# Angewandte <br> Eine Zeitschrijt der Gesellschaft Deutscher Chemiker Chemie 

## Supporting Information

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# On the Origin of the Haouamine Alkaloids 

Noah Z. Burns and Phil S. Baran*<br>Contribution from the Department of Chemistry, The Scripps Research Institute, 10550<br>North Torrey Pines Road, La Jolla, California 92037

General Procedures. All reactions were carried out under a nitrogen atmosphere with dry solvents under anhydrous conditions, unless otherwise noted. Dry tetrahydrofuran (THF), triethylamine (TEA), dichloromethane (DCM), methanol $(\mathrm{MeOH})$, dimethylformamide (DMF), and benzene were obtained by passing commercially available pre-dried, oxygen-free formulations through activated alumina columns. Yields refer to chromatographically and spectroscopically ( ${ }^{1} \mathrm{H}$ NMR) homogeneous materials, unless otherwise stated. Reagents were purchased at the highest commercial quality and used without further purification, unless otherwise stated. Reactions were monitored by thin-layer chromatography (TLC) carried out on 0.25 mm E. Merck silica gel plates ( $60 \mathrm{~F}-254$ ) using UV light as visualizing agent and $p$ anisaldehyde in ethanol/aqueous $\mathrm{H}_{2} \mathrm{SO}_{4} / \mathrm{CH}_{3} \mathrm{CO}_{2} \mathrm{H}$ and heat as developing agents. NMR spectra were recorded on either a Bruker DRX 600, DRX 500 or an AMX 400 and calibrated using residual undeuterated solvent as an internal reference. The following abbreviations were used to explain the multiplicities: $\mathrm{s}=\operatorname{singlet}, \mathrm{d}=\operatorname{doublet}, \mathrm{t}=$ triplet, $\mathrm{q}=$ quartet, $\mathrm{dd}=$ doublet of doublets, $\mathrm{m}=$ multiplet, $\mathrm{b}=$ broad. IR spectra were recorded on a Perkin-Elmer Spetrum BX spectrometer. High resolution mass spectra (HRMS)
were recorded on an Agilent Mass spectrometer (at Scripps) using ESI-TOF (electrospray ionization-time of flight) or a ThermoFinnigan Mass spectrometer (at UCSD) using FAB (fast atom bombardment), or EI (electron impact). Low resolution mass spectra (LRMS) were recorded on an Agilent (at Scripps) or ThermoFinnigan Mass spectrometer (at UCSD) GC-MS. Melting points (m.p.) are uncorrected and were recorded on a FisherJohns 12-144 melting point apparatus. Circular dichroism measurements were obtained on an AVIV model 62DS spectrophotometer.

## General procedure for the abnormal Chichibabin pyridine synthesis:





A solution of an arylacetaldehyde $(2.0 \mathrm{mmol})$ and an alkylamine $(0.5 \mathrm{mmol}$, as either the hydrochloride salt or free base) in a solvent of water, water/1,4-dioxane, or DCM (0.1 $0.5 \mathrm{M})$ is treated with ytterbium triflate $(0.25 \mathrm{mmol})$, and the reaction mixture is stirred vigorously for 24 h . The reaction mixture is diluted with water ( 20 mL ), extracted with EtOAc $(2 \times 20 \mathrm{~mL})$, dried over $\mathrm{MgSO}_{4}$, filtered, and concentrated. Purification by flash column chromatography (silica gel, $99: 1 \rightarrow 95: 5 \mathrm{DCM} / \mathrm{MeOH}$ ) affords the product 3,5diarylpyridinium.

## Phenyl pyridinium 7:



The general procedure with phenylacetaldehyde and benzylammonium chloride on 0.21 mmol scale in water ( 0.5 M ) afforded the title pyridinium as a white solid ( $66 \mathrm{mg}, 66 \%$ );
m.p. $=159-160{ }^{\circ} \mathrm{C}$;
$\mathbf{R}_{\mathrm{f}}=0.30($ silica gel, $9: 1 \mathrm{DCM} / \mathrm{MeOH}) ;$
IR (film) $\boldsymbol{v}_{\text {max }} 3063,1252(\mathrm{~s}), 1156,1027(\mathrm{~s}), 901,758,695,636(\mathrm{~s}) \mathrm{cm}^{-1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \delta 9.15(\mathrm{~s}, 2 \mathrm{H}), 8.52(\mathrm{~s}, 1 \mathrm{H}), 7.74(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.60$
$-7.59(\mathrm{~m}, 2 \mathrm{H}), 7.53-7.48(\mathrm{~m}, 6 \mathrm{H}), 7.34-7.32(\mathrm{~m}, 3 \mathrm{H}), 6.08(\mathrm{~s}, 2 \mathrm{H})$;
${ }^{13} \mathbf{C}$-APT NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\delta 141.9,140.4,139.9,132.8,132.7,130.6,130.0$,
129.8, 129.6, 129.5, 127.6, 65.1;

HRMS (ESI) calcd. for $\mathrm{C}_{24} \mathrm{H}_{20} \mathrm{~N}\left[\mathrm{M}^{+}\right]$322.1596, found 322.1596.
meta-Methoxyphenyl pyridinium 9:


The general procedure with $m$-methoxyphenylacetaldehyde and benzylammonium chloride on a 0.12 mmol scale in $1: 1$ water $/ 1,4$-dioxane $(0.5 \mathrm{M})$ afforded the product as clear needles ( $43 \mathrm{mg}, 68 \%$ );
m.p. $=139-140^{\circ} \mathrm{C}$;
$\mathbf{R}_{\mathrm{f}}=0.28$ (silica gel, 9:1 DCM/MeOH);
IR (film) $\boldsymbol{v}_{\text {max }} 1587,1480,1255(\mathrm{~s}), 1159,1027(\mathrm{~s}), 909,783,754,728,691,636(\mathrm{~s}) \mathrm{cm}^{-}$ ${ }^{1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(600 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \delta 9.17(\mathrm{~s}, 2 \mathrm{H}), 8.49(\mathrm{~s}, 1 \mathrm{H}), 7.59(\mathrm{dd}, J=3.1,5.9 \mathrm{~Hz}, 2$ H), $7.41(\mathrm{t}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.32(\mathrm{~m}, 3 \mathrm{H}), 7.28-7.26(\mathrm{~m}, 4 \mathrm{H}), 7.01(\mathrm{dd}, J=2.3,8.3$ $\mathrm{Hz}, 2 \mathrm{H}), 6.11$ ( $\mathrm{s}, 2 \mathrm{H}$ ), 3.88 (s, 6 H );
${ }^{13}$ C-APT NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 160.6,141.8,140.5,140.1,134.1,132.9,130.9$, 129.9, 129.6, 129.4, 119.8, 116.7, 112.6, 65.0, 55.7;

HRMS (ESI) calcd. for $\mathrm{C}_{26} \mathrm{H}_{24} \mathrm{NO}_{2}\left[\mathrm{M}^{+}\right]$382.1807, found 382.1803.
para-Bromophenyl pyridinium 10:



The general procedure with $p$-bromophenylacetaldehyde and benzylammonium chloride on 0.098 mmol scale in $2: 1$ 1,4-dioxane/water $(0.3 \mathrm{M})$ yielded the title pyridinium as colorless cubes ( $40.3 \mathrm{mg}, 65 \%$ );
m.p. $=241-242{ }^{\circ} \mathrm{C}$;
$\mathbf{R}_{\mathrm{f}}=0.38$ (silica gel, 9:1 DCM/MeOH);
IR (film) $\boldsymbol{v}_{\text {max }} 1594,1476,1252(\mathrm{~s}), 1222,1159,1027,1005,817,710,636 \mathrm{~cm}^{-1}$;
${ }^{1} \mathbf{H}$ NMR ( $600 \mathbf{M H z}, \mathbf{C D}_{3} \mathbf{C N}$ ) $\delta 8.99(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 2 \mathrm{H}), 8.87(\mathrm{t}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.79$ $(\mathrm{d}, J=8.6 \mathrm{~Hz}, 4 \mathrm{H}), 7.73(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 4 \mathrm{H}), 7.56(\mathrm{~m}, 2 \mathrm{H}), 7.49-7.47(\mathrm{~m}, 3 \mathrm{H}), 5.84$ (s, 2 H );
${ }^{13}$ C-APT NMR ( $\mathbf{1 5 0} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{C N}$ ) $\delta 142.4,141.8,141.5,133.8,133.6,133.3,130.8$, 130.6, 130.3, 130.0, 125.6, 65.9;

HRMS (ESI) calcd. for $\mathrm{C}_{24} \mathrm{H}_{18} \mathrm{Br}_{2} \mathrm{~N}\left[\mathrm{M}^{+}\right]$477.9806, found 477.9805 .
meta-Trifluoromethylpyridinium 11:


The general procedure with $m$-trifluoromethylphenylacetaldehyde and benzylammonium chloride on a 0.17 mmol scale in $1: 1$ water/1,4-dioxane ( 0.4 M ) yielded the title pyridinium as clear needles ( $55.6 \mathrm{mg}, 54 \%$ );
m.p. $=58-60^{\circ} \mathrm{C}$;
$\mathbf{R}_{\mathrm{f}}=0.40$ (silica gel, 9:1 DCM/MeOH);

IR film $\boldsymbol{v}_{\text {max }} 2354,1329,1252(\mathrm{~s}), 1159,1123(\mathrm{~s}), 1071,1027(\mathrm{~s}), 802,699,636 \mathrm{~cm}^{-1}$;
${ }^{1} \mathbf{H}$ NMR ( $600 \mathbf{M H z}, \mathbf{C D}_{2} \mathbf{C l}_{\mathbf{2}}$ ) $9.22(\mathrm{~s}, 2 \mathrm{H}), 8.68(\mathrm{~s}, 1 \mathrm{H}), 8.05(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.98$ $(\mathrm{s}, 2 \mathrm{H}), 7.82(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.74(\mathrm{t}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.60-7.59(\mathrm{~m}, 2 \mathrm{H}), 7.42-$ 7.40 (m, 3 H), 6.11 (s, 2 H);
${ }^{13}$ C-APT NMR ( $150 \mathrm{MHz}, \mathrm{CD}_{2} \mathrm{Cl}_{2}$ ) 142.1, 141.7, 141.5, 134.3, 132.8, 132.5 (d, $J=32.8$
$\mathrm{Hz}), 132.1,131.2,130.8,130.3,130.1,127.9,125.0,124.2(\mathrm{q}, J=272.7 \mathrm{~Hz}), 66.2$;
HRMS (ESI) calcd. for $\mathrm{C}_{26} \mathrm{H}_{18} \mathrm{~F}_{6} \mathrm{~N}\left[\mathrm{M}^{+}\right] 458.1343$, found 458.1339.
o-Methylphenyl pyridinium 12:



The general procedure with o-methylphenylacetaldehyde and benzylammonium chloride on a 0.2 mmol scale in $1: 1$ water/1,4-dioxane $(0.5 \mathrm{M})$ yielded the title pyridinium as clear needles (62 mg, 62\%);
m.p. $=164-167^{\circ}$;
$\mathbf{R}_{\mathrm{f}}=0.4$ (silica gel, 9:1 DCM/MeOH);
IR (film) $\boldsymbol{v}_{\text {max }} 1476,1454,1255(\mathrm{~s}), 1222,1152,1027(\mathrm{~s}), 761,635(\mathrm{~s}) \mathrm{cm}^{-1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathbf{M H z}, \mathbf{C D C l}_{3}\right) \delta 8.78(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 2 \mathrm{H}), 8.20(\mathrm{t}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.57-$ $7.55(\mathrm{~m}, 2 \mathrm{H}), 7.43-7.41(\mathrm{~m}, 3 \mathrm{H}), 7.39-7.35(\mathrm{~m}, 4 \mathrm{H}), 7.32-7.30(\mathrm{~m}, 4 \mathrm{H}), 6.06(\mathrm{~s}$, 2H), 2.29 (s, 6 H);
${ }^{13}$ C-APT NMR (500 MHz, $\mathbf{C D C l}_{3}$ ) $\delta 145.4,142.0,141.9,135.5,133.1,132.3,131.2$, $130.22,130.17$ (2 C), 129.9, 129.8, 127.0, 65.3, 20.1;

HRMS (ESI) calcd. for $\mathrm{C}_{26} \mathrm{H}_{24} \mathrm{~N}\left[\mathrm{M}^{+}\right]$350.1909, found 350.1908.

## Trideuteropyridinium 13 :



The general procedure with phenylacetaldehyde- $d$ and benzylammonium chloride in water ( 0.5 M ) yielded the title pyridinium as a white solid.
m.p. $=159-160^{\circ} \mathrm{C}$;
$\mathbf{R}_{\mathrm{f}}=0.33$ (silica gel, 9:1 DCM/MeOH)
IR (film) $\boldsymbol{v}_{\text {max }} 2354,2332,1554,1418,1252,1152,1027,750,695,636,514 \mathrm{~cm}^{-1}$;
${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ): $\delta 7.75-7.73(\mathrm{~m}, 4 \mathrm{H}), 7.61-7.59(\mathrm{~m}, 2 \mathrm{H}), 7.52-7.44$ (m, 6 H ), $7.32(\mathrm{t}, J=3.2 \mathrm{~Hz}, 3 \mathrm{H}), 6.05(\mathrm{~s}, 2 \mathrm{H})$;
${ }^{13} \mathbf{C}$-APT NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta 141.7,132.8,132.6,130.6,130.0,129.8,129.6$, $129.5,127.6,64.9$.

HRMS (ESI) calcd. for $\mathrm{C}_{24} \mathrm{H}_{17} \mathrm{D}_{3} \mathrm{~N}\left[\mathrm{M}^{+}\right]$325.1781, found 325.1775.

## Dideuteropyridine 14:



The general procedure with phenylacetaldehyde and benzylammonium chloride- $d_{2}$ yielded the title pyridinium as a white solid.
m.p. $=154-155^{\circ} \mathrm{C}$;
$\mathbf{R}_{\mathrm{f}}=0.24$ (silica gel, $9: 1 \mathrm{DCM} / \mathrm{MeOH}$ );
IR (film) $\boldsymbol{v}_{\text {max }} 3067,1595,1498,1475,1442,1257(\mathrm{~s}), 1225,1157,1029(\mathrm{~s}), 759,696$ $\mathrm{cm}^{-1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \delta 9.16(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 2 \mathrm{H}), 8.52(\mathrm{t}, J=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.75-$ $7.73(\mathrm{~m}, 4 \mathrm{H}), 7.61-7.59(\mathrm{~m}, 2 \mathrm{H}), 7.54-7.47(\mathrm{~m}, 6 \mathrm{H}), 7.34(\mathrm{t}, J=3.1 \mathrm{~Hz}, 3 \mathrm{H})$;
${ }^{13}$ C-APT NMR ( $\mathbf{1 5 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta 141.9,140.3,139.9,132.68,132.65,130.6,130.0$, $129.8,129.6,129.4,127.5 ;$

HRMS (ESI) calcd. for $\mathrm{C}_{24} \mathrm{H}_{18} \mathrm{D}_{2} \mathrm{~N}\left[\mathrm{M}^{+}\right] 324.1719$, found 324.1717.


The general procedure with 4-(tert-butylphenyl)-phenylacetaldehyde and benzylamine on a 0.054 mmol scale in DCM ( 0.1 M ) yielded the known 4-(tert-butylphenyl)benzaldehyde ${ }^{1}$ ( $6 \mathrm{mg}, 0.7$ equiv based on isolated product) and the title pyridinium as a white solid (26 mg, 65\%);
m.p. $>300^{\circ} \mathrm{C}$;
$\mathbf{R}_{\mathrm{f}}=0.39$ (silica gel, 9:1 DCM/MeOH);
IR (film) $\boldsymbol{\nu}_{\text {max }} 3067,2923,1595,1498,1475,1452,1442,1412,1341,1254$ (s), 1224, 1155, 1029, 758, $695 \mathrm{~cm}^{-1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \delta 9.18(\mathrm{~s}, 2 \mathrm{H}), 8.45(\mathrm{~s}, 1 \mathrm{H}), 7.79(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 4 \mathrm{H}), 7.73$ $(\mathrm{d}, J=8.4 \mathrm{~Hz}, 4 \mathrm{H}), 7.62-7.6-(\mathrm{m}, 2 \mathrm{H}), 7.52(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 4 \mathrm{H}), 7.47(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 4$ H), $7.40-7.36(\mathrm{~m}, 3 \mathrm{H}), 6.16(\mathrm{~s}, 2 \mathrm{H}), 1.36(\mathrm{~s}, 18 \mathrm{H})$;
${ }^{13}$ C-APT NMR (150 MHz, $\left.\mathbf{C D C l}_{3}\right) \delta 151.4,143.1,141.2,139.9,138.9,136.2,132.9$, $131.0,130.0,129.7,129.4,128.0,127.9,126.6,126.0,65.3,34.6,31.3$;

HRMS (ESI) calcd. for $\left[\mathrm{M}^{+}\right] 586.3474$ found 586.3469.

[^0]
## Scheme S1.




## Pyridinium 23:



The general procedure with m-methoxyphenylacetaldehyde 24 and 2-(meta-methoxyphenyl)-ethylamine hydrochloride $\mathbf{2 5}$ on a 0.54 mmol scale in water ( 0.5 M ) yielded the title pyridinium as a white solid (131 mg, 57\%); ${ }^{2}$

[^1]m.p. $=158-160^{\circ} \mathrm{C}$;
$\mathbf{R}_{\mathrm{f}}=0.22$ (silica gel, 9:1 DCM/MeOH);
IR (film) $\boldsymbol{v}_{\text {max }} 2937,1735,1594,1583,1488,1454,1255(\mathrm{~s}), 1152,1027,783,691,636$ $\mathrm{cm}^{-1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathbf{~ M H z}\right.$, CDCl $\left._{3}\right) \delta 8.76(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 2 \mathrm{H}), 8.50(\mathrm{t}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.41$ $(\mathrm{t}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.17-7.14(\mathrm{~m}, 3 \mathrm{H}), 7.10(\mathrm{t}, J=2.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.03(\mathrm{dd}, J=2.4,8.3$ $\mathrm{Hz}, 2 \mathrm{H}), 6.74(\mathrm{dd}, J=2.4,8.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.68(\mathrm{t}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.63(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1$ $\mathrm{H}), 5.15(\mathrm{t}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.89(\mathrm{~s}, 6 \mathrm{H}), 3.72(\mathrm{~s}, 3 \mathrm{H}), 3.30(\mathrm{t}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H})$;
${ }^{13}$ C-APT NMR ( $\mathbf{1 5 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\delta 160.7,141.4,140.5,139.9,136.9,134.0,130.8$, $130.2,121.1,119.6,116.7,114.2,113.5,112.4,63.6,55.7,55.3,37.9$;

HRMS (ESI) calcd. for $\mathrm{C}_{28} \mathrm{H}_{28} \mathrm{NO}_{3}\left[\mathrm{M}^{+}\right] 426.2069$, found 426.2062.

## Tetrahydropyridine S1:



To a solution of pyridinium $23(316 \mathrm{mg}, 0.74 \mathrm{mmol})$ in methanol $(3.7 \mathrm{~mL})$ and DCM (3.7 mL ) at $0{ }^{\circ} \mathrm{C}$ was added cerium trichloride ( $276 \mathrm{mg}, 0.74 \mathrm{mmol}, 1.0$ equiv). Sodium borohydride was added CAUTIOUSLY ( $560 \mathrm{mg}, 14.8 \mathrm{mmol}, 20.0$ equiv). A distinct bright color is observed upon addition. The reaction mixture was stirred for 20 min , and acetic acid ( 3.7 mL ) was then added CAUTIOUSLY dropwise. A disappearance in color
signified the completion of the reaction (more sodium borohydride was added if complete decoloration did not occur within 30 min ). The reaction mixture was then diluted with $1.0 \mathrm{M} \mathrm{NaOH}(40 \mathrm{~mL})$, and the aqueous layer was extracted with EtOAc $(2 \times 50 \mathrm{~mL})$. The combined organics were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. Flash column chromatography (silica gel, 5:1 hexanes/EtOAc) afforded the product as a clear oil (229 $\mathrm{mg}, 72 \%)$.
$\mathbf{R}_{\mathrm{f}}=0.29$ (silica gel, 3:1 hexanes/EtOAc)
IR (film) $\boldsymbol{v}_{\text {max }} 2930,2827,1598,1580,1484,1462,1451,1429,1285,1259,1148,1045$, $780,691 \mathrm{~cm}^{-1}$;
${ }^{1} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ) $\delta 7.28(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.25(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.21$ $(\mathrm{t}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.02(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.96(\mathrm{~s}, 1 \mathrm{H}), 6.90(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.86$ - $6.75(\mathrm{~m}, 6 \mathrm{H}), 6.19(\mathrm{~s}, 1 \mathrm{H}), 3.83(\mathrm{~s}, 3 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.79(\mathrm{~m}, 1 \mathrm{H})$, $3.68(\mathrm{~d}, J=15.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.37(\mathrm{dt}, J=2.5,15.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.16(\mathrm{dd}, J=5.5,11.2 \mathrm{~Hz}, 1$ H), $2.88(\mathrm{dd}, J=11.3,9.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.81(\mathrm{dd}, J=9.7,10.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.43(\mathrm{dd}, J=9.0$, 11.1 Hz, 1 H );
${ }^{13}$ C-APT NMR ( $\mathbf{1 5 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\delta 159.7,159.6$ (2C), 145.2, 141.9, 141.3, 136.1, $129.42,129.35,129.34,126.2,121.1,120.5,117.7,114.5,114.0,112.5,111.8,111.3$, $111.2,59.8,58.3,55.23,55.19,55.1,54.6,43.2,33.9$;

HRMS (ESI) calcd. for $\mathrm{C}_{28} \mathrm{H}_{31} \mathrm{NO}_{3}\left[\mathrm{M}+\mathrm{H}^{+}\right] 430.2377$, found 430.2370.

## 2-Benzyltetrahydropyridine 27:



To a solution of tetrahydropyridine $\mathbf{S 1}(51.8 \mathrm{mg}, 0.12 \mathrm{mmol})$ in acetone ( $600 \mu \mathrm{~L}$ ) was added 3-methoxylbenzylbromide ( $167 \mu \mathrm{~L}, 1.2 \mathrm{mmol}, 10.0$ equiv), and the reaction mixture was heated in a $60{ }^{\circ} \mathrm{C}$ oil bath overnight. Concentration and flash column chromatography (silica gel, 95:5 DCM/MeOH) afforded the quaternary ammonium salt as a white solid ( $75 \mathrm{mg}, 99 \%$ ). This salt was dissolved in THF ( 4.0 mL ) and placed in a $45{ }^{\circ} \mathrm{C}$ oil bath. LHMDS ( $720 \mu \mathrm{~L}, 0.5 \mathrm{M}, 0.36 \mathrm{mmol}, 3.0$ equiv) was added at once, and after 5 min the reaction was cooled and quenched with water $(10 \mathrm{~mL})$. The organic layer was extracted twice with EtOAc ( 20 mL ), dried over $\mathrm{MgSO}_{4}$, filtered, and concentrated. Flash column chromatography (silica gel, 99:1 to $95: 5$ benzene/EtOAc) afforded the product (as a 7:3 mixture of diastereomers) that appeared as a slightly yellow oil ( 32 mg , 48\%).
$\mathbf{R}_{\mathrm{f}}=0.38$ (silica gel, 4:1 hexanes/EtOAc)
IR (film) $\boldsymbol{v}_{\max } 2998,2938,1833,1598,1583,1486,1464,1453,1433,1286,1260(\mathrm{~s})$, 1151. 1047, 777, $696 \mathrm{~cm}^{-1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(600 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \delta 7.31(\mathrm{dt}, J=2.9,8.0 \mathrm{~Hz}, 1.3 \mathrm{H}), 7.25-7.20(\mathrm{~m}, 0.7 \mathrm{H})$, $7.19-7.15(\mathrm{~m}, 1.7 \mathrm{H}), 7.10(\mathrm{t}, J=7.8 \mathrm{~Hz}, 0.3 \mathrm{H}), 7.01(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 0.7 \mathrm{H}), 6.95-6.92$ (m, 1 H), $6.87-6.61$ (m, 10.3 H ), 6.12 (s, 0.7 H ), 6.05 ( $\mathrm{s}, 0.3 \mathrm{H}), 4.05(\mathrm{~s}, 0.3 \mathrm{H}), 3.90(\mathrm{~d}$,
$J=8.8 \mathrm{~Hz}, 0.7 \mathrm{H}), 3.85-3.74(\mathrm{~m}, 12.7 \mathrm{H}), 3.67(\mathrm{t}, J=8.1 \mathrm{~Hz}, 0.7 \mathrm{H}), 3.46(\mathrm{~s}, 0.3 \mathrm{H})$, 3.33 (dd, $J=11.4,5.0 \mathrm{~Hz}, 0.3 \mathrm{H}), 3.10-3.02(\mathrm{~m}, 1 \mathrm{H}), 2.98-2.90(\mathrm{~m}, 2 \mathrm{H}), 2.85-2.73$ (m, 2 H ), $2.69-2.58(\mathrm{~m}, 2 \mathrm{H})$;
${ }^{13}$ C-APT NMR (150 MHz, $\mathbf{C D C l}_{3}$ ) $\delta$ 159.72, 159.67, 159.51, 159.45, 159.44, 159.4, $159.1,158.8,145.9,145.2,142.5,142.4,142.3,142.2,142.0,141.39,141.37,140.4$, $129.6,129.5,129.3,129.2,128.84,128.79,128.4,127.4,122.2,121.6,121.1,121.0$, $120.4,120.3,119.1,118.7,115.3,115.0,114.3,114.2,113.9,113.8,112.5,112.4,112.2$, $111.5,111.3,111.24,111.19,111.1,111.0,61.7,61.0,59.0,56.9,55.8,55.24,55.18$, $55.14,55.09,55.01,54.97,54.5,52.2,41.9,38.8,37.3,35.3,33.8 ;$

HRMS (ESI) calcd. for $\mathrm{C}_{36} \mathrm{H}_{40} \mathrm{NO}_{4}\left[\mathrm{M}+\mathrm{H}^{+}\right]$550.2957, found 550.2945.

Note- The following compound was synthesized in an analogous manner (substituting 2-bromo-3-methoxybenzyl bromide for 3-methoxybenzyl bromide) and its structure proven by x-ray crystallographic analysis:



## 2-Benzylpyridinium 22:



To a solution of $27(100 \mathrm{mg}, 0.18 \mathrm{mmol})$ in $\mathrm{DCM}(1.8 \mathrm{~mL})$ was added N bromoacetamide ( $55 \mathrm{mg}, 0.40 \mathrm{mmol}, 2.2$ equiv) at room temperature. The reaction mixture was stirred for 1.5 h and quenched with sat. aq. $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{4}(10 \mathrm{~mL})$. The organic layer was extracted with EtOAc $(2 \times 20 \mathrm{~mL})$, dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. Flash column chromatography (99:1 to $95: 5 \mathrm{DCM} / \mathrm{MeOH}$ ) afforded the pyridinium bromide as a yellowish solid ( $90 \mathrm{mg}, 79 \%$ ). This compound was dissolved in DCM ( 1.0 mL ), and treated with silver triflate $(40.6 \mathrm{mg}, 0.16 \mathrm{mmol}, 1.1$ equiv). The reaction mixture was stirred for 1 h and quenched with water ( 10 mL ). The aqueous layer was extracted twice with EtOAc ( 20 mL ), dried over $\mathrm{MgSO}_{4}$, filtered, and concentrated. Flash column chromatography (silica gel, 99:5 DCM/MeOH) afforded the pyridinium triflate as a yellowish semisolid ( $97 \mathrm{mg}, 97 \%$ ).
$\mathbf{R}_{\mathrm{f}}=0.32$ (silica gel, 9:1 DCM/MeOH);
IR (film) $\boldsymbol{v}_{\text {max }} 2926,1600,1584,1490,1468,1258(\mathrm{~s}), 1224,1153,1030,787,698 \mathrm{~cm}^{-1}$;
${ }^{1} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ) $\delta 8.96(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.36(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.41$ $(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.37(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.26(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.20(\mathrm{t}, J=7.9 \mathrm{~Hz}$, $1 \mathrm{H}), 7.16-7.14(\mathrm{~m}, 2 \mathrm{H}), 7.04(\mathrm{ddd}, J=8.3,2.4,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.01(\mathrm{ddd}, J=8.4,2.5$, $0.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.93$ (ddd, $J=7.5,1.4,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.89(\mathrm{~m}, 1 \mathrm{H}), 6.83-6.79(\mathrm{~m}, 2 \mathrm{H})$,
$5.00(\mathrm{t}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.29(\mathrm{~s}, 2 \mathrm{H}), 3.91(\mathrm{~s}, 3 \mathrm{H}), 3.761(\mathrm{~s}, 3 \mathrm{H}), 3.760(\mathrm{~s}, 3 \mathrm{H}), 3.75$ ( $\mathrm{s}, 3 \mathrm{H}$ ), $3.02(\mathrm{t}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H} 0$;
${ }^{13}$ C-APT NMR ( $\mathbf{1 5 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta 160.7,160.5,160.2,159.9,152.1,144.0,143.2$, $139.0,136.8,136.7,136.4,133.5,130.8,130.7,130.4,130.3,121.1,120.7,119.8,119.6$, $116.9,115.7,114.6,114.1,114.0,113.4,112.9,112.3,60.1,55.8,55.4,55.33,55.32$, 37.1, 35.6.
${ }^{13}$ C-APT NMR ( $\mathbf{1 5 0} \mathbf{~ M H z}, \mathbf{C D}_{2} \mathbf{C l}_{2}$ ): $\delta 161.3,161.1,160.9,160.6,152.6,144.8,144.4$, $144.3,139.6,137.4,137.1,136.8,134.2,131.4,131.3,131.0,130.9,121.7,121.3,120.3$, $120.1,117.0,116.0,115.2,114.8,114.6,113.8,113.5,113.1,60.8,56.3,55.95,55.88$, 55.8, 37.7, 36.1.

HRMS (ESI) calcd. for $\mathrm{C}_{36} \mathrm{H}_{36} \mathrm{NO}_{4}\left[\mathrm{M}^{+}\right] 546.2644$, found 546.2635.

## Scheme S2.




## Diol 29:



To a solution of unstable (gradually decomposes upon exposure to air/moisture, as such it is stored in a frozen benzene solution) indene $28(2.78 \mathrm{~g}, 11.0 \mathrm{mmol}$, synthesized as previously described ${ }^{3}$ ) in $t$ - $\mathrm{BuOH}(55 \mathrm{~mL}$ ) was added sequentially: methanesulfonamide ( $5.2 \mathrm{~g}, 55.0 \mathrm{mmol}, 5.0$ equiv), (DHQD) ${ }_{2} \operatorname{PHAL}(428.0 \mathrm{mg}, 0.55 \mathrm{mmol}, 0.05$ equiv), $\mathrm{K}_{2} \mathrm{CO}_{3}\left(4.56 \mathrm{~g}, 33.0 \mathrm{mmol}, 3.0\right.$ equiv), and $\mathrm{H}_{2} \mathrm{O}(55.0 \mathrm{~mL})$. This solution was cooled to 0 ${ }^{\circ} \mathrm{C}$ and treated with potassium ferricyanide ( $10.9 \mathrm{~g}, 33.0 \mathrm{mmol}, 3.0$ equiv) followed by osmium tetroxide ( $2.5 \%$ in $t$ - $\mathrm{BuOH} ; 1.38 \mathrm{~mL}, 0.11 \mathrm{mmol}, 0.01$ equiv). The reaction flask was placed in a $5{ }^{\circ} \mathrm{C}$ cold room and stirred for 44 h . The reaction was quenched with saturated aqueous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(100 \mathrm{~mL})$, removed from the cold room, and allowed to warm to room temperature and stir for 1 h . The aqueous layer was then extracted with EtOAc ( $3 \times 150 \mathrm{~mL}$ ), dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. Flash column chromatography (silica gel, $4: 1 \rightarrow 3: 1$ hexanes/EtOAc) afforded the diol as a brown semisolid ( $3.10 \mathrm{~g}, 98 \%$ ). To the resulting diol mixture was added hexanes ( 40 mL ) and EtOAc ( 10 mL ), followed by heating to a brief reflux with a heat gun in order to assure complete dissolution. The hot solution was then seeded with a sample of crystalline racemic diol and allowed to cool to room temperature. Clear needles of racemic diol crystallized, and the supernatant was then removed and concentrated to afford the enantiomerically pure title diol as a thick oil $(1.89 \mathrm{~g}, 60 \%)$.

[^2]$\mathbf{R}_{\mathrm{f}}=0.24$ (silica gel, 2:1 hexanes/EtOAc);
$[\boldsymbol{\alpha}]_{\mathrm{D}}=+43.4^{\circ}\left(\mathrm{CHCl}_{3}, c 0.53\right) ;$
IR (film) $\boldsymbol{v}_{\text {max }} 3447$ (br), 2937, 2835, 1589, 1480 (s), 1262 (s), 1078 (s), 1044 (s) $\mathrm{cm}^{-1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathbf{M H z}\right.$, CDCl $\left._{3}\right) \delta 7.32(\mathrm{t}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.19(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.95$ $(\mathrm{d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.83-6.78(\mathrm{~m}, 3 \mathrm{H}), 6.70-6.68(\mathrm{~m}, 1 \mathrm{H}), 4.29(\mathrm{~d}, J=4.6 \mathrm{~Hz}, 1 \mathrm{H})$, $4.15\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{D}_{2} \mathrm{O}\right.$ exchangeable), $3.77(\mathrm{~s}, 3 \mathrm{H}), 3.75(\mathrm{~s}, 3 \mathrm{H}), 3.48\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{D}_{2} \mathrm{O}\right.$ exchangeable), $3.02(\mathrm{dd}, J=16.5,4.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.90(\mathrm{~d}, J=16.4 \mathrm{~Hz}, 1 \mathrm{H})$;
${ }^{13}$ C-APT NMR (150 MHz, $\mathbf{C D C l}_{3}$ ) $\delta 159.6,156.7,145.8,143.6,130.5,130.0,129.1$, $118.3,118.2,112.5,111.8,109.1,85.8,80.5,55.3,55.1,38.1$;

HRMS (ESI) calcd. for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{O}_{4}\left[\mathrm{M}+\mathrm{Na}^{+}\right]$309.1097, found 309.1092.

## $\alpha$-Hydroxy ketone 30 :



To a solution of diol 29 ( $599 \mathrm{mg}, 2.09 \mathrm{mmol}$ ) in $\mathrm{DCM}(10.5 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was added sequentially: saturated aqueous $\mathrm{NaHCO}_{3}(4.2 \mathrm{~mL})$, potassium bromide $(12.0 \mathrm{mg}, 0.10$ mmol, 0.05 equiv), TEMPO ( $16.0 \mathrm{mg}, 0.10 \mathrm{mmol}, 0.05$ equiv), and aqueous sodium hypochlorite ( $6.2 \mathrm{~mL}, 4.18 \mathrm{mmol}, 2.0$ equiv). The reaction mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min then quenched with saturated aqueous $\mathrm{NaHSO}_{4}(10 \mathrm{~mL})$. The ice bath was removed and the reaction mixture was allowed to warm to room temperature. The aqueous layer was extracted with EtOAc $(3 \times 30 \mathrm{~mL})$, and the combined organic layers
were washed once with brine ( 20 mL ). The organic layer was dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. Flash column chromatography (silica gel, 3:1 hexanes/EtOAc) afforded the title ketone as an off-white solid ( $570 \mathrm{mg}, 96 \%$ ).
m.p. $=117-119^{\circ} \mathrm{C}$;
$\mathbf{R}_{\mathbf{f}}=0.24$ (silica gel, 2:1 hexanes/EtOAc);
$[\boldsymbol{\alpha}]_{\mathbf{D}}=+37.9^{\circ}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}, c 0.43\right) ;$
IR (film) $\boldsymbol{v}_{\text {max }} 3467$ (br) 3011, 1744 (s), 1587 (s), 1484 (s), 1291 (s), 1053 (s) $\mathrm{cm}^{-1}$;
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{M H z}$, CDCl $\left._{3}\right) \delta 7.39(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.19(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.01$ $(\mathrm{d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.97(\mathrm{~m}, 1 \mathrm{H}), 6.89(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{~m}, 1 \mathrm{H}), 6.77(\mathrm{~m}, 1$ H), $3.78(\mathrm{~s}, 3 \mathrm{H}), 3.77(\mathrm{~s}, 3 \mathrm{H}), 3.67(\mathrm{~d}, J=21.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.57(\mathrm{~s}, 1 \mathrm{H}), 3.49(\mathrm{~d}, J=21.6$ Hz, 1 H);
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 5 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\delta 210.4,160.0,156.8,141.5,137.5,130.8,129.6,129.5$, $117.9,117.4,113.7,111.3,110.0,82.2,55.5,55.2,40.2$;

HRMS (ESI) calcd. for $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{O}_{4}\left[\mathrm{M}+\mathrm{Na}^{+}\right]$307.0941, found 307.0929.

## Allyl stannane 31:



The corresponding allyl iodide ( $1.41 \mathrm{~g}, 3.99 \mathrm{mmol}$, synthesized as previously described ${ }^{3}$ ) was dissolved in dry THF ( 20 mL ) and treated with bis(tributyltin) $(2.0 \mathrm{~mL}, 3.99 \mathrm{mmol}$, 1.0 equiv) followed by $\mathrm{Pd}_{2} \mathrm{dba}_{3} \cdot \mathrm{CHCl}_{3}$ ( $207 \mathrm{mg}, 0.20 \mathrm{mmol}, 0.05$ equiv). Argon was bubbled through the solution for 25 min and it was then heated to $55^{\circ}$ for 3 h . Sat. aq.
$\mathrm{NaHCO}_{3}(50 \mathrm{~mL})$ was added and the mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(2 \times 100 \mathrm{~mL})$. The combined organic layers were washed with water ( 50 mL ) and brine ( 50 mL ) and dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. Flash column chromatography (triethylamine neutralized silica gel, $100 \%$ hexanes) afforded the title allyl stannane as a clear oil (1.78 g, 86\%).
$\mathbf{R}_{\mathrm{f}}=0.15$ (silica gel, hexanes);
IR (film) $\boldsymbol{v}_{\text {max }} 2953$ (s), 1560, 1462 (s), 1300, 1231 (s), 1016 (s) $\mathrm{cm}^{-1}$;
${ }^{1} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ) $\delta 7.41(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.74(\mathrm{~d}, J=3.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.66$ (dd, $J=8.7,3.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), $4.99(\mathrm{~m}, 1 \mathrm{H}), 4.71(\mathrm{~m}, 1 \mathrm{H}), 3.78(\mathrm{~s}, 3 \mathrm{H}), 2.23(\mathrm{~s}, 2 \mathrm{H}), 1.35$ (m, 6 H), 1.23 (qd, $J=14.6,7.3,6 \mathrm{H}), 0.84(\mathrm{t}, J=7.3 \mathrm{~Hz}, 9 \mathrm{H}), 0.76$ (m, 6 H$)$;
${ }^{13} \mathbf{C}$ NMR ( $150 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ) $\delta 158.8,150.4,146.3,133.7,115.8,114.1,112.4,110.9$, 55.6, 29.1, 27.5, 20.1, 13.8, 9.8.

## Ketone 3:



To a solution of $\alpha$-hydroxy ketone $\mathbf{3 0}(490 \mathrm{mg}, 1.72 \mathrm{mmol})$ and allyl stannane $\mathbf{3 1}(1.78 \mathrm{~g}$, 3.45 mmol, 2.0 equiv) in THF ( 11.5 mL ) at $0{ }^{\circ} \mathrm{C}$ was added indium(III) trifluoromethanesulfonate ( $1.16 \mathrm{~g}, 2.06 \mathrm{mmol}, 1.2$ equiv). After complete dissolution, the reaction mixture was allowed to warm to room temperature and stirred for 1.5 h . Saturated aqueous sodium potassium tartrate $(20 \mathrm{~mL})$ and $\mathrm{EtOAc}(50 \mathrm{~mL})$ were added,
and the organic layer was washed with brine $(20 \mathrm{~mL})$. The combined aqueous layers were extracted with EtOAc ( 50 mL ), and the combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered, and concentrated. Flash column chromatography (silica gel, 6:1 $\rightarrow 4: 1$ hexanes/EtOAc) afforded the homoallylic diol as a clear oil ( $755 \mathrm{mg}, 86 \%$ ). To a solution of this diol ( $755 \mathrm{mg}, 1.48 \mathrm{mmol}$ ) in $\mathrm{DCM}(14.8 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ was added boron triflouride diethyl etherate ( $204 \mu \mathrm{~L}, 1.62 \mathrm{mmol}, 1.1$ equiv) dropwise. After stirring for 10 $\min$ at $0{ }^{\circ} \mathrm{C}$, sat. aq. $\mathrm{NaHCO}_{3}(20 \mathrm{~mL})$ and EtOAc $(20 \mathrm{~mL})$ were added. The organic layer was washed with brine ( 20 mL ), and the combined aqueous layers were extracted with EtOAc (20 mL). The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. Flash column chromatography (silica gel, 9:1 $\rightarrow$ 6:1 hexanes/EtOAc) afforded the title ketone as a white solid ( $603 \mathrm{mg}, 83 \%$ ).
m.p. $=88-90^{\circ} \mathrm{C}$;
$\mathbf{R}_{\mathrm{f}}=0.39$ (silica gel, $1: 1$ hexanes $/ \mathrm{Et}_{2} \mathrm{O}$ );
$[\boldsymbol{\alpha}]_{\mathrm{D}}=+6.8^{\circ}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}, c 0.5\right) ;$
IR (film) $\boldsymbol{v}_{\text {max }} 1749$ (s), 1586 (s), 1482 (s), 1463 (s), 1289 (s), 1264 (s), 1050 (s), 1016 $\mathrm{cm}^{-1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(600 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \delta 7.27(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.23(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.16$ $(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.91(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.76(\mathrm{dd}, J=8.4$, $6.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.74(\mathrm{dd}, J=8.1,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.53(\mathrm{dd}, J=8.8,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.51(\mathrm{~d}, J=$ $10.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.63(\mathrm{~d}, J=3.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.24(\mathrm{~s}, 1 \mathrm{H}), 4.88(\mathrm{~d}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.91(\mathrm{~d}, J$ $=13.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.74(\mathrm{~s}, 3 \mathrm{H}), 3.72(\mathrm{~d}, J=13.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.57(\mathrm{~s}, 3 \mathrm{H}), 3.48(\mathrm{~d}, J=22.5$ Hz, 1 H ), 3.42 (m, 3 H ), 3.11 (d, $J=22.5 \mathrm{~Hz}, 1 \mathrm{H}$ );
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 5 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\delta 215.2,159.6,158.0,157.3,147.5,143.7,142.4,138.5$, $132.7,130.3,129.6,129.3,120.4,119.2,116.8,115.6,114.6,113.2,112.7,111.8,109.2$, 63.4, 55.3, 55.2, 54.6, 42.9, 41.3;

HRMS (ESI) calcd. for $\mathrm{C}_{27} \mathrm{H}_{25} \mathrm{BrO}_{4}\left[\mathrm{M}+\mathrm{H}^{+}\right]$493.1014, found 493.0988 .

The following compounds were spectroscopically identical to those we reported previously ${ }^{1}$ except for optical rotation data given below:



$$
[\boldsymbol{\alpha}]_{\mathrm{D}}=-13.5^{\circ}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}, c 0.45\right)
$$

$$
[\boldsymbol{\alpha}]_{\mathrm{D}}=+18.8^{\circ}\left(2: 1 \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}, c 0.47\right)
$$




$$
[\boldsymbol{\alpha}]_{\mathbf{D}}=+66.3^{\circ}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}, c 0.40\right)
$$

(+)-Haouamine A: $[\boldsymbol{\alpha}]_{\mathrm{D}}=+45.8^{\circ}(\mathrm{MeOH}, c 0.05)$

Figure S1. Circular dichroism spectrum of natural-(-)- and synthetic-(+)-haouamine A.


Scheme S3.


## Homoallylic diol S2:



To a solution of racemic $\alpha$-hydroxy ketone $\mathbf{3 0}(50 \mathrm{mg}, 0.18 \mathrm{mmol})$ in toluene ( 3.6 mL ) at $-78^{\circ}$ was added allylmagnesium bromide ( $880 \mu \mathrm{~L}, 1.0 \mathrm{M}, 0.88 \mathrm{mmol}, 5.0$ equiv). The reaction mixture was allowed to stir at $-78^{\circ}$ for 20 min and then warmed to room temperature before being quenched with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(5 \mathrm{~mL})$. This mixture was extracted with EtOAc $(2 \times 20 \mathrm{~mL})$, and the combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered, and concentrated. Flash column chromatography (silica gel, 8:1 hexanes/EtOAc) afforded the title diol as a clear oil ( $47 \mathrm{mg}, 82 \%$ ).
$\mathbf{R}_{\mathrm{f}}=0.32$ (silica gel, $1: 1$ hexanes $/ \mathrm{Et}_{2} \mathrm{O}$ );
IR (film) $v_{\text {max }} 3509$ (br), 2937, 2834, 1592, 1480 (s), 1260 (s), 1085, $1039 \mathrm{~cm}^{-1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(600 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \delta 7.29(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.17(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.93$ $(\mathrm{d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.85(\mathrm{~m}, 1 \mathrm{H}), 6.78(\mathrm{dd}, J=8.2,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.75(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1$ H), $6.65(\mathrm{~m}, 1 \mathrm{H}), 5.93(\mathrm{~m}, 1 \mathrm{H}), 5.04(\mathrm{dd}, J=9.2,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.96(\mathrm{dd}, J=17.1,0.6$ $\mathrm{Hz}, 1 \mathrm{H}), 4.25(\mathrm{~s}, 1 \mathrm{H}), 3.77(\mathrm{~s}, 3 \mathrm{H}), 3.71(\mathrm{~s}, 3 \mathrm{H}), 3.63(\mathrm{~s}, 1 \mathrm{H}), 2.97(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1$ H), $2.91(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.19(\mathrm{dd}, J=14.1,6.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.65(\mathrm{dd}, J=14.1,7.9 \mathrm{~Hz}$, 1 H );
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 5 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\delta 159.6,156.3,144.6,142.7,134.7,132.7,130.3,129.2$, $118.5,118.3,117.6,112.3,112.2,109.4,87.7,84.5,55.5,55.3,42.7,41.6 ;$

HRMS (ESI) calcd. for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{O}_{4}\left[\mathrm{M}+\mathrm{Na}^{+}\right]$349.1416, found 349.1399.

## Homoallylic diol S2- $\boldsymbol{d}_{\mathbf{2}}$ :



Diol S2 $(66 \mathrm{mg}, 0.20 \mathrm{mmol})$ was dissolved in $\mathrm{MeOH}(3 \mathrm{~mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$ and cooled to $-78^{\circ}$. Ozone was bubbled through the solution for 5 min , and the reaction mixture was then quenched with $\mathrm{Me}_{2} \mathrm{~S}(2 \mathrm{~mL})$ and warmed to room temperature. $\mathrm{Et}_{2} \mathrm{O}$ $(20 \mathrm{~mL})$ and $5 \% \mathrm{NaHCO}_{3}(3 \mathrm{~mL})$ were added and the organic phase was washed with $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL})$ and brine ( 10 mL ), dried over $\mathrm{MgSO}_{4}$, filtered, and concentrated. The
crude aldehyde was dissolved in dry THF ( 3 mL ) and a solution of $\mathrm{Ph}_{3} \mathrm{P}=\mathrm{CD}_{2}(2.0 \mathrm{mmol}$, 10 equiv) in dry THF ( 4.4 mL ) was added at RT. The resulting solution was stirred for 1.5 hours and then quenched with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(5 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ and diluted with $\mathrm{Et}_{2} \mathrm{O}(20 \mathrm{~mL})$. The organic phase was washed with brine $(20 \mathrm{~mL})$ and the aqueous phase was extracted with $\mathrm{Et}_{2} \mathrm{O}(20 \mathrm{~mL})$. The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered, and concentrated. Flash column chromatography (silica gel, 3:1 hexanes/EtOAc) afforded the title diol as colorless/white needles ( $34 \mathrm{mg}, 51 \%$ ).
m.p. $=124^{\circ} \mathrm{C}$;
$\mathbf{R}_{\mathrm{f}}=0.67$ (silica gel, $1: 1$ hexanes/EtOAc);
IR (film) $\boldsymbol{v}_{\text {max }} 3509$ (br), 1592, 1480 (s), 1261 (s), 1083, 1042 (s) $\mathrm{cm}^{-1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(600 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \delta 7.29(\mathrm{~m}, 1 \mathrm{H}), 7.17(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.94(\mathrm{~d}, J=7.5$ $\mathrm{Hz}, 1 \mathrm{H}), 6.86(\mathrm{~m}, 1 \mathrm{H}), 6.78(\mathrm{dd}, J=8.2,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.63(\mathrm{~m}, 1 \mathrm{H}), 6.75(\mathrm{~d}, J=8.2$ $\mathrm{Hz}, 1 \mathrm{H}), 5.94(\mathrm{~m}, 1 \mathrm{H}), 5.04(\mathrm{~m}, 0.2 \mathrm{H}[80 \% \mathrm{D}]), 4.96(\mathrm{~m}, 0.2 \mathrm{H}[80 \% \mathrm{D}]), 4.27(\mathrm{~s}, 1 \mathrm{H})$, $3.77(\mathrm{~s}, 3 \mathrm{H}), 3.70(\mathrm{~s}, 3 \mathrm{H}), 3.66(\mathrm{~s}, 1 \mathrm{H}), 2.98(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.92(\mathrm{~d}, J=16.2 \mathrm{~Hz}$, $1 \mathrm{H}), 2.20(\mathrm{dd}, J=14.1,6.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.66(\mathrm{dd}, J=14.1,7.9 \mathrm{~Hz}, 1 \mathrm{H})$;
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 5 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\delta 159.4,156.1,144.4,142.5,134.3,132.5,130.1,129.0$, $118.3,118.06,118.05,112.1,112.0,109.2,87.4,84.3,55.3,55.1,42.5,41.3 ;$

GC/MS calcd. for $\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{D}_{2} \mathrm{O}_{4}\left[\mathrm{M}^{+}\right] 328$, found 328 .

## Ketone S3:



To a solution of diol $\mathbf{S 2}(40.0 \mathrm{mg}, 12 \mathrm{mmol})$ in $\mathrm{DCM}(1.2 \mathrm{~mL})$ at $0^{\circ}$ was added boron triflouride diethyl etherate ( $17 \mu \mathrm{~L}, 0.13 \mathrm{mmol}, 1.1$ equiv) dropwise. After stirring for 10 min at $0^{\circ}$, sat. aq. $\mathrm{NaHCO}_{3}(5 \mathrm{~mL})$ and $\mathrm{EtOAc}(20 \mathrm{~mL})$ were added. The organic layer was washed with brine ( 10 mL ), and the combined aqueous layers were extracted with EtOAc ( 20 mL ). The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. Flash column chromatography (silica gel, $9: 1$ hexanes $/ \mathrm{Et}_{2} \mathrm{O}$ ) afforded the title ketone as a clear oil ( $34 \mathrm{mg}, 92 \%$ ).
$\mathbf{R}_{\mathrm{f}}=0.35$ (silica gel, $4: 1$ hexanes $\left./ \mathrm{Et}_{2} \mathrm{O}\right) ;$
IR (film) $\boldsymbol{v}_{\text {max }} 2915,2841,1746(\mathrm{~s}), 1583,1480,1292,1248(\mathrm{~s}), 1053,769,695 \mathrm{~cm}^{-1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \delta 7.33(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.17(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.97$ $(\mathrm{d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.77-6.74(\mathrm{~m}, 3 \mathrm{H}), 5.39-5.30(\mathrm{~m}, 1 \mathrm{H})$, $5.01(\mathrm{dd}, J=17.0,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.82(\mathrm{dd}, J=10.1,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.75(\mathrm{~s}, 3 \mathrm{H}), 3.59(\mathrm{~d}, J$ $=22.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.35(\mathrm{~d}, J=22.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.26(\mathrm{dd}, J=13.0,7.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.18(\mathrm{dd}, J=$ $13.0,6.9 \mathrm{~Hz}, 1 \mathrm{H})$;
${ }^{13} \mathbf{C}$ NMR ( $150 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ) $\delta 215.4,159.5,157.0,142.0,138.0,134.1,130.3,129.3$, $129.2,119.0,117.8,117.1,113.0,111.8,109.6,63.3,55.2,55.1,42.9,39.4 ;$

HRMS (ESI) calcd. for $\mathrm{C}_{20} \mathrm{H}_{21} \mathrm{O}_{3}\left[\mathrm{M}+\mathrm{H}^{+}\right]$309.1491, found 309.1485.

## Ketone S3- $\boldsymbol{d}_{\mathbf{2}}$ :



To a solution of $\mathbf{S} 2-\boldsymbol{d}_{\mathbf{2}}(25 \mathrm{mg}, 0.076 \mathrm{mmol})$ in $\mathrm{DCM}(1.0 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was added boron triflouride diethyl etherate ( $11 \mu \mathrm{~L}, 0.084 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) dropwise. After stirring for 10 $\min$ at $0^{\circ} \mathrm{C}$, sat. aq. $\mathrm{NaHCO}_{3}(5 \mathrm{~mL})$ and $\mathrm{EtOAc}(10 \mathrm{~mL})$ were added. The organic layer was washed with brine ( 10 mL ), and the combined aqueous layers were extracted with EtOAc (10 mL). The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. Flash column chromatography (silica gel, $2: 1$ hexanes $/ \mathrm{Et}_{2} \mathrm{O}$ ) afforded the title ketone as a white film ( $21.2 \mathrm{mg}, 90 \%$ ).
$\mathbf{R}_{\mathrm{f}}=0.58$ (silica gel, $1: 1$ hexanes $/ \mathrm{Et}_{2} \mathrm{O}$ );
IR (film) $\boldsymbol{v}_{\text {max }} 2936,1749$ (s), 1584 (s), 1482 (s), 1289 (s), 1252 (s), 1143, $1053 \mathrm{~cm}^{-1}$;
${ }^{1} \mathbf{H} \operatorname{NMR}\left(600 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \delta 7.33(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.17(\mathrm{t}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.97$ $(\mathrm{d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.76(\mathrm{~m}, 1 \mathrm{H}), 5.34(\mathrm{~m}, 1 \mathrm{H}), 5.00(\mathrm{~m}, 0.2$ H [80\% D] ), $4.82(\mathrm{~m}, 0.2 \mathrm{H}[80 \% \mathrm{D}]), 3.75(\mathrm{~s}, 6 \mathrm{H}), 3.59(\mathrm{~d}, J=22.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.36(\mathrm{~d}, J$ $=22.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.26(\mathrm{dd}, J=13.0,7.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.18(\mathrm{dd}, J=12.9,6.8 \mathrm{~Hz}, 1 \mathrm{H})$;
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 5 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\delta 215.4,159.5,157.0,142.0,138.0,134.0,133.9,130.3$, $129.3,129.2,119.0,117.0,113.0,111.8,109.6,63.3,55.2,55.1,42.9,39.3$;

HRMS (ESI) calcd. for $\mathrm{C}_{20} \mathrm{H}_{18} \mathrm{D}_{2} \mathrm{O}_{3}\left[\mathrm{M}+\mathrm{H}^{+}\right]$311.1616, found 311.1609.
















































[^0]:    ${ }^{1}$ Yang, J.; Gabriele, B.; Belvedere, S.; Huang, Y.; Breslow, R. J. Org. Chem. 2002, 67, 5057-5067.

[^1]:    ${ }^{2}$ Poupon and coworkers (Gravel, E.; Poupon, E.; Hocquemiller, R. Chem. Commun. 2007, 719 - 721) conducted this reaction in DCM with the free amine. Identical results are obtained regardless of the procedure (vide infra).

[^2]:    ${ }^{3}$ P. S. Baran, N. Z. Burns, J. Am. Chem. Soc. 2006, 128, 3908 - 3909.

