

Supporting Information © Wiley-VCH 2007

● Wilcy-VOI1 2007

69451 Weinheim, Germany

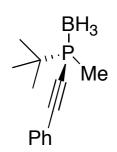
Synthesis and Enantioselectivity of P-Chiral Phosphine Ligands Possessing Alkynyl Groups

Tsuneo Imamoto,* Youichi Saitoh, Aya koide, Tomokazu Ogura, and Kazuhiro Yoshida Department of Chemistry, Graduate School of Science, Chiba University Yayoi-cho, Inage-ku Chiba 263-8522, Japan

General.

All solvents used in reactions were dried and purified according to standard procedure. NMR spectra were measured with JEOL JMN-GXS-500 (500 MHz) or JEOR JMN-LA-400 (400 MHz) spectrometer in CDCl₃. Chemical shifts were reported in δ ppm. Optical rotations were measured with JASCO DIP-370 polarimeter. Enantiomeric excesses were determined by HPLC analysis using Chiralcel AD, OJ-H, OD-H, OB, IA columns and varying concentrations of 2-propanol/hexane as the mobile phase. X-ray crystal structure data were collected using a Bruker SMART APEX II diffractmeter with Mo-Kα radiation. Silica gel (Kanto Chemical, Silica Gel 60N for flash chromatography) was used for column chromatography.

(S)-tert-Butylmethyl(phenylethynyl)phosphine-borane (3a)

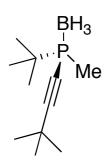


To a solution of (*S*)-*tert*-butylmethylphosphine–borane (**1**) (>99% ee, 1.8 g, 15.4 mmol) in dry Et₂O (60 mL) was added *n*BuLi (11.6 mL of a 1.60 M solution in hexane, 18.5 mmol) at –78 °C under nitrogen, and the mixture was stirred for 15 min. 1,2-Dibromoethane (2.0 mL, 23.1 mmol) was added dropwise, and the reaction mixture was stirred at –78 °C. After 2 h, lithium phenylacetylide (30.8 mmol, 30 mL of Et₂O solution) was added to the reaction mixture, and the mixture was stirred

at room temperature. After 1.5 h, the reaction was quenched with 1 M HCl. The mixture was extracted with EtOAc three times. The combined organic layers were washed with saturated NaHCO₃ and brine, and dried over Na₂SO₄. After filtration, the filtrate was concentrated under reduced pressure, and the residue was purified by chromatography on silica gel (hexane/EtOAc = 5/1) to give (*S*)-*tert*-butylmethyl(phenylethynyl)phosphine–borane (**3a**) (97% ee, 2.8 g, 12.8 mmol, 83%) as white solid. ¹H NMR (500 MHz, CDCl₃) δ 0.42–0.98 (m, 4H), 1.29 (d, ³J(H,P) = 15.6 Hz, 9H), 1.51 (d, ²J(H,P) = 10.0 Hz, 3H), 7.33–7.37 (m, 2H), 7.41 (tt, ³J(H,H) = 15.3, ⁴J(H,H) = 1.7 Hz, 1H), 7.50–7.52 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 8.37 (d, J(CP) = 40 Hz), 24.96 (d, J(CP) = 4 Hz), 29.03 (d, J(CP) = 37 Hz), 79.42 (d,

J(CP) = 90 Hz), 106.00 (d, J(CP) = 12 Hz), 120.77, 128.42, 130.01, 132.20. ³¹P NMR (162 MHz, CDCl₃) δ 18.0. HRMS (FAB) calcd for C₁₃H₁₉BP (M⁺-H) 217.1320, found 217.1313. [α]²²_D = 0.97 cm³ g⁻¹ dm⁻¹ (c = 0.0100 g cm⁻³ in CHCl₃), HPLC: Daicel Chiralcel OD-H, Hexane/iPrOH = 199/1, Flow rate = 0.5 mL/min, UV = 254 nm, t_R = 11.9 min (R), t_R = 13.1 min (S).

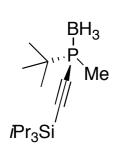
(S)-tert-Butyl(3,3-dimethyl-1-butynyl)methylphosphine-borane (3b)



In a similar manner (*S*)-*tert*-butyl(3,3-dimethyl-1-butynyl)methyl-phosphine–borane was prepared in 82% yield. ¹H NMR (400 MHz, CDCl₃) δ 0.23–0.89 (m, 3H), 1.21 (d, ³*J*(H,H) = 15.1 Hz, 9H), 1.27 (s, 9H), 1.38 (d, ²*J*(H,H) = 10.0 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 8.54 (d, *J*(CP) = 40 Hz), 24.77 (d, *J*(CP) = 3 Hz), 28.62 (d, *J*(CP) = 35 Hz), 30.20 (d, *J*(CP) = 2 Hz), 68.70 (d, *J*(CP) = 95 Hz), 116.79 (d, *J*(CP) = 10 Hz). ³¹P NMR (162 MHz, CDCl₃) δ 15.7. HRMS (FAB) calcd for C₁₁H₂₄BP (M⁺+K) 237.1348, found 237.1351. [α]²²_D = 9.4 cm³ g⁻¹ dm⁻¹

(c = 0.0013 g cm⁻³ in CHCl₃). Its enantiomeric excess was determined to be 98% ee by HPLC analysis. HPLC: Daicel Chiralpak IA, Hexane/*i*PrOH=1000/1, Flow rate = 0.5 mL/min, UV = 230 nm, $t_R = 13.7$ min (R), $t_R = 15.3$ min (S).

(S)-tert-Butylmethyl(triisopropylsilylethynyl)phosphine-borane (3c)



In a similar manner (*S*)-tert-butylmethyl(triisopropylsilylethynyl)-phosphine–borane was prepared in 86% yield. ¹H NMR (400 MHz, CDCl₃) δ 0.27–0.73 (m, 3H), 1.09–1.10 (m, 21H), 1.24 (d, ³*J*(HP) = 15.4 Hz, 9H), 1.43 (d, ²*J*(HP) = 10.2 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 8.45 (d, *J*(CP) = 40 Hz), 10.90, 18.43, 24.82 (d, *J*(CP) = 3 Hz), 28.62 (d, *J*(CP) = 38 Hz), 98.26 (d, *J*(CP) = 77 Hz), 113.26 (d, *J*(CP) = 3 Hz). ³¹P NMR (162 MHz, CDCl₃) δ 17.5. HRMS (FAB) calcd for C₁₆H₃₅BPSi (M⁺–H) 297.2342, found 297.2332. [α]²²_D = –8.9

cm³ g⁻¹ dm⁻¹ (c = 0.0101 g cm⁻³ in CHCl₃). Its enantiomeric excess was determined to be 98% ee by HPLC analysis. Daicel Chiralpak IA, Hexane/*i*PrOH = 1000/1, Flow rate = 0.5 mL/min, UV = 230 nm, $t_R = 11.0$ min (R), $t_R = 13.6$ min (S).

(S)-tert-Butylmethyl(trimethylsilylethynyl)phosphine-borane (3d)

In a similar manner (*S*)-tert-butylmethyl(trimethylsilylethynyl)phosphine–borane was prepared in 81% yield. ¹H NMR (400 MHz, CDCl₃) δ 0.22 (s, 9H), 1.23 (d, ³*J*(HP) = 15.4 Hz, 9H), 1.42 (d, ²*J*(HP) = 10.2 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 8.92 (d, *J*(CP) = 40 Hz), 25.45 (d, *J*(CP) = 4 Hz), 29.28 (d, *J*(CP) = 38 Hz), 96.79 (d, *J*(CP) = 77 Hz), 116.66 (d, *J*(CP)

= 3 Hz). ³¹P NMR (162 MHz, CDCl₃) δ 17.1. $\left[\alpha\right]^{22}_{D}$ = -1.59 cm³ g⁻¹ dm⁻¹ (c = 0.0100 g cm⁻³ in CHCl₃). Its enantiomeric excess was determined to be 99% ee by HPLC analysis. Daicel Chiralcel OJ-H, Hexane/iPrOH = 99/1, Flow rate = 0.5 mL/min, UV = 230 nm, t_{R} = 8.8 min (R), t_{R} = 9.7 min (S).

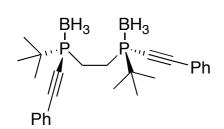
(S)-tert-Butylmethylphenylphosphine-borane (3e), [1-4] (S)-tert-Butyl(o-methoxyphenyl)-methylphosphine-borane (3f), [2,5] and (S)-Benzyl(tert-butyl)methylphosphine-borane (4b) [6,7]

The absolute configurations and enantiomeric excesses of the products (**3e**, **3f**, and **4b**) were determined by HPLC analysis in comparison with the reported data. ^[1-7] **3e**: Daicel Chiralcel OJ-H, Hexane/*i*PrOH = 9/1, Flow rate = 0.5 mL/min, UV = 230 nm, t_R = 25.5 min (*R*), t_R = 27.7 min (*S*); **3f**: Daicel Chiralcel OD-H, Hexane/*i*PrOH = 99/1, Flow rate = 0.5 mL/min, UV = 230 nm, t_R = 11.0 min (*R*), t_R = 11.6 min (*S*); **4b**: Daicel Chiralcel OD-H, Hexane/*i*PrOH = 9/1, Flow rate = 0.5 mL/min, UV = 230 nm, t_R = 16.9 min (*R*), t_R = 18.9 min (*S*).

(S)-n-Butyl(tert-butyl)methylphosphine-borane (4a)

The absolute configuration and the optical purity of compound $\mathbf{4a}$ ($[\alpha]_D = -4.1 \text{ cm}^3 \text{ g}^{-1} \text{ dm}^{-1}$) ($c = 0.010 \text{ g cm}^{-1}$ in CHCl₃)) were determined by comparison of the sign of rotation and specific rotation of the authentic sample ($[\alpha]_D = -4.5 \text{ cm}^3 \text{ g}^{-1} \text{ dm}^{-1}$) ($c = 0.010 \text{ g cm}^{-3}$ in CHCl₃)) prepared by the reaction of compound $\mathbf{1}$ with nBuLi and bromobutane.

(S,S)-1,2-Bis(boranato(*tert*-butyl)(phenylethynyl)phosphino)ethane (6a)

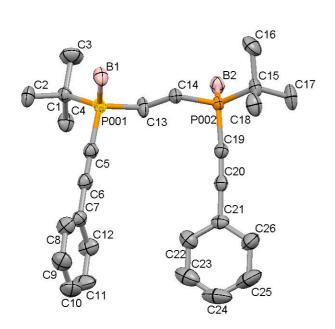


To a solution of (*S*)-tert-butylmethyl(phenylethynyl)-phosphine-borane (**3a**) (97% ee, 0.95 g, 4.4 mmol) and TMEDA (0.8 mL, 5.2 mmol) in dry Et₂O (13 mL) was added *s*-BuLi (5.2 mL of a 1.0 M solution in hexane, 5.2 mmol) at -78 °C under nitrogen, and the mixture was stirred for 1 h and -50 °C for 10 min. Copper(II) chloride

(1.5 g, 11 mmol) was added with vigorous stirring and the resulting mixture was gradually warmed to room temperature. After 2 h, the reaction was quenched with saturated NH₄Cl. The mixture was extracted with EtOAc three times. The combined organic layers were washed with brine, and dried over Na₂SO₄. After filtration, the filtrate was concentrated under reduced pressure, and the residue was purified by chromatography on silica gel (hexane/EtOAc = 4/1) to give (S,S)-1,2-bis(boranato(tert-butyl)(phenylethynyl)phosphino)ethane (**6a**) (0.69 g, 1.6 mmol, 73%) as white solid. ¹H NMR (400 MHz, CDCl₃) δ 0.57–0.80 (m, 6H), 1.32–1.35 (m, 18H), 2.13–2.28 (m, 4H), 7.29–7.32 (m, 4H), 7.41 (tt, 3 J(H,H) = 15.0 Hz, 4 J(H,H) = 2.5 Hz, 2H), 7.46–7.47 (m, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 17.01 (d, J(CP) = 34 Hz), 25.30 (t,

J(CP) = 3 Hz), 30.11 (d, J(CP) = 36 Hz), 77.88 (d, J(CP) = 88 Hz), 107.38, 120.39, 128.48, 130.22, 132.24. ³¹P NMR (162 MHz, CDCl₃) δ 29.6. [α]²²_D = 111 cm³ g⁻¹ dm⁻¹ (c =0.0100 g cm⁻³ in CHCl₃).

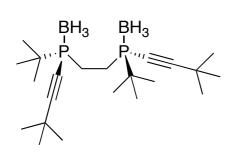
Recrystallization of compound **6a** from a mixed solvent of EtOAc and *n*-hexane gave analysis. Empirical prisms. Absolute configuration of this compound was determined by



single crystal X-ray Formula C₂₆H₄₈B₂P₂; Formula weight 444.20; Temperature 173 K; Wavelength Å; 0.71073 Crystal system Orthorhombic; Space group P2(1)2(1)2; Unit cell dimensions a =22.202(5) Å, $\alpha = 90^{\circ}$, b = $10.932(3) \text{ Å}, \beta = 90^{\circ}, c = 11.379(3)$ Å, $\gamma = 90^{\circ}$; Volume 2761.8(11) Å³; Z = 5; Density (calculated) 1.335 Mg/m³; Crystal size 0.45 x 0.35 x 0.15 mm^3 ; GOF = 1.034; Final R indice $[I>2\sigma(I)] R1 = 0.0446$, wR2 = 0.1045; R indices (all data) R1 =

0.0524, wR2 = 0.1095; Absolute structure parameter 0.06(10). CCDC-641148.

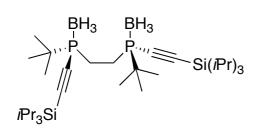
(S,S)-1,2-Bis(boranato(tert-butyl)(3,3-dimethyl-1-butynyl)phosphino)ethane (6b)



This compound was prepared in 73% yield by the oxidative coupling of **4b**. ¹H NMR (400 MHz, CDCl₃) δ 0.22–0.88 (m, 6H), 1.22–1.26 (d, ³J(H,P) = 15.1 Hz, 18H), 1.28 (s, 18H), 1.94–2.02 (m, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 17.05 (d, J(CP) = 34 Hz), 25.12, 28.58, 29.58 (d, J(CP) = 37 Hz), 30.26, 67.35 (d, J(CP) = 93 Hz), 118.42 (d, J(CP) = 10 Hz). ³¹P NMR (162 MHz,

CDCl₃) δ 27.8. HRMS (FAB) calcd for C₂₂H₄₆B₂P₂ (M⁺+K) 433.2906, found 433.2881. [α]²²_D = 19.4 cm³ g⁻¹ dm⁻¹ (c = 0.0096 g cm⁻³ in CHCl₃).

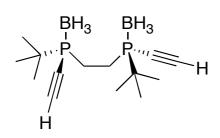
(S,S)-1,2-Bis(boranato(tert-butyl)triisopropylsilylethynylphosphino)ethane (5c)



This compound was prepared in 78% yield by the oxidative coupling of **4c**. 1 H NMR (400 MHz, CDCl₃) δ 0.27–0.73 (m, 6H), 1.06–1.14 (m, 42H), 1.25–1.29 (m, 18H), 1.98–2.13 (m, 4H). 13 C NMR

(100 MHz, CDCl₃) δ 10.92, 16.73 (d, J(CP) = 32 Hz), 18.50 (d, J(CP) = 2 Hz), 25.19, 29.63 (d, J = 36 Hz), 96.47 (d, J = 74 Hz), 115.49. ³¹P NMR (162 MHz, CDCl₃) δ 29.3. HRMS (FAB) calcd for $C_{32}H_{70}B_2P_2Si_2$ (M⁺+K) 633.4327, found 633.4327. [α]²²_D = 25.1 cm³ g⁻¹ dm⁻¹ (c = 0.0100 g cm⁻³ in CHCl₃).

(S,S)-1,2-Bis(boranato(tert-butyl)ethynylphosphino)ethane (6d)

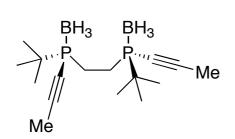


Tetrabutylammonium fluoride (0.4 mL of a 1.0 M solution in THF, 0.4 mmol) was added to (*S*,*S*)-1,2-bis(boranato(*tert*-

butyl)(triisopropylsilylethynyl)phosphino)ethane (**6c**) (60 mg, 0.1 mmol) under nitrogen, and the mixture was stirred at room temperature. After 6 h, the reaction was quenched by the addition of water. The mixture was

extracted with EtOAc three times. The combined organic layers were washed with brine, and dried over Na₂SO₄. After filtration, the filtrate was concentrated under reduced pressure, and the residue was purified by chromatography on silica gel (hexane/EtOAc = 3/1) to give (*S*,*S*)-1,2-bis(boranato(*tert*-butyl)ethynylphosphino)ethane (**6d**) (15 mg, 0.052 mmol, 52%) as white solid. 1 H NMR (400 MHz, CDCl₃) δ 0.24–0.92 (m, 6H), 1.27–1.31 (m, 18H), 2.01–2.15 (m, 4H), 3.08–3.10 (m, 2H). 13 C NMR (100 MHz, CDCl₃) δ 16.31 (d, *J*(CP) = 33 Hz), 25.08, 29.66 (d, *J*(CP) = 35 Hz), 74.06 (d, *J*(CP) = 81 Hz), 96.28. 31 P NMR (162 MHz, CDCl₃) δ 30.9. HRMS (FAB) calcd for $C_{14}H_{30}B_{2}P_{2}Si_{2}$ (M⁺+K) 321.1651, found 321.1662. [α]²²_D = -13.9 cm³ g⁻¹ dm⁻¹ (c = 0.0024 g cm⁻³ in CHCl₃).

(S,S)-1,2-Bis(boranato(*tert*-butyl)(1-propynyl)phosphino)ethane (6e)



To a solution of (S,S)-1,2-bis(boranato(*tert*-butyl)ethynylphosphino)ethane (**6d**) 31 mg, 0.11 mmol) in THF (0.55 mL) was added sBuLi (0.28 mL of a 1.0 M solution in hexane, 0.28 mmol) at -78 °C under nitrogen, and the mixture was stirred for 1 h. Iodomethane (34 μ L, 0.55 mmol) was added to the

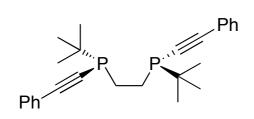
solution with stirring and the mixture was warmed to room temperature during 2 h. The reaction was quenched with saturated NH₄Cl. The mixture was extracted with EtOAc three times. The combined organic layers were washed with brine, and dried over Na₂SO₄. After filtration, the filtrate was concentrated under reduced pressure, and the residue was purified by chromatography on silica gel (hexane/EtOAc = 4/1) to give (S,S)-1,2-bis(boranato(tert-butyl)(1-propynyl)phosphino)ethane (**6e**) (30 mg, 88%) as white solid. ¹H NMR (400 MHz, CDCl₃) δ 0.24–0.86 (m, 6H), 1.23–1.27 (m, 18H), 2.00–2.06 (m, 10H). ¹³C NMR (100 MHz,

CDCl₃) δ 5.17, 16.86 (d, J(CP) = 35 Hz), 25.22, 29.70 (d, J(CP) = 37 Hz), 68.20 (d, J(CP) = 95 Hz), 106.53. ³¹P NMR (162 MHz, CDCl₃) δ 28.1. HRMS (FAB) calcd for $C_{16}H_{34}B_2P_2$ (M⁺+K) 349.1965, found 349.1953. [α]²²_D = 21.6 cm³ g⁻¹ dm⁻¹ (c = 0.00050 g cm⁻¹ in CHCl₃).

Preparation of (S,S)-1,2-Bis(tert-butyl)(1-alkynyl)phosphine)ethanes (AlkynylP*). A typical Procedure

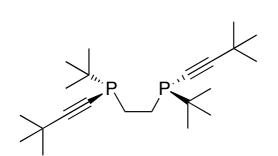
Compound **6a** (36 mg, 0.084 mmol) and 1,4-diazabicyclo[2.2.2]octane (56 mg, 0.51 mmol) were dissolved in degassed THF (0.5 mL) under nitrogen. The solution was stirred under nitrogen at 60 °C for 1 h and the reaction mixture concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel using degassed diethyl ether as the eluent under nitrogen to give diphosphine **5a** (32 mg, 94%). In a similar manner, ligands **5b**, **5c**, **5d**, and **5e** were prepared in 94–98% yield.

(S,S)-1,2-Bis(tert-butyl(phenylethynyl)phosphino)ethane (5a)



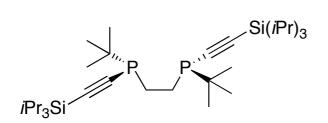
¹H NMR (400 MHz, CDCl₃) δ 1.19–1.26 (m, 18H), 1.67–1.80 (m, 2H), 2.09–2.19 (m, 2H), 7.23–7.31 (m, 3H), 7.40–7.43 (m, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 20.27, 27.55, 29.99, 87.96, 105.38, 123.25, 128.19, 128.25, 131.58. ³¹P NMR (202 MHz, CDCl₃) δ –16.43.

(S,S)-1,2-Bis(*tert*-butyl(3,3-dimethyl-1-butynyl)phosphino)ethane (5b)



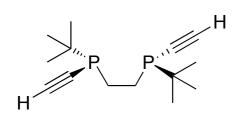
¹H NMR (400 MHz, CDCl₃) δ 1.11–1.19 (m, 18H), 1.25 (s, 18H), 1.48–1.58 (m, 2H), 1.82–1.90 (m, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 20.24, 27.28, 28.56, 29.35, 30.99, 75.61, 115.28. ³¹P NMR (202 MHz, CDCl₃) δ –15.48.

(S,S)-1,2-Bis(tert-butyl(triisopropylsilylethynyl)phosphino)ethane (5c)



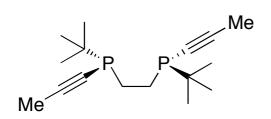
 $\square\square\square\square\square$ ³¹P NMR (202 MHz, CDCl₃) δ –13.09.

(S,S)-1,2-Bis(t-butyl(ethynyl)phosphino)ethane (5d)



¹H NMR (400 MHz, CDCl₃) δ 1.1–1.18 (m, 18H), 1.54–1.65 (m, 2H), 1.96–2.03 (m, 2H), 2.86 (s, CCH). (500 MHz, CDCl₃/CD₃OD) δ 1.14–1.18 (m, 18H), 1.55–1.65 (m, 2H), 1.96–2.03 (m, 2H) (The signal of CCH was not observed.). ¹³C NMR (100 MHz, CDCl₃) δ 19.76, 27.31, 29.40, 83.30–83.53 (m), 93.52. ³¹P NMR (202 MHz, CDCl₃) δ –14.25.

(S,S)-1,2-Bis(tert-butyl(1-propynyl)phosphino)ethane (5e)



¹H NMR (400 MHz, CDCl₃) δ 1.10–1.15 (m, 18H), 1.49–1.60 (m, 2H), 1.84–1.94 (m, 2H), 1.98 (s, 6H). (Diethyl ether was strongly bound to this ligand and could not be removed in vacuo). ¹³C NMR (125 MHz, CDCl₃) δ 5.14, 20.36 27.39, 29.39 (d, J(CP) = 3 Hz), 76.61, 102.52. ³¹P NMR (202 MHz, CDCl₃) δ –13.00.

General Procedure for Rhodium-Catalyzed Asymmetric Hydrogenation of Methyl (Z)- α -Acetamidocinnamate

A 50 mL-hydrogenation tube was charged with methyl (Z)- α -acetamidocinnamate (0.5 mmol). The tube was connected to the hydrogen tank via stainless steel tubing. The vessel was evacuated and filled with 1 atm of hydrogen gas (Nippon Sanso, 99.9999%). A solution of $[Rh(nbd)_2]BF_4$ (1.9 mg, 5.0 μ mol) and (S,S)-1,2-bis(1-alkynyl(tert-butyl)phosphino)ethane (AlkynylP*) (5.5 μ mol) in degassed MeOH (1 mL) was added via a syringe to the tube, and the hydrogen pressure was increased to 1 atm. After 3 h, the reaction mixture was evaporated and the residue was purified by flash chromatography on silica gel using EtOAc as an eluent. The product was characterized by specral data and analyzed by HPLC using a chiral column. The absolute configuration and enantiomeric excess of the product were determined by comparison of the retention times with reported values. Conditions for the HPLC analysis of N-acetylphenylalanine methyl ester: Chiralcel OJ, hexane/2-propanol = 9:1, 0.5 mL/min, wavelength = 254 nm, retention times: 22.3 min (R), 31.7 min (S).

General Procedure for Rhodium-Catalyzed Asymmetric 1,4-Addition of Organoboronic Acids to Enones

A solution of $[Rh(nbd)_2]BF_4$ (1.9 mg, 10 µmol) and ligand (*S*,*S*)-1,2-bis(1-alkynyl(*tert*-butyl)phosphino)ethane (AlkynylP*) (11 µmol) in dioxane (1 mL) was stirred at 40 °C for 15 min under nitrogen. To the reaction mixture was added KOH (0.1 mL, 1.5M, 0.15 mmol) in water and the solution was stirred for 15 min. Arylboronic acid (1.0 mmol) and α , β -unsaturated carbonyl compound (0.50 mmol) was added to the solution. After stirring at 40 °C for 2 h, the reaction mixture was quenched with saturated NaHCO₃ and extracted with ether five times. The combined organic layers were dried over sodium sulfate and concentrated under reduced pressure. The residue was purified by preparative TLC (silica gel, hexane/EtOAc = 3/1). The products were characterized by their spectral data and analyzed by HPLC using chiral columns. The absolute configurations and ee values of the products were determined by comparison of the retention times with reported values.

Conditions for the determination of the ee's by HPLC analysis. 3-Phenylcyclohexanone: Chiralcel OD-H, hexane:2-propanol = 98:2, 0.5 mL/min, wavelength 254 nm, retention times: 26.4 min (*S*), 28.2 min (*R*). 3-(4-Trifluoromethylphenyl)cyclohexanone: Chiralcel OJ, hexane:2-propanol = 98:2, 1.0 mL/min, wavelength 254 nm, retention times: 18.1 min (*R*), 20.5 min (*S*). 3-(4-Methoxyphenyl)cyclohexanone: Chiralcel OJ, hexane:2-propanol = 98:2, 1.0 mL/min, wavelength 254 nm, retention times: 34.6 min (*R*), 42.9 min (*S*). 3-Phenylcyclopentanone: Chiralcel OB, hexane:2-propanol = 98:2, 1.0 mL/min, wavelength 254 nm, retention times: 16.5 min (*S*), 22.4 min (*R*). 5-Methyl-4-phenylhexan-2-one: Chiralcel OJ, hexane:2-propanol = 98:2, 0.5 mL/min, wavelength 254 nm, retention times: 16.3 min (*R*), 20.0 min (*S*).

In a similar manner, 2-cyclohexenone (0.50 mmol) was reacted with phenylboronic acid in the presence of [Rh(nbd)₂]BF₄/tBu-BisP* (2 mol%) at 40 °C. The reaction was sluggish and after 2 h the reaction mixture was worked up to give 3-phenylcyclohexanone in 37% yield. The ee of this product was determined to be 20% by HPLC analysis.

General Procedure for the Alkylative Ring-Opening of Oxabenzonorbornadiene Derivatives

A solution of PdCl₂(cod) (1.1 mg, 8.0 μmol) and (*S*,*S*)-1,2-bis(1-alkynyl(*tert*-butyl)phosphino)ethane (AlkynylP*) (8.8 μmol) in CH₂Cl₂ (1 mL) was stirred at 40 °C under nitrogen for 15 min. To the solution was added a solution of oxabenzonorbornadiene (58 mg, 0.4 mmol) in CH₂Cl₂ (3 mL), followed by dimethylzinc (0.6 mL of 1.0 M hexane solution). The resulting solution was stirred at room temperature until completion of the reaction. The reaction was quenched by the addition of a few drops of water and the mixture was passed through a short plug of Celite and then concentrated. The residue was purified by preparative TLC (hexane/EtOAc).

The products were characterized by their spectral data and analyzed by HPLC using chiral columns. The absolute configurations and ee values of the products were determined by comparison of the retention times with reported values.¹⁰

Conditions for the determination of the ee's by HPLC analysis. 2-Methyl-1,2dihydronaphthalen-1-ol: Chiralcel OD-H, hexane:2-propanol = 199:1, 1 mL/min, wavelength 254 nm, retention times: 28.5 min (1R,2R), 30.6 min (1S,2S); 2-Ethyl-1,2-dihydronaphthalen-1-ol: Chiralcel OD-H, hexane:2-propanol = 199:1, 1 mL/min, wavelength 254 nm, retention times: 26.0 min (1R,2R), 29.6 min (1S,2S). 2-Methyl-1,2-dihydro-5,8-bis(methoxymethoxy)naphthalen-1-ol: Chiralcel OD-H, hexane:2-propanol = 9:1, 1.0 mL/min, wavelength 254 nm, retention times: 9.0 min (1R,2R),11.1 min (1S,2S). 2-Ethyl-1,2-dihydro-5,8bis(methoxymethoxy)-naphthalen-1-ol: Chiralcel OD-H, hexane:2-propanol = 9:1, 0.5 mL/min, wavelength 254 nm, retention times: 15.5 min (1R,2R), 17.0 min (1S,2S).

In a similar manner, oxabenzonorbornadiene (0.4 mmol) was reacted with diethylzinc in the presence of $PdCl_2(cod)/tBu-BisP*$ (2 mol%) at room temperature for 6 h to give (1*S*,2*S*)-2-ethyl-1,2-dihydronaphthalen-1-ol in 93% yield. The ee of this product was determined to be 94% by HPLC analysis.

References

- [1] T. Imamoto, T. Oshiki, T. Onozawa, T. Kusumoto, K. Sato, *J. Am. Chem. Soc.* **1990**, *112*, 5244–5252.
- [2] T. Imamoto, M. Matsuo, T. Nonomura, K. Kishikawa, M. Yanagawa, *Heteroatom Chem.* **1993**, *4*, 475–486.
- [3] E. J. Corey, Z. Chen, G. J. Tanoury, J. Am. Chem. Soc. 1993, 115, 11000–11001.
- [4] M. Stankevic, K. M. Pietrusiewicz, J. Org. Chem. 2007, 72, 816–822.
- [5] Y. Takahashi, Y. Yamamoto, K. Katagiri, H. Danjo, K. Yamaguchi, T. Imamoto, J. Org. Chem. 2005, 70, 9009–9012.
- [6] T. Miura, H. Yamada, S. Kikuchi, T. Imamoto, J. Org. Chem. 2000, 65, 1877–1880.
- [7] H. Danjo, W. Sasaki, T. Miyazaki, T. Imamoto, *Tetrahedron Lett.* **2003**, 44, 3467–3469.
- [8] Imamoto, T.; Watanabe, J.; Wada, Y.; Masuda, H.; Yamada, H.; Tsuruta, H.; Matsukawa, S.; Yamaguchi, K. *J. Am. Chem. Soc.* **1998**, *120*, 1635–1636.
- [9] Takaya, Y.; Ogasawara, M.; Hayashi, T.; Sakai, M.; Miyaura, N. J. Am. Chem. Soc. 1998, 120, 5579–5580.
- [10] Lautens, M.; Renaud, J.-L.; Hiebert, S. J. Am. Chem. Soc. 2000, 122, 1804–1805.

