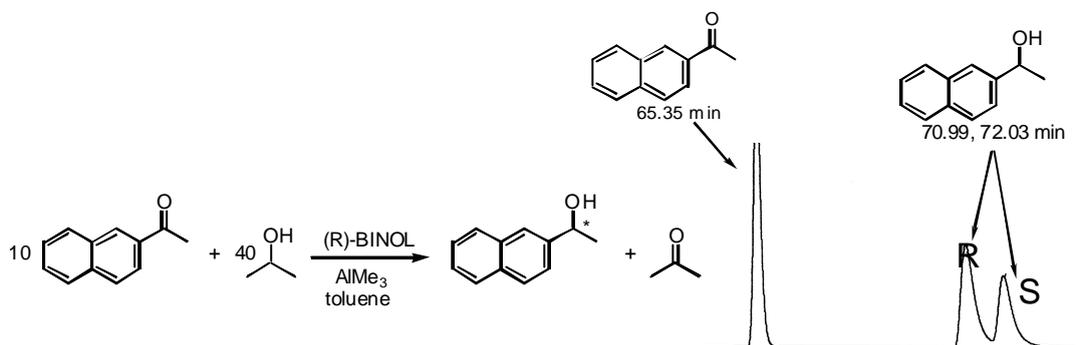
Temp. Program: initial temp. = 80 °C, initial time = 20 min., ramp = 5 °C/min., final temp = 180 °C, final time = 10 min.

The chiral GC trace for the MSPV reduction of *a*-methoxyacetophenone catalyzed by (S)-BINOL and AlMe₃ (Table 1, entry 7).



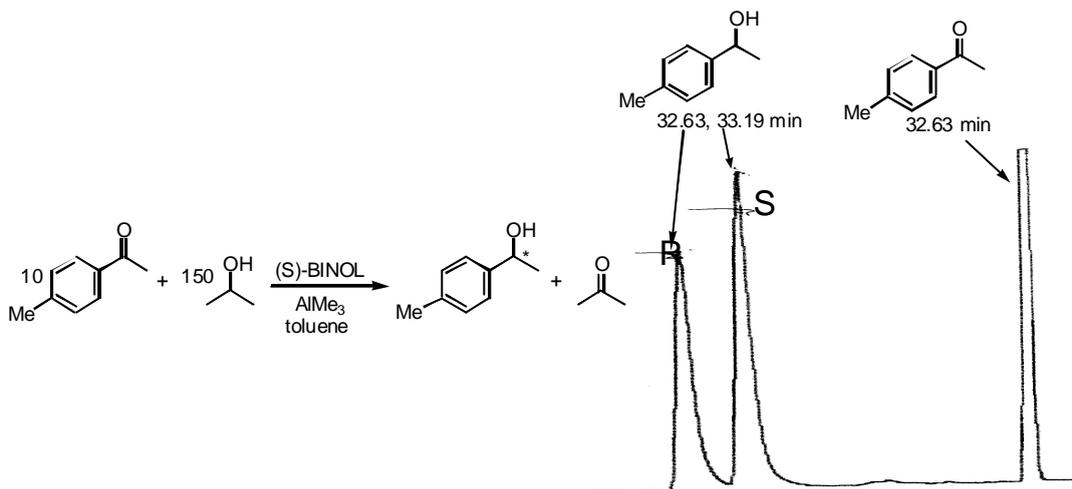
Temp. Program: initial temp. = 100 °C, initial time = 0 min., ramp = 2 °C/min., final temp = 180 °C, final time = 10 min.

The chiral GC trace for the MSPV reduction of acetophenone catalyzed by (R)-BINOL and AlMe₃ (Table 1, entry 8).



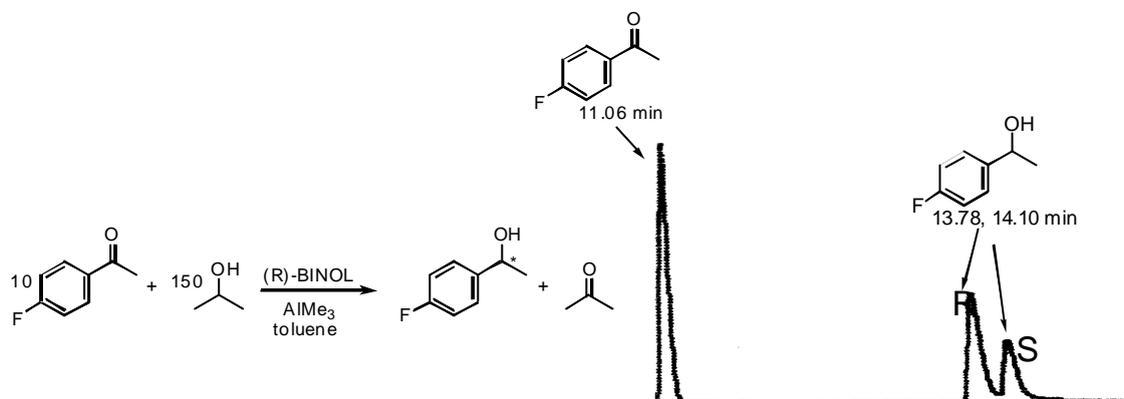
Temp. Program: initial temp. = 100 °C, initial time = 0 min., ramp = 1 °C/min., final temp = 180 °C, final time = 10 min.

The chiral GC trace for the MSPV reduction of 4-methylacetophenone catalyzed by (S)-BINOL and AlMe₃ (Table 2, entry 2).



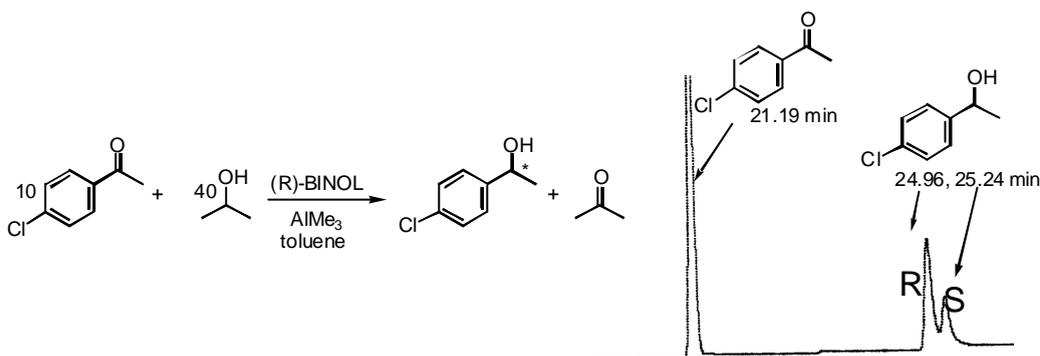
Temp. Program: initial temp. = 80 °C, initial time = 20 min., ramp = 5 °C/min., final temp = 180 °C, final time = 10 min.

The chiral GC trace for the MSPV reduction of 4-fluoroacetophenone catalyzed by (R)-BINOL and AlMe₃ (Table 2, entry 3).



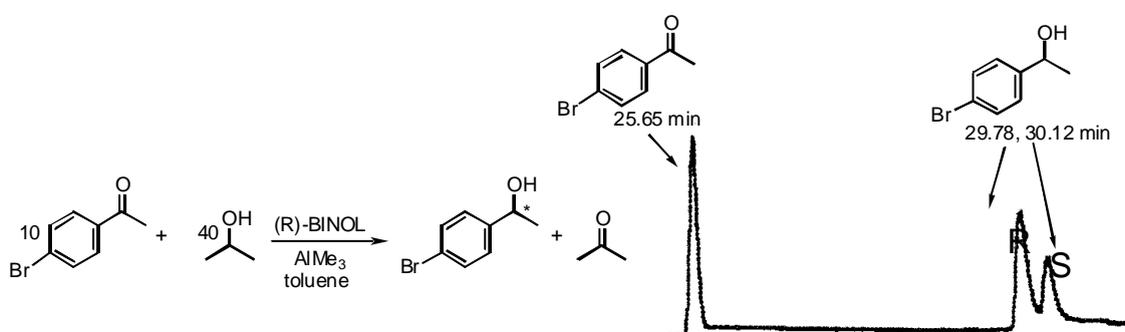
Temp. Program: initial temp. = 80 °C, initial time = 20 min., ramp = 5 °C/min., final temp = 180 °C, final time = 10 min.

The chiral GC trace for the MSPV reduction of 4-chloroacetophenone catalyzed by (R)-BINOL and AlMe₃ (Table 2, entry 4).



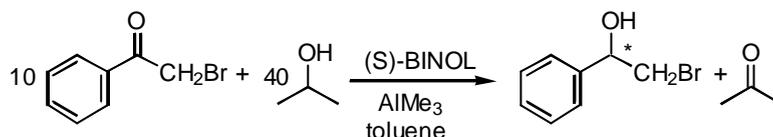
Temp. Program: initial temp. = 80 °C, initial time = 20 min., ramp = 5 °C/min., final temp = 180 °C, final time = 10 min.

The chiral GC trace for the MSPV reduction of 4-bromoacetophenone catalyzed by (R)-BINOL and AlMe₃ (Table 2, entry 5).

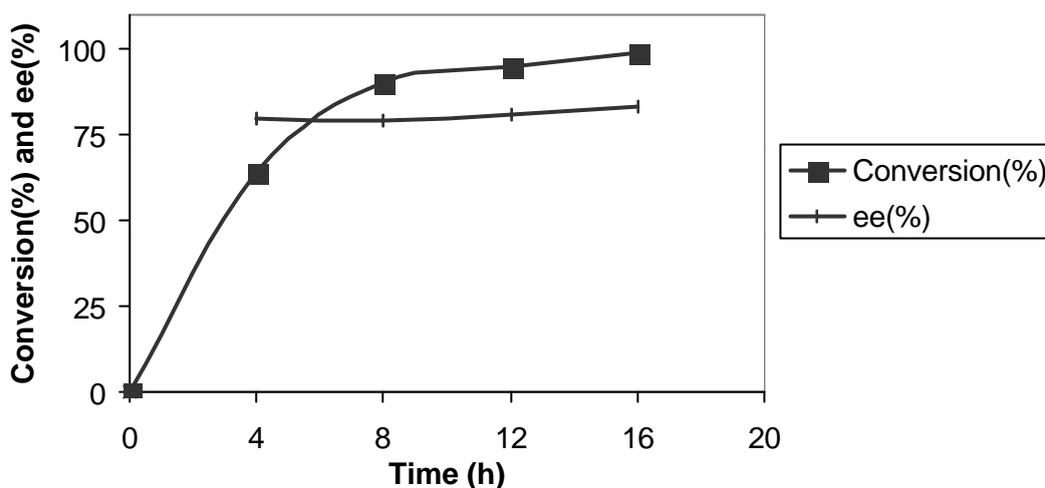


Temp. Program: initial temp. = 80 °C, initial time = 20 min., ramp = 5 °C/min., final temp = 180 °C, final time = 10 min.

The time evolution of the selectivity and conversion for the asymmetric MSPV reduction of α -bromoacetophenone.

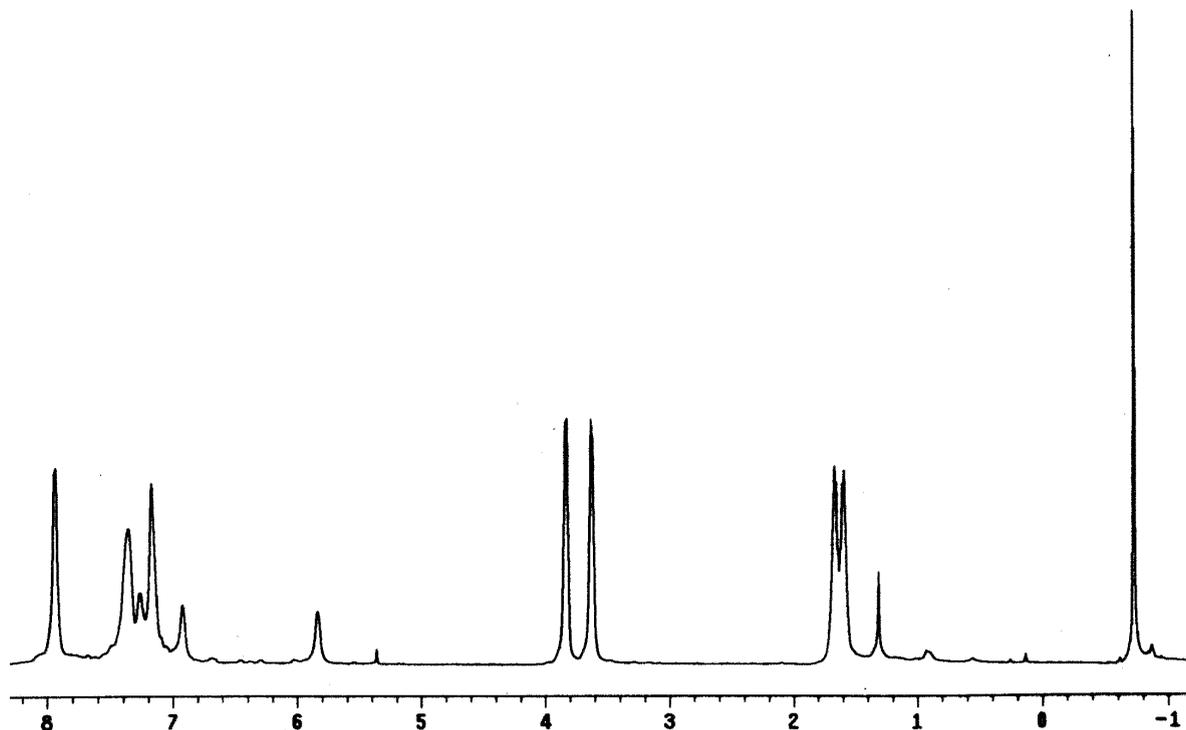
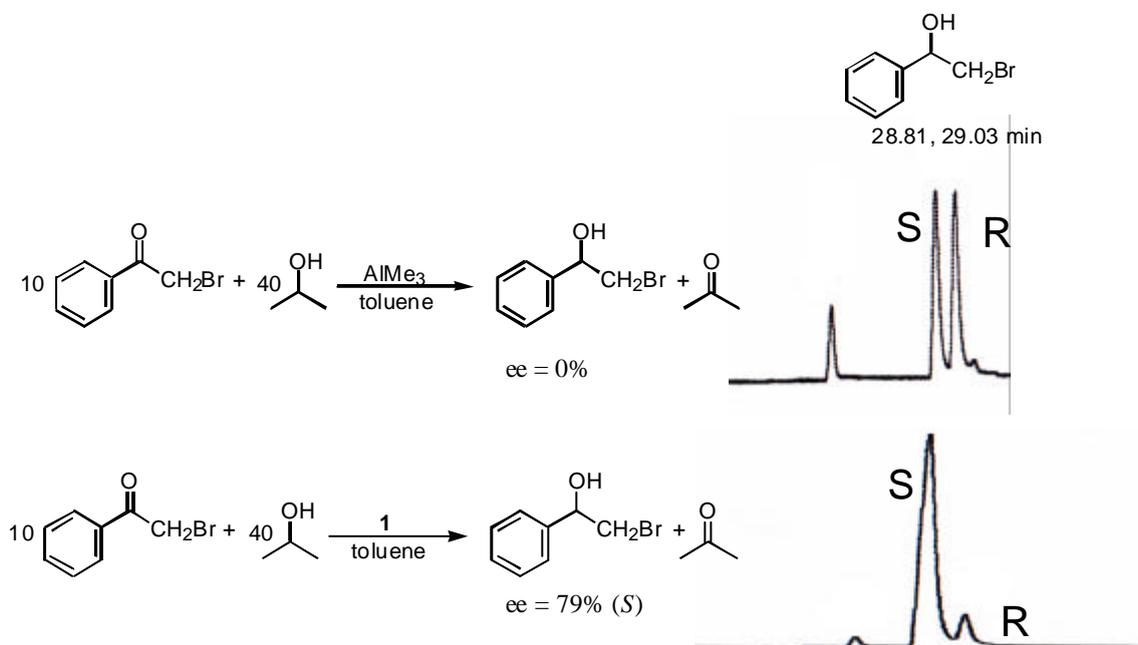


Conversion and ee vs. Time



Time evolution study of chirally pure (*R*)-1-phenylethanol. Into a 4-mL-vial equipped with a magnetic stir bar was added toluene (500 μ L) and enantiopure (*R*)-BINOL (5.8 mg, 0.02 mmol). AlMe_3 (1.9 ml, 0.02 mmol) was added to the mixture via syringe and the reaction was allowed to stir for 5 minutes when a white precipitate formed. (*R*)-1-phenylethanol (24 μ L, 10 equiv), Acetone (31 μ L, 20 equiv) and 2-propanol (30 μ L, 20 equiv) were added and the vial was sealed with a teflon-lined silicone septa. The reaction was stirred at room temperature under nitrogen for 16 hours. Aliquots (20 μ L) were passed through a plug of neutral aluminum oxide [activated, Brockmann activity 1, ~150 mesh] and analyzed on GC to determine selectivity and conversion data.

Synthesis of ((*S*)-BINOL) $\text{AlMe}_3\cdot\text{THF}$ (1**).** In the drybox and into a 50-mL Schlenk flask equipped with a magnetic stir bar was added Toluene (10 mL) THF (160 μ L, 2 equiv) and (*S*)-(-)-BINOL (0.287 g, 1 mmol). AlMe_3 (96.5 μ L, 1 mmol) was added to the mixture via syringe and the reaction was allowed to stir for 3 h. The solvent was removed under vacuum and a white solid was collected, yield = 350 mg (88%). ^1H NMR (500 MHz, $\text{DMSO-}d_6$): δ 7.92 (b, 3H, naphthyl), 7.32-6.85 (b, 8H, naphthyl), 5.81 (b, 1H, naphthyl), 3.82 (m, 2H, OCH_2), 3.59 (m, 2H, OCH_2), 1.60 (m, 2H, $\text{OCH}_2\text{CH}=\text{CH}$), -0.76 (s, 3H, Al-CH_3). ^{13}C NMR (500 MHz, $\text{DMSO-}d_6$): δ 157.69, 150.03, 134.40, 133.96, 130.64, 128.98, 128.67, 127.68, 126.10, 125.85, 125.19, 124.39, 123.86, 122.81, 122.21, 117.02, 70.63, 25.53, -11.12 (Al-CH_3).

The ^1H NMR spectra of complex 1 in $\text{DMSO-}d_6$.The chiral GC trace for the MSPV reduction of α -bromoacetophenone catalyzed by ((S)-BINOL)AlMe \cdot THF.

References.

- (1) Wipf, P.; Jung, J.-K. *J. Org. Chem.* **2000**; *65*, 6319-6337.
- (2) Pirkle, W. H.; Schreiner, J. L. *J. Org. Chem.* **1981**, *46*, 4988-4991.
- (3) Ko, B.-T.; Chao, Y.-C.; Lin, C.-C. *Inorg. Chem.* **2000**, *39*, 1463-1469.
- (4) Campbell, E. J.; Zhou, H.; Nguyen, S. T. *Org. Lett.* **2001**, *3*, 2391-2393.