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## Supporting Information

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# Synthesis of the Putative Structure of 7-deoxycylindrospermopsin: C7-oxygenation is not required for the inhibition of protein synthesis.

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Unless otherwise noted, materials were obtained from commercial sources and used without purification. All reactions requiring anhydrous conditions were performed under a positive pressure of argon using flame-dried glassware. Dichloromethane, diisopropylamine, triethylamine, and *N,N*-diisopropylethylamine were distilled from CaH<sub>2</sub> immediately prior to use. Tetrahydrofuran, diethylether, toluene, and dimethylformamide were degassed with argon and passed through a solvent purification system (J.C. Meyer of Glass Contour) containing either alumina or molecular sieves. Flash chromatography was performed on Merk silica gel Kieselgel 60 (230-400 mesh) from EM science with the indicated solvent. Alkyllithium reagents were standardized in THF with diphenylacetic acid as the acid and indicator.<sup>1</sup>

<sup>1</sup><sup>1</sup>H NMR spectra were recorded on Varian 300, 400, or 500 MHz spectrometers as indicated. The chemical shifts ( $\delta$ ) of proton resonances are reported relative to CHCl<sub>3</sub>, DMSO-*d*<sub>5</sub>, HOD, or HD<sub>2</sub>COD using the following format: chemical shift [multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, app = apparent), coupling constant(s) (*J* in Hz), integral].<sup>2,3</sup> <sup>13</sup>C NMR spectra were recorded at 75, 100, or 125 MHz. The chemical shifts of carbon resonances are reported relative to the deuterated solvent peak, except those in D<sub>2</sub>O which are referenced to methanol.<sup>2</sup>

Infrared spectra were recorded on a Nicolet Avatar 320-FT IR spectrometer. All absorptions are reported in cm<sup>-1</sup> relative to polystyrene (1601 cm<sup>-1</sup>). Spectra that were recorded 'neat' refer to a thin film of pure liquid on NaCl plates. Spectra were also recorded as films deposited from CDCl<sub>3</sub> (dep. CDCl<sub>3</sub>) or CH<sub>2</sub>Cl<sub>2</sub> (dep. CH<sub>2</sub>Cl<sub>2</sub>) solutions on NaCl plates followed by solvent evaporation. Peaks reported in the IR spectrum are described using the following conventions: w = weak, m = medium, s = strong, vs = very strong, sh = shoulder, and br = broad.

Mass spectra were obtained at the Colorado State University CIF on a Fisons VG Autospec. Optical rotations were obtained with a 2 mL, 1 dm cell on a Rudolf Research Autopol III polarimeter operating at 589 nm. CHCl<sub>3</sub> was distilled from CaCl<sub>2</sub> for optical rotations where indicated. HPLC data was obtained on a Waters 600 HPLC system Interfaced with Varian Dynamax Integration software using the indicated column and eluent conditions. Melting points are uncorrected.

Experimental details for the preparation of compounds **7-10** can be found at:

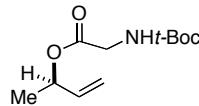
[http://www.wiley-vch.de/contents/jc\\_2002/2004/z54208\\_s.pdf](http://www.wiley-vch.de/contents/jc_2002/2004/z54208_s.pdf)

<sup>1</sup> Kofron, W. G.; Baclawski, L. M. *J. Org. Chem.* **1976**, 41, 1879-1880.

<sup>2</sup> Gottlieb, H. E.; Kotlyar, V.; Nudelman, A. *J. Org. Chem.* **1997**, 62(21), 7512-7515.

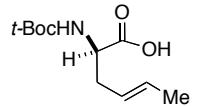
<sup>3</sup> Hoye, T.R.; Hansen, P.R.; Vyvyan, J.R. *J. Org. Chem.* **1994**, 59(15), 4096-4103.

**tert-Butoxycarbonylamino-acetic acid 1-methyl-allyl ester.**



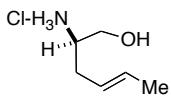
To a solution of 3-buten-2-ol (2.00 g, 27.7 mmol), 4-dimethylamino pyridine (10 mol%, 346 mg, 2.77 mmol), and *N*-tert-butoxycarbonyl glycine (5.35 g, 30.5 mmol) in  $\text{CH}_2\text{Cl}_2$  (50 mL) was added diisopropylcarbodiimide (4.78 mL, 30.5 mmol) in  $\text{CH}_2\text{Cl}_2$  (10 mL) at 0 °C. The mixture was stirred for 2h and filtered through Celite with  $\text{CH}_2\text{Cl}_2$  (100 mL). The combined organics were washed with 10% HCl, sat.  $\text{NaHCO}_3$ , brine, and dried ( $\text{Na}_2\text{SO}_4$ ). The concentrated organics were purified by flash chromatography (6:1 Hex: EtOAc) to give the ester as a colourless oil (6.12 g, 96%).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz):  $\delta$  5.83 (ddd,  $J$  = 17.3, 10.5, 6.6 Hz, 1H), 5.40 (qd (app quintet),  $J$  = 6.6, 6.6 Hz, 1H), 5.25 (dd,  $J$  = 17.2, 1.2 Hz, 1H), 5.15 (dd,  $J$  = 10.5, 1.2 Hz, 1H), 5.00 (br s, 1H), 3.90 (app d,  $J$  = 3.9 Hz, 2H), 1.45 (s, 9H), 1.33 (d,  $J$  = 6.6 Hz, 3H).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 75 MHz):  $\delta$  169.69, 155.84, 137.19, 116.52, 80.07, 72.39, 42.82, 28.53, 20.12. IR (Dep.  $\text{CDCl}_3$ ): 3381 (m); 2980 (m); 1751 (s, sh); 1719 (s); 1520 (m); 1368 (m); 1168 (s). HRMS (FAB $^+$ ): Calc. For  $\text{C}_{11}\text{H}_{20}\text{NO}_4$  [M+H]: (*m/z*) 230.1393; Found (*m/z*) 230.1392.

**rac-2-tert-Butoxycarbonylamino-hex-4-(E)-enoic acid.**



To a solution of the ester (2.72 g, 11.9 mmol) in THF (30 mL) under an argon atmosphere was added a 1M solution of Sodium bis(trimethylsilyl)amide in THF (2.2 eq., 26.1 mL, 26.1 mmol) at 0°C. The mixture was allowed to warm to r.t.. After 2h the reaction was quenched with sat.  $\text{NH}_4\text{Cl}$  (5 mL) and brought to pH = 2 by the addition of 10% HCl. The mixture was extracted with  $\text{Et}_2\text{O}$  (3 x 50 mL), the combined organics were washed with brine and dried ( $\text{Na}_2\text{SO}_4$ ). Concentration gave the acid as a light yellow oil (2.69g, 99%).  $^1\text{H}$ NMR ( $\text{CDCl}_3$ , 300MHz):  $\delta$  10.25 (br s, 1H), 5.60 (dq,  $J$  = 15.0, 6.3 Hz, 1H), 5.40-5.24 (m, 1H), 5.00 (d,  $J$  = 7.7 Hz, 1H), 4.34 (br m, 1H), 2.58-2.40 (m, 2H), 1.66 (dd,  $J$  = 6.3, 0.9 Hz, 3H), 1.44 (s, 9H).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 75 MHz):  $\delta$  177.22, 155.66, 130.48, 124.54, 80.45, 52.23, 35.44, 28.55, 18.27. IR (Dep.  $\text{CDCl}_3$ ): 3330 (m, br); 2978 (m); 1716 (s, br); 1508 (m); 1165 (s). HRMS (FAB $^+$ ): Calc. for  $\text{C}_{11}\text{H}_{20}\text{NO}_4$  [M+H]: (*m/z*) 230.1392; Found (*m/z*) 230.1393.

**2-(E)-aminohex-4-en-1-ol- hydrochloride.**

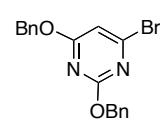


To a solution of the acid (13.5g, 58.9 mmol) was added  $\text{Et}_3\text{N}$  (9.09 mL, 64.7 mmol) in THF (100 mL) and the mixture cooled to 0°C. Ethyl chloroformate (6.16 mL, 64.7 mmol) was then added dropwise over 10 min. After stirring for 20 min the mixture was filtered to remove the  $\text{Et}_3\text{NHCl}$ . This solution was added dropwise to a slurry of  $\text{NaBH}_4$  (3.34 g, 88.4 mmol) in  $\text{H}_2\text{O}$  (70 mL) over 0.5 h, after the addition another portion of  $\text{NaBH}_4$  (1.0 g, 26.4 mmol) was added. The mixture was stirred for 3h at rt and quenched by the addition of AcOH. After concentration the mixture was partitioned between  $\text{Et}_2\text{O}$  and  $\text{H}_2\text{O}$ . The organics were further washed 1 x 10% HCl, 1 x sat.  $\text{NaHCO}_3$ , 1 x brine and dried ( $\text{Na}_2\text{SO}_4$ ). [an analytical sample was purified on silica gel eluting with 1:1 hexanes:EtOAc] Concentration gave a light yellow oil that was added to a mixture of AcCl (6.31 mL, 88.4 mmol) in MeOH (100 mL) that had been stirred for 15 min. The combined solutions were stirred at rt for 8h and concentrated.  $\text{CH}_2\text{Cl}_2$  (100 mL) was added and the mixture allowed to stand at 0°C until crystallization occurred. The solid was filtered off and washed with  $\text{Et}_2\text{O}$  to give the amine salt 7 as a white solid (4.82 g). The mother liquor was concentrated and diluted with  $\text{CH}_2\text{Cl}_2$  to afford an additional crop (531 mg, 5.35 g combined, 60%).

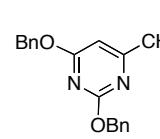
**N-Boc-crotylglycinol:**  $^1\text{H}$  NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  5.57 (dq,  $J$  = 15.6, 6.3 Hz, 1H), 5.42 (dt,  $J$  = 15.6, 7.1 Hz, 1H), 3.66 (app t,  $J$  = 9 Hz, 2H), 3.60 (burried m, 1H), 2.27 (ddd,  $J$  = 13.2, 7.2, 1.2 Hz, 1H), 2.17 (ddd,  $J$  = 13.2, 6.9, 1.2 Hz, 1H), 1.71 (d,  $J$  = 6.3 Hz, 3H), 1.48 (s, 9H).  $^{13}\text{C}$  NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  151.94, 124.12, 121.98, 75.17, 60.815, 48.06, 30.32, 23.95, 13.58. IR (Dep. CDCl<sub>3</sub>): 3350 (br, s), 1694 (s), 1520, 1366 (both m), 1172 (s), 1056, 967 (both m). HRMS (FAB+): Calc. For C<sub>11</sub>H<sub>21</sub>NO<sub>3</sub> [M+H]: (*m/z*) 216.1600; Found (*m/z*) 216.1599.

**7:**  $^1\text{H}$  NMR (D<sub>2</sub>O, 300 MHz):  $\delta$  5.73 (dq,  $J$  = 15, 6.3 Hz, 1H), 5.45 (ddd,  $J$  = 15, 7.5, 7.5 Hz, 1H), 3.84 (dd,  $J$  = 12.6, 4.2 Hz, 1H), 3.64 (dd,  $J$  = 12.6, 7.2 Hz, 1H), 3.38 (dddd,  $J$  = 7.8, 7.2, 6.6, 4.2 Hz, 1H), 2.40 (ddd,  $J$  = 15, 7.5, 6.6 Hz, 1H), 2.30 (ddd,  $J$  = 15, 7.8, 7.5 Hz, 1H), 1.70 (d,  $J$  = 6.3 Hz, 3H).  $^{13}\text{C}$  NMR (D<sub>2</sub>O, 75 MHz):  $\delta$  132.12, 124.22, 61.44, 53.41, 32.65, 17.90. HRMS (FAB+): Calc. for C<sub>6</sub>H<sub>13</sub>NO [M+H]: (*m/z*) 116.1075; Found (*m/z*) 116.1080.

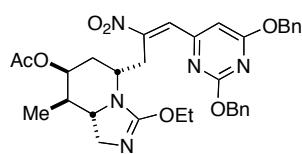
### 2,4-bis(benzylxy)-6-bromopyrimidine.

 To a solution of benzyl alcohol (0.11 mL, 1.03 mmol) in THF (0.5 mL) under an argon atmosphere at 0 °C was added a 1.6 M solution of *n*BuLi in hexanes (0.62 mL, 0.99 mmol). The mixture was stirred 10 min and DMF (5 mL) added. A solution of the tribromopyrimidine in DMF (1 mL) was added and the mixture stirred at 0 °C for 3h. The reaction was quenched with sat. NH<sub>4</sub>Cl and diluted with H<sub>2</sub>O (10 mL). The aqueous phase was extracted with Et<sub>2</sub>O (3 x 10 mL) and the combined organics washed with brine and dried (Na<sub>2</sub>SO<sub>4</sub>). The crude oil was purified on silica gel eluting with 15 : 1 hexanes : EtOAc to give the dibenzylloxypyrimidine as a clear oil (137 mg, 80%).  $^1\text{H}$  NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  7.47-7.32 (m, 10H), 6.66 (s, 1H), 5.43 (s, 2H), 5.40 (s, 2H).  $^{13}\text{C}$  NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  171.10, 163.77, 152.26, 135.89, 135.60, 128.71, 128.58, 128.53, 128.43, 128.34, 128.25, 105.54, 70.09, 69.14. IR (Dep. CDCl<sub>3</sub>): 2952 (w), 1549, 1404, 1323 (all s), 1130, 1003 (both m). HRMS (FAB+): Calc. for C<sub>18</sub>H<sub>16</sub>N<sub>2</sub>O<sub>2</sub><sup>81</sup>Br<sub>1</sub> [M+H]: (*m/z*) 373.0375; Found (*m/z*) 373.0363. Calc. for C<sub>18</sub>H<sub>16</sub>N<sub>2</sub>O<sub>2</sub>Br<sub>1</sub> [M+H]: (*m/z*) 371.0395; Found (*m/z*) 371.0383.

### 2,6-bis(benzylxy)pyrimidine-4-carbaldehyde (12).

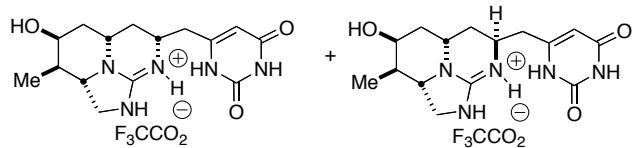
 To a solution of the bromopyrimidine (491 mg, 1.32 mmol) in Et<sub>2</sub>O (30 mL) under argon at -100°C was added a 1.6 M solution of *n*BuLi in hexanes (1.07 mL, 1.71 mmol). The mixture was stirred for 20 min. and then DMF (0.51 mL, 6.60 mmol) was added. The mixture was allowed to warm to rt over 0.5 h and then refluxed briefly with a heat gun. After cooling to rt 10% HCl (5 mL) was added and the mixture stirred vigorously for 10 min. The mixture was partitioned between 10% HCl and Et<sub>2</sub>O and the aqueous phase extracted again with Et<sub>2</sub>O. The combined organics were washed with brine and dried (Na<sub>2</sub>SO<sub>4</sub>). The crude oil was purified on silica gel eluting with 9 : 1 hexanes : EtOAc to give the aldehyde as a clear oil (308 mg, 73%) which solidified upon standing.  $^1\text{H}$  NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  9.90 (s, 1H), 7.53-7.35 (m, 10H), 6.96 (s, 1H), 5.51 (s, 2H), 5.46 (s, 2H).  $^{13}\text{C}$  NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  191.99, 172.29, 165.65, 160.56, 136.12, 135.54, 128.73, 128.62, 128.55, 128.34, 128.23, 99.99, 69.94, 69.33. IR (Dep. CDCl<sub>3</sub>): 2953, 2836 (both w), 1721, 1590, 1566, 1401, 1337 (all s), 1248, 1097 (both m). HRMS (FAB+): Calc. for C<sub>19</sub>H<sub>17</sub>N<sub>2</sub>O<sub>3</sub> [M+H]: (*m/z*) 321.1239; Found (*m/z*) 321.1238.

**5-((E)-3-(2,6-bis(benzyloxy)pyrimidin-4-yl)-2-nitroallyl)-3-ethoxy-8-methyl-1,5,6,7,8,8a-hexahydroimidazo[1,5-a]pyridin-7-yl acetate (13).**



To a solution of the isourea **10** (23 mg, 73  $\mu$ mol) and the pyrimidine aldehyde (26 mg, 81  $\mu$ mol, 1.1 eq.) in  $\text{CH}_2\text{Cl}_2$  (1 mL) under argon was added  $\text{Ac}_2\text{O}$  (34  $\mu$ L, 0.35 mmol, 5 eq.).  $\text{CsF}$  (110 mg, 0.73 mmol) was then added as a solid in one portion. The reaction was diluted with  $\text{MeCN}$  (3 mL) and the mixture stirred for 4h. The reaction was concentrated under reduced pressure, taken up in  $\text{CH}_2\text{Cl}_2$  and filtered to remove the cesium salts. This mixture was again concentrated and purified on silica gel eluting with 10%  $\text{MeOH} / \text{CH}_2\text{Cl}_2$  to give the nitroalkene as a yellow oil (30 mg, 67%) as a single geometric isomer. *This compound is unstable, decomposing overnight at rt.*  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz):  $\delta$  7.68 (s, 1H), 7.48-7.30 (m, 10H), 6.58 (s, 1H), 5.52-5.40 (m, 4H), 4.98 (br 2,  $J$  = 3.2 Hz), 4.28-4.18 (m, 3H), 4.00 (dd,  $J$  = 14, 5 Hz, 1H), 3.66 (ddd,  $J$  = 15, 10, 5 Hz, 1H), 3.55 (dd,  $J$  = 10, 8 Hz, 1H), 3.40-3.30 (m, 1H), 3.12 (dd,  $J$  = 10, 8 Hz, 1H), 1.98 (s, 3H), 1.78-1.64 (m, 1H), 1.62-1.60 (m, 2H), 1.25 (t,  $J$  = 7.2 Hz, 3H), 0.76 (d,  $J$  = 6.8 Hz, 3H).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz):  $\delta$  172.3, 170.6, 165.0, 164.3, 159.8, 155.2, 136.2, 135.7, 130.4, 128.9, 128.7, 128.5, 128.4, 127.8, 106.9, 71.6, 69.8, 69.1, 65.3, 64.1, 52.9, 50.2, 36.7, 35.4, 31.1, 21.2, 14.7, 13.0. HRMS (FAB $+$ ): Calc. for  $\text{C}_{33}\text{H}_{38}\text{N}_5\text{O}_7$  [M $+$ H]: (m/z) 616.2771; Found: (m/z) 616.2795.

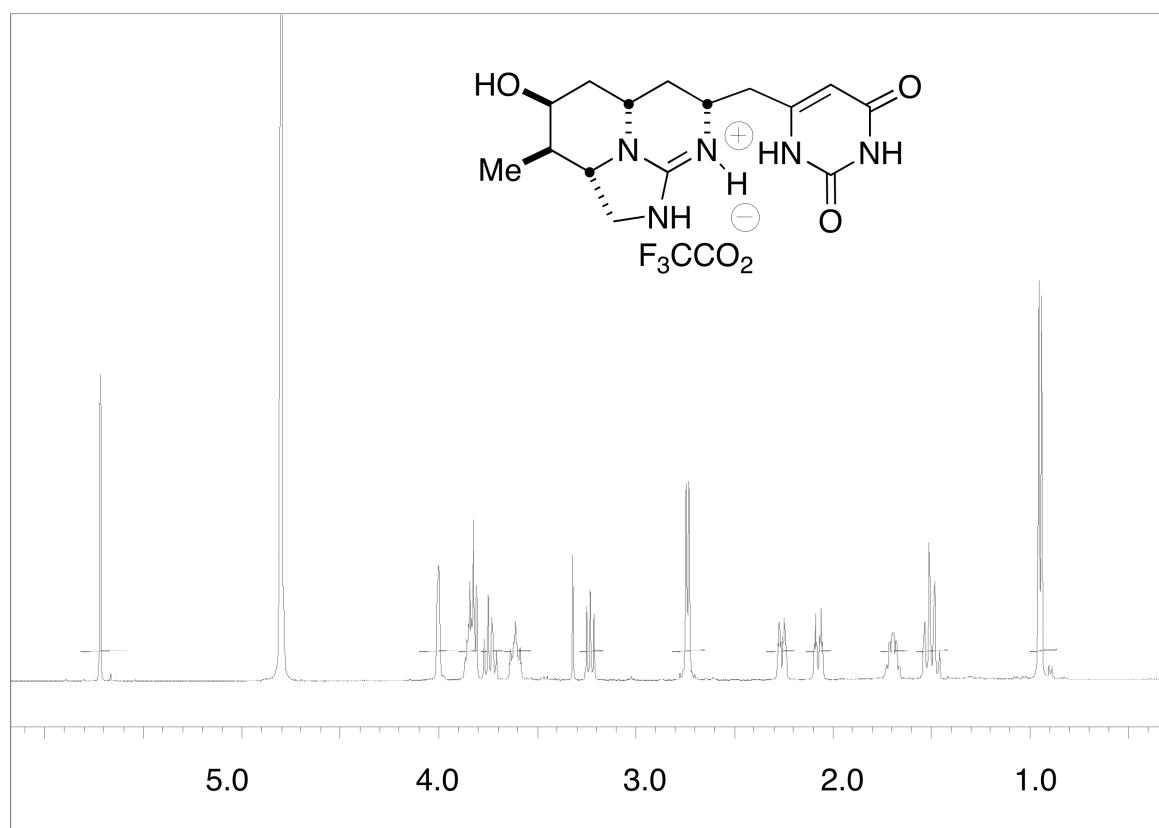
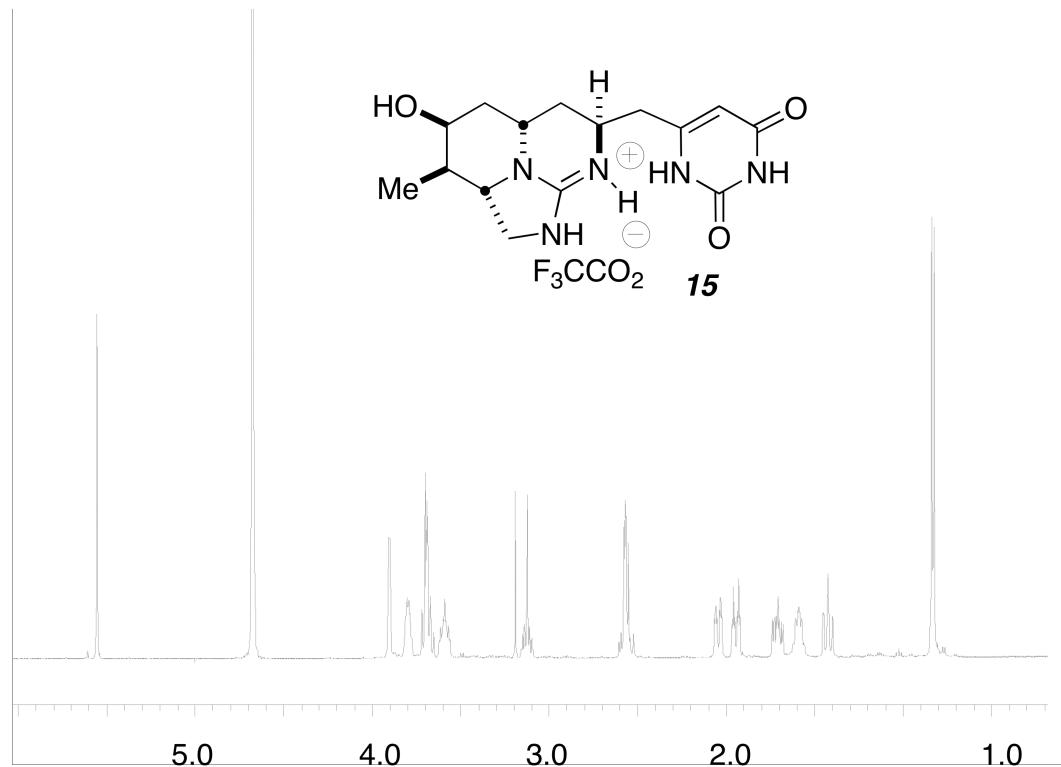
**7-Deoxycylindrospermopsin diol**



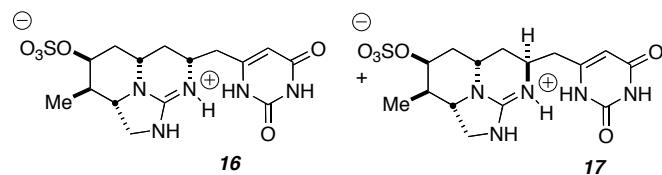
A solution of the nitroalkene **13** (18 mg, 29.2  $\mu$ mol) in  $\text{EtOH}$  (0.5 mL) was added dropwise to a slurry of  $\text{NaBH}_4$  (5 mg, 146  $\mu$ mol) in  $\text{EtOH}$  (0.5 mL) over 20 min. After stirring for 1.5 h the reaction was quenched by the addition of 1:1  $\text{H}_2\text{O}:\text{AcOH}$  (0.1 mL)

and concentrated. The concentrate was diluted with 5%  $\text{AcOH}:\text{MeOH}$  (5.8 mL, to be 5 mM) and purged with argon.  $\text{Pd}(\text{OH})_2$  (20% / C, 6 mg) was added and the mixture stirred under a hydrogen atmosphere for 12 h, filtered through a 0.45  $\mu$ m Acrodisc<sup>®</sup> and concentrated. The residue was dissolved in conc.  $\text{HCl}$  and refluxed for 1h and concentrated. Purification of the uracils was achieved by HPLC using a Waters Symmetry<sup>®</sup> C-18 column (4.6 x 250 mm) eluting with 8%  $\text{MeOH} / \text{H}_2\text{O}$  with 1% TFA at 1.5 mL/min, monitoring at 263 nm to give 7-deoxy-cylindrospermopsin diol **14** as a white solid (3.7 mg, 38%,  $t_{\text{R}}$  = 22.1 min) preceded by the C8 diastereomer **15** also obtained as a white crystalline solid (4 mg, 38 %,  $t_{\text{R}}$  = 12.6 min). A small sample of **15** (~1 mg) was recrystallized from methanol (layered with pentane) to give X-ray quality crystals.

**15 (8S $^*$ ):**  $^1\text{H}$  NMR ( $\text{D}_2\text{O}$ , 500 MHz):  $\delta$  5.68 (s, 1H), 4.03 (br s, 1H), 3.92 (m, 1H), 3.82 (dd,  $J$  = 9, 9 Hz, 1H), 3.78 (dd,  $J$  = 9, 9 Hz, 1H), 3.72 (dd,  $J$  = 11, 11, 4, 4 Hz, 1H), 3.25 (m, 1H), 2.71 (dd,  $J$  = 14, 5.5 Hz, 1H), 2.67 (dd,  $J$  = 14, 9 Hz, 1H), 2.16 (dt,  $J$  = 14, 4, 4 Hz, 1H), 2.06 (dt,  $J$  = 15, 3 Hz, 1H), 1.83 (ddd,  $J$  = 15, 11, 5 Hz, 1H), 1.72 (ddq,  $J$  = 14, 7, 3 Hz, 1H), 1.55 (ddd,  $J$  = 14, 14, 1.5 Hz, 1H), 0.95 (d,  $J$  = 7 Hz, 3H). HRMS (FAB $+$ ): Calc. for  $\text{C}_{15}\text{H}_{22}\text{N}_5\text{O}_3$  [M $+$ H]: (m/z) 320.1723; Found: (m/z) 320.1723. **14 (8R $^*$ ):**  $^1\text{H}$  NMR ( $\text{D}_2\text{O}$ , 500 MHz):  $\delta$  5.72 (s, 1H), 4.00 (br s, 1H), 3.86 (buried m, 1H), 3.82 (dd,  $J$  = 9.0, 9.0 Hz, 1H), 3.74 (dd,  $J$  = 10, 10 Hz, 1H), 3.61 (ddt,  $J$  = 11, 11, 3.5 Hz, 1H), 3.23 (dd,  $J$  = 10, 10 Hz, 1H), 2.73 (app d,  $J$  = 5 Hz, 1H), 2.26 (dt,  $J$  = 15, 5, 5 Hz, 1H), 2.07 (dt,  $J$  = 15, 3, 3 Hz, 1H), 1.70 (ddq,  $J$  = 9, 6.5, 2.5, 1H), 1.50 (app q,  $J$  = 11 Hz, 2H), 0.95 (d,  $J$  = 6.5 Hz, 3H). HRMS (FAB $+$ ): Calc. for  $\text{C}_{15}\text{H}_{22}\text{N}_5\text{O}_3$  [M $+$ H]: (m/z) 320.1723; Found: 320.1712.

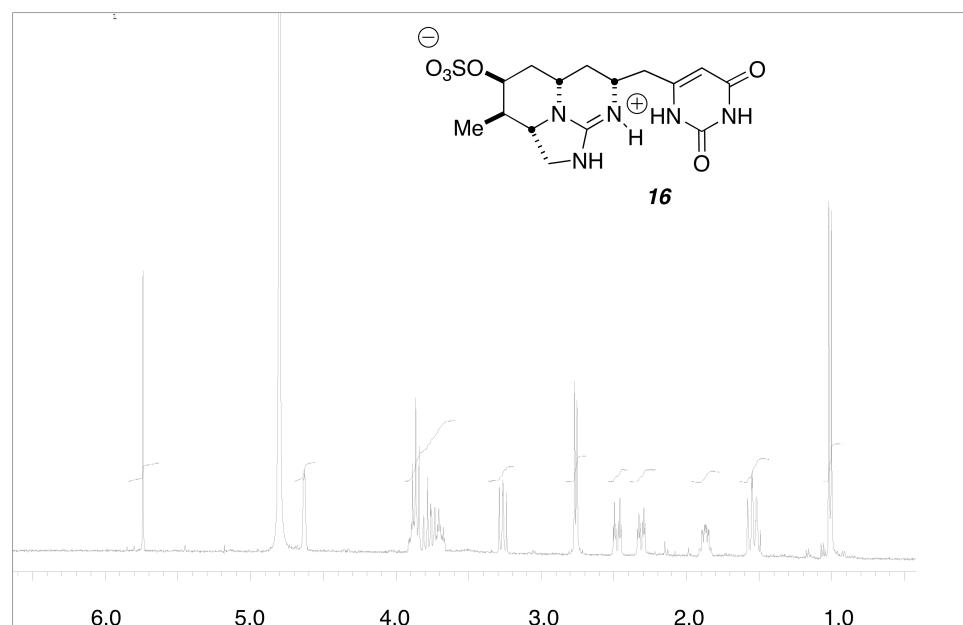
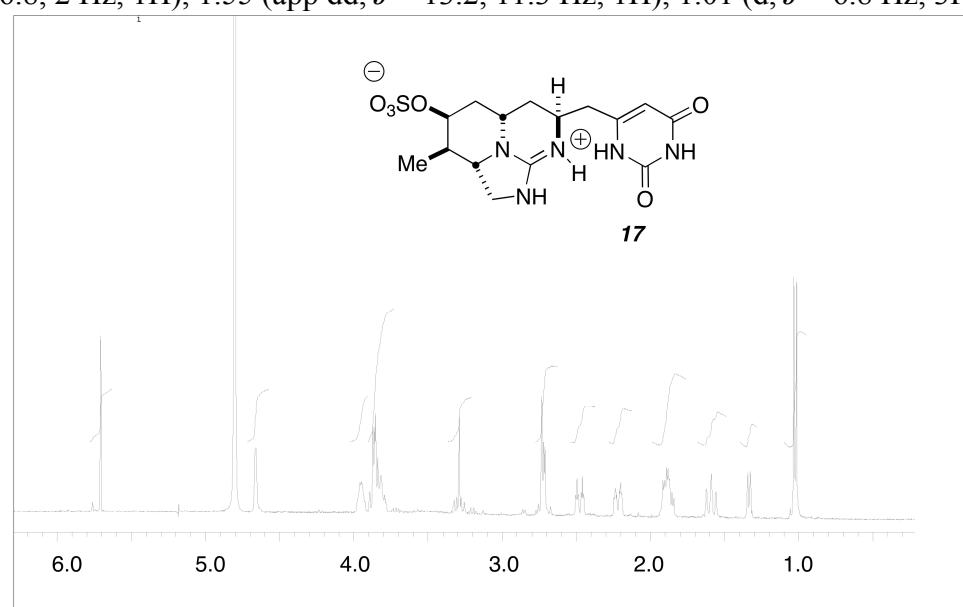


**7-deoxycylindrospermopsin.**

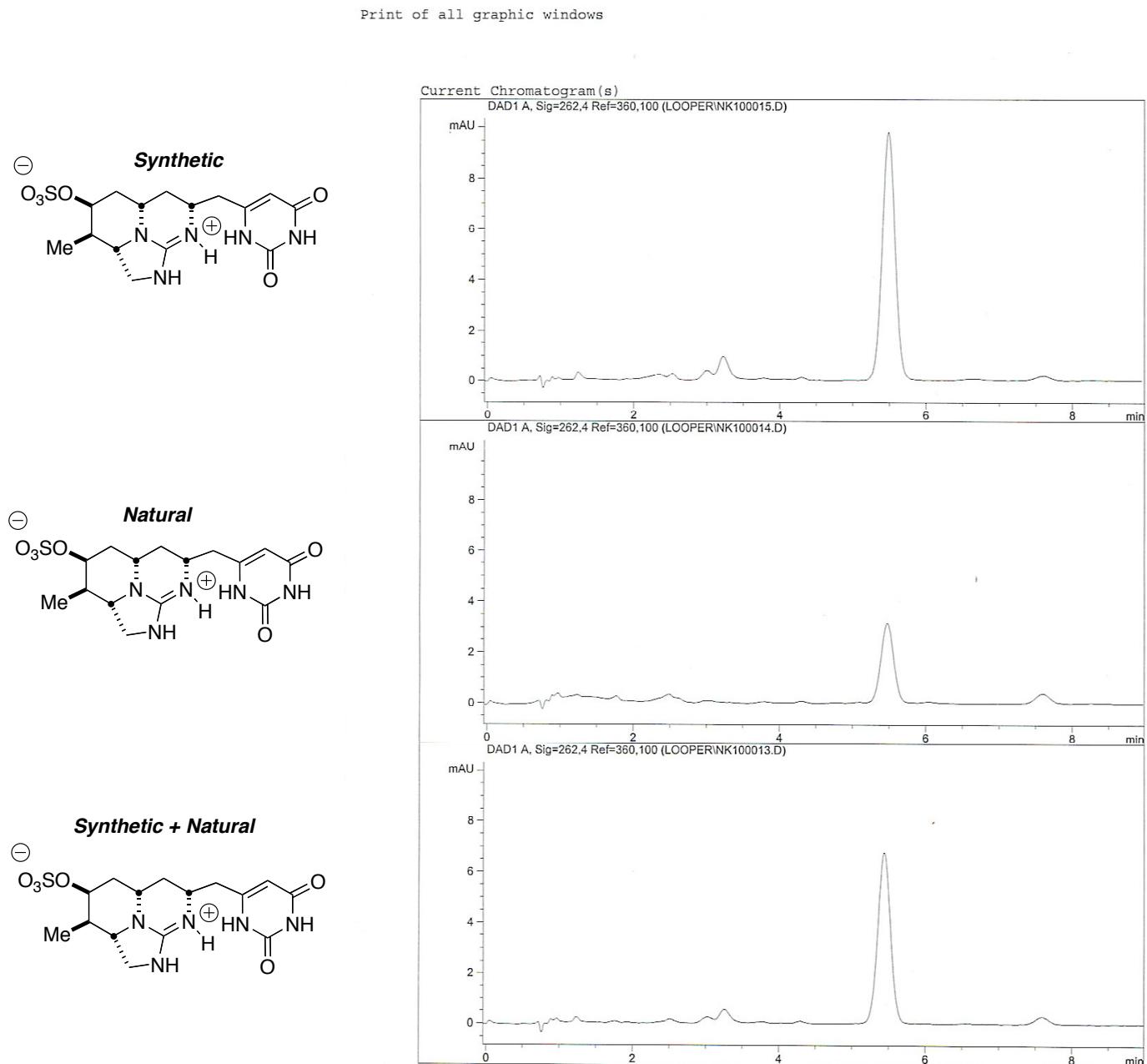


eluting with 8% MeOH / H<sub>2</sub>O with 1% TFA at 1.5 mL/min, monitoring at 263 nm to give 7-deoxycylindrospermopsin **16** as a white solid (1 mg, 33%, *t*<sub>R</sub> = 8.25 min) preceded by the C8 diastereomer **17** also obtained as a white crystalline solid (1 mg, 33 %, *t*<sub>R</sub> = 4.91 min). **16**: <sup>1</sup>H NMR (D<sub>2</sub>O, 400 MHz):  $\delta$  5.74 (s, 1H), 4.63 (br s, 1H), 3.92-3.85 (buried m, 1H), 3.86 (dd, *J* = 8.9, 8.9 Hz, 1H), 3.78 (dd, *J* = 10.7, 10.7 Hz, 1H), 3.70 (dd, *J* = 11.3, 11.3, 3.8, 3.8 Hz, 1H), 3.26 (dd, *J* = 10.8, 8.9 Hz, 1H), 2.76 (app d, *J* = 6.8 Hz, 2H), 2.48 (ddd, *J* = 14.3, 3.8, 3.8 Hz, 1H), 2.32 (ddd, *J* = 13.2, 3.6, 3.6 Hz, 1H), 1.87 (ddd, *J* = 8.9, 6.8, 2 Hz, 1H), 1.55 (app dd, *J* = 13.2, 11.3 Hz, 1H), 1.01 (d, *J* = 6.8 Hz, 3H).

Alternatively a mixture of the C12-hydroxy uracils (3.2 mg, 7.9  $\mu$ mol) can be directly sulfonated by treatment with SO<sub>3</sub>·pyr (19 mg, 120  $\mu$ mol) in DMF (300  $\mu$ L). Purification of the uracils after concentration was achieved by HPLC using a Waters Symmetry<sup>®</sup> C-18 colum (4.6 x 250 mm)



Comparison of synthetic and natural 7-deoxycylindrospermopsin. HPLC: Agilent ZORBAX® C-18 column (4.6 x 150 mm) eluting with 4% (MeOH w/ 0.1% TFA) / H<sub>2</sub>O at 2.0 mL/min, monitoring at 262 nm.



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