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# Nucleophilicities and Carbon Basicities of Pyridines

Frank Brotzel, Bernhard Kempf, Thomas Singer, Hendrik Zipse and Herbert Mayr\*

*Department Chemie und Biochemie*

*Ludwig-Maximilians-Universität München*

*Butenandtstraße 5-13 (Haus F)*

*81377 München (Germany)*

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## Determination of Rate and Equilibrium Constants for Pyridines

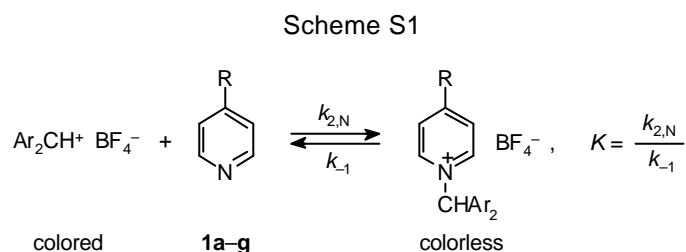
**Instrumentation.** The rates of slow reactions ( $\tau_{1/2} > 10$  s) were determined by using a J&M TIDAS diode array spectrophotometer, which was controlled by Labcontrol Spectacle software and connected to a Hellma 661.502-QX quartz Suprasil immersion probe (5 mm light path) via fiber optic cables and standard SMA connectors.

UV-Vis kinetic measurements of rapid reactions ( $\tau_{1/2} < 10$  s) were performed on a Hi-Tech SF-61DX2 stopped-flow spectrophotometer system. The kinetic experiments were initiated by fast mixing equal volumes of solutions of the nucleophile and the diarylcarbenium salt. The pseudo-first-order rate constants  $k_{\text{obs}}$  for the reactions were obtained from at least five runs at each nucleophile concentration.

The temperature of solutions during all kinetic studies was kept constant ( $\pm 0.2$  °C) by using a circulating bath thermostat and monitored with a thermocouple probe.

For the evaluation of rate constants, absorption-time curves at wavelengths in proximity to the absorption maxima of the diarylcarbenium ions  $\text{Ar}_2\text{CH}^+$  were fitted to the single exponential function  $A = A_0 \exp(-k_{\text{obs}} t) + C$ .

**Determination of the equilibrium constants in dichloromethane.** For the determination of the equilibrium constants  $K$ , the pyridine was added to a solution of the benzhydrylium tetrafluoroborate. After reaching the equilibrium (Scheme S1),



the absorption of the remaining benzhydrylium ions ( $\text{Ar}_2\text{CH}^+$ ) was determined photometrically. The corresponding concentration  $[\text{EI}]$  was calculated from a calibration curve. The equilibrium constants were calculated from the Equation (S1), where  $[\text{EI}]_0$  is the initial concentration of the benzhydrylium ions in the solution,  $[\text{EI}]$  is the concentration of the benzhydrylium ions in the equilibrium mixture and  $[\text{Nu}]_0$  is the initial concentration of the pyridine.

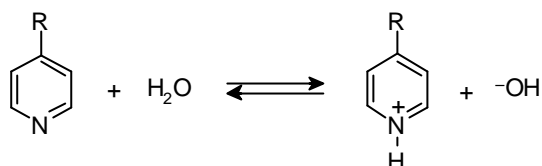
$$K = \frac{[\text{EI}]_0 - [\text{EI}]}{[\text{EI}]( [\text{Nu}]_0 - [\text{EI}]_0 + [\text{EI}] )} \quad (\text{S1})$$

Equilibrium constants  $K$  that have only been measured at 20 °C have been averaged and are given with standard deviations.

When measurements were made at variable temperatures, the equilibrium constants  $K(20\text{ }^\circ\text{C})$  have been derived from extrapolations of the linear plots of  $\ln K$  versus  $1/T$  (van't Hoff plot).

**Kinetic measurements in water.** If pyridine is dissolved in water, the concentration of hydroxide increases by protolysis (Scheme S2). For that reason we have to calculate the concentration of the free pyridines  $[\text{Nu}]_{\text{eff}}$  and of hydroxide  $[\text{OH}^-]$  with the  $pK_B$  of the pyridines.

Scheme S2.



$$K_B = \frac{[\text{pyridinium}] [\text{OH}^-]}{[\text{pyridine}]_{\text{eff}}} \quad (\text{S2})$$

$$[\text{pyridine}]_0 = [\text{pyridine}]_{\text{eff}} + [\text{pyridinium}] = [\text{pyridine}]_{\text{eff}} + [\text{OH}^-] \quad (\text{S3})$$

(S3) in (S2)

$$K_B = \frac{[\text{OH}^-]^2}{[\text{pyridine}]_0 - [\text{OH}^-]} \quad (\text{S4})$$

Solving of the quadratic equation (S4) leads to one logic solution for  $[\text{OH}^-]$ , that is Equation S5 (the one with the "+" in the numerator).

$$[\text{OH}^-] = -\frac{K_B}{2} + \sqrt{\left(\frac{K_B}{2}\right)^2 + K_B[\text{pyridine}]_0} \quad (\text{S5})$$

The rates of the combination reactions were determined by mixing a colored aqueous solution of a benzhydrylium salt with an aqueous solution of a pyridine (>10 equivalents). As the products are colorless, we studied the reactions by UV-Vis spectroscopy and followed the decay of absorption with time [Eq. (S6)].

$$-\frac{d[\text{R}^+]}{dt} = k_{\text{obs}}[\text{R}^+] \quad (\text{S6})$$

The consumption of the benzhydrylium cations may be due to the reaction with the pyridines, hydroxide ions and the solvent water [Eq. (S7)].

$$k_{\text{obs}} = k_{2,\text{N}} [\text{pyridine}]_{\text{eff}} + k_{2,\text{OH}}[\text{OH}^-] + k_{\text{W}} + k_{-1} \quad (\text{S7})$$

$$= k_{1\Psi} + k_{2,\text{OH}}[\text{OH}^-], \text{ with } k_{1\Psi} = k_{2,\text{N}} [\text{pyridine}]_{\text{eff}} + k_{\text{W}} + k_{-1} \text{ and } [\text{OH}^-] \text{ from eq. (5)}$$

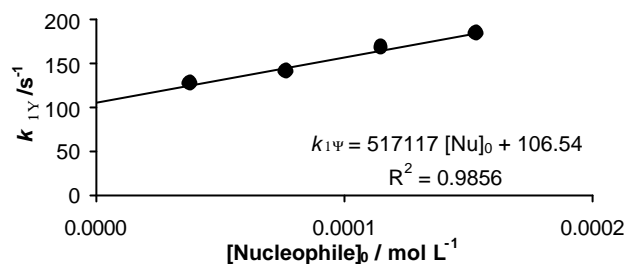
The pyridines are usually used in more than 10-fold excess over the benzhydrylium cations in order to run the reactions under pseudo first-order conditions. The concentrations of the pyridines as well as that of hydroxide remains therefore constant during the reactions. With the already published second-order rate constants  $k_{2,\text{OH}}$  for the reactions of hydroxide with benzhydrylium ions<sup>[S1]</sup> and the first-order rate constants  $k_{\text{W}}$  for the reactions of water with benzhydrylium ions,<sup>[S2]</sup> we get the second-order rate constants for the reactions of the pyridines with the benzhydrylium ions  $k_{2,\text{N}}$  from a plot of  $k_{1\Psi}$  versus  $[\text{pyridine}]_{\text{eff}}$ .

**Determination of the equilibrium constants in water.** We were not able to directly measure the equilibrium constants  $K$  (Scheme S1) in water. When an aqueous solution of the benzhydrylium salt was combined with the pyridine, the UV/Vis spectrum for the benzhydrylium ion did not show a constant end absorption, because of the slow reactions of the benzhydrylium ions with hydroxide and water in the presence of pyridine. However, the ratio of the second-order rate constants  $k_{2,\text{N}}$  for the forward reaction and the first order rate constant  $k_{-1}$  for the backward reaction, gave equilibrium constants, which are less accurate than the directly measured ones and are written in parentheses.

## 4-Chloropyridine (1a)

Reaction of 4-chloropyridine (1a) with (pfa)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>2</sub>Cl<sub>2</sub> (at 20 °C, J&M, detection at 601 nm)

[(pfa) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> /[EI] <sub>0</sub>	k <sub>1Ψ</sub> / s <sup>-1</sup>
4.80 × 10 <sup>-6</sup>	3.82 × 10 <sup>-5</sup>	8	1.28 × 10 <sup>2</sup>
4.80 × 10 <sup>-6</sup>	7.64 × 10 <sup>-5</sup>	16	1.42 × 10 <sup>2</sup>
4.80 × 10 <sup>-6</sup>	1.15 × 10 <sup>-4</sup>	24	1.69 × 10 <sup>2</sup>
4.80 × 10 <sup>-6</sup>	1.53 × 10 <sup>-4</sup>	32	1.85 × 10 <sup>2</sup>



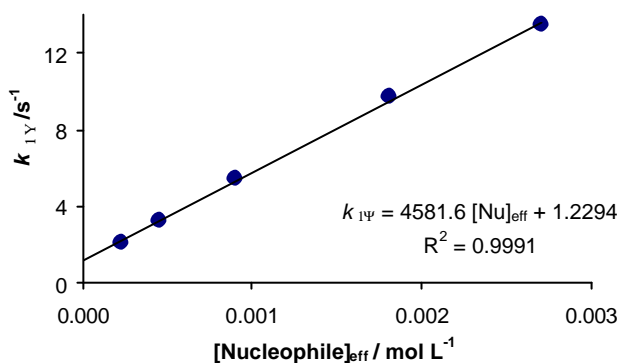
$$k_{2,N} = 5.17 \times 10^5 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{-1} = 1.07 \times 10^2 \text{ s}^{-1}$$

$$K = 4.83 \times 10^3 \text{ M}^{-1}$$

Reaction of 4-chloropyridine (1a) with (mor)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in water (at 20 °C, cosolvent: 9 vol-% CH<sub>3</sub>CN, stopped-flow, detection at 610 nm)

[(mor) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> /[EI] <sub>0</sub>	k <sub>1Ψ</sub> / s <sup>-1</sup>
2.57 × 10 <sup>-5</sup>	2.71 × 10 <sup>-3</sup>	105	13.5
2.57 × 10 <sup>-5</sup>	1.81 × 10 <sup>-3</sup>	70	9.71
2.57 × 10 <sup>-5</sup>	9.03 × 10 <sup>-4</sup>	35	5.46
2.57 × 10 <sup>-5</sup>	4.51 × 10 <sup>-4</sup>	18	3.27
2.57 × 10 <sup>-5</sup>	2.26 × 10 <sup>-4</sup>	9	2.15



$$k_{2,N} = 4.58 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{-1} = 1.23 \text{ s}^{-1}$$

$$K = 3.73 \times 10^3 \text{ M}^{-1}$$

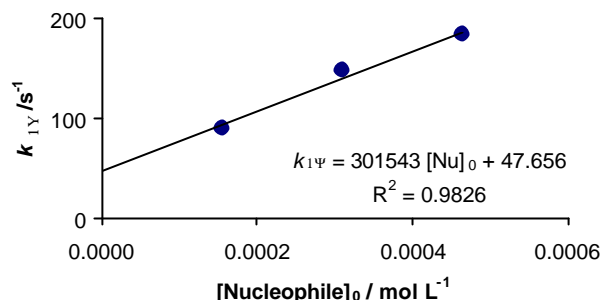
$$k_{2,\text{OH}} = 1.06 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_{\text{B}} = 10.17$$

## Pyridine (1b)

Reaction of pyridine (1b) with (dpa)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>2</sub>Cl<sub>2</sub> (at 20 °C, stopped-flow, detection at 674 nm)

[(dpa) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> /[EI] <sub>0</sub>	k <sub>1Ψ</sub> / s <sup>-1</sup>
7.51 × 10 <sup>-6</sup>	4.64 × 10 <sup>-4</sup>	62	1.84 × 10 <sup>2</sup>
7.51 × 10 <sup>-6</sup>	3.09 × 10 <sup>-4</sup>	41	1.48 × 10 <sup>2</sup>
7.51 × 10 <sup>-6</sup>	1.55 × 10 <sup>-4</sup>	21	90.8



$$k_{2,N} = 3.02 \times 10^5 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{-1} = 47.7 \text{ s}^{-1}$$

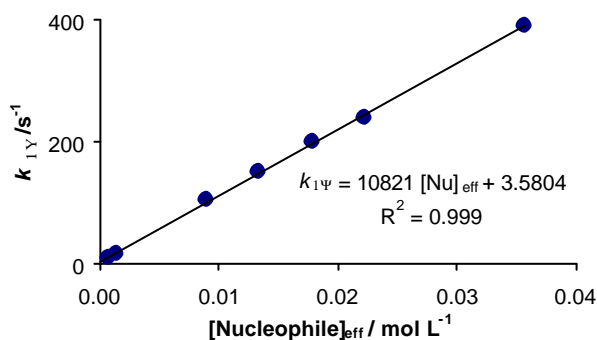
Equilibrium constants for the reaction of pyridine (1b) with (dpa)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>2</sub>Cl<sub>2</sub> (at 20 °C, J&M, detection at 672 nm)

[(dpa) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[(dpa) <sub>2</sub> CH <sup>+</sup> ] / mol L <sup>-1</sup>	K / L mol <sup>-1</sup>
1.08 × 10 <sup>-5</sup>	1.36 × 10 <sup>-4</sup>	2.96 × 10 <sup>-6</sup>	2.07 × 10 <sup>4</sup>
1.65 × 10 <sup>-5</sup>	1.03 × 10 <sup>-4</sup>	5.28 × 10 <sup>-6</sup>	2.30 × 10 <sup>4</sup>
1.18 × 10 <sup>-5</sup>	6.86 × 10 <sup>-5</sup>	9.47 × 10 <sup>-6</sup>	2.32 × 10 <sup>4</sup>

$$K (20 \text{ °C}) = (2.23 \pm 0.11) \times 10^4 \text{ L mol}^{-1}$$

Reaction of pyridine (1b) with (mor)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in water (at 20 °C, cosolvent: 9 vol-% CH<sub>3</sub>CN, stopped-flow, detection at 607 nm)

[(mor) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> /[EI] <sub>0</sub>	k <sub>1Ψ</sub> / s <sup>-1</sup>
2.29 × 10 <sup>-5</sup>	3.57 × 10 <sup>-2</sup>	1560	3.90 × 10 <sup>2</sup>
2.29 × 10 <sup>-5</sup>	2.23 × 10 <sup>-2</sup>	974	2.39 × 10 <sup>2</sup>
2.29 × 10 <sup>-5</sup>	1.79 × 10 <sup>-2</sup>	781	2.00 × 10 <sup>2</sup>
2.29 × 10 <sup>-5</sup>	1.34 × 10 <sup>-2</sup>	585	1.50 × 10 <sup>2</sup>
2.29 × 10 <sup>-5</sup>	8.93 × 10 <sup>-3</sup>	390	1.07 × 10 <sup>2</sup>
2.29 × 10 <sup>-5</sup>	1.43 × 10 <sup>-3</sup>	62	16.4
2.29 × 10 <sup>-5</sup>	7.13 × 10 <sup>-4</sup>	31	8.57



$$k_{2,\text{OH}} = 1.06 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_{\text{B}} = 8.77$$

$$k_{2,N} = 1.08 \times 10^4 \text{ M}^{-1}\text{s}^{-1}$$

Reaction of pyridine (**1b**) with  $(\text{dma})_2\text{CH}^+ \text{BF}_4^-$  in water (at 20 °C, cosolvent: 0.4 vol-%  $\text{CH}_3\text{CN}$ , stopped-flow, detection at 607 nm)

$[(\text{dma})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$7.62 \times 10^{-5}$	$3.57 \times 10^{-2}$	468	41.4
$7.62 \times 10^{-5}$	$2.23 \times 10^{-2}$	293	29.0
$7.62 \times 10^{-5}$	$1.79 \times 10^{-2}$	235	25.7
$7.62 \times 10^{-5}$	$1.34 \times 10^{-2}$	176	21.5
$7.62 \times 10^{-5}$	$8.93 \times 10^{-3}$	117	17.8

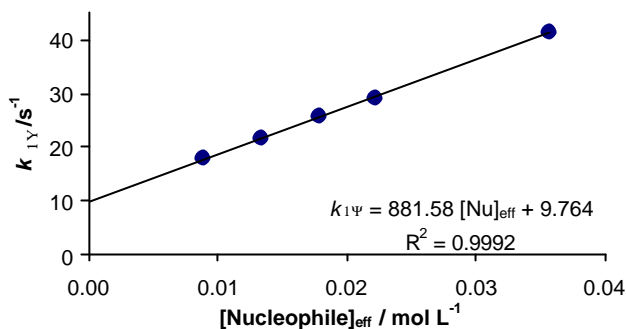
$$k_{2,N} = 8.82 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{-1} = 9.76 \text{ s}^{-1}$$

$$K = 9.04 \times 10^1 \text{ M}^{-1}$$

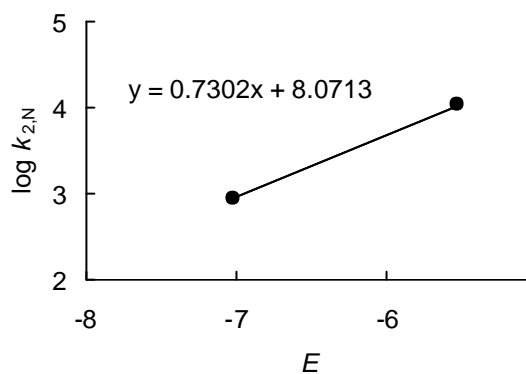
$$k_{2,\text{OH}} = 1.31 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_{\text{B}} = 8.77$$



Reactivity parameters of pyridine (**1b**) in water:  $N = 11.05$ ;  $s = 0.730$

Reference electrophile	$E$ parameter	$k_{2,N}(20^\circ\text{C}) / \text{M}^{-1} \text{s}^{-1}$
$(\text{mor})_2\text{CH}^+$	-5.53	$1.08 \times 10^4$
$(\text{dma})_2\text{CH}^+$	-7.02	$8.82 \times 10^2$

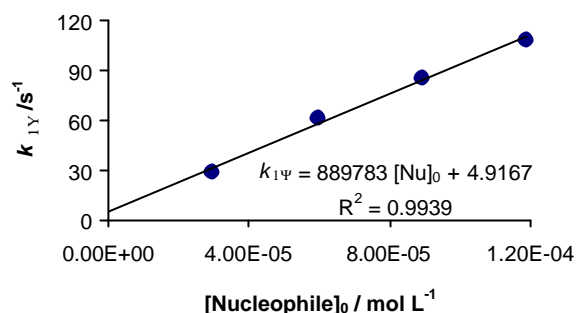




## 4-Methylpyridine (1c)

Reaction of 4-methylpyridine (1c) with  $(\text{dpa})_2\text{CH}^+ \text{BF}_4^-$  in  $\text{CH}_2\text{Cl}_2$  (at 20 °C, stopped-flow, detection at 674 nm)

$[(\text{dpa})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$4.30 \times 10^{-6}$	$1.19 \times 10^{-4}$	28	$1.09 \times 10^2$
$4.30 \times 10^{-6}$	$8.96 \times 10^{-5}$	21	85.7
$4.30 \times 10^{-6}$	$5.97 \times 10^{-5}$	14	61.3
$4.30 \times 10^{-6}$	$2.99 \times 10^{-5}$	7	29.0



$$k_{2,N} = 8.90 \times 10^5 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{-1} = 4.92 \text{ s}^{-1}$$

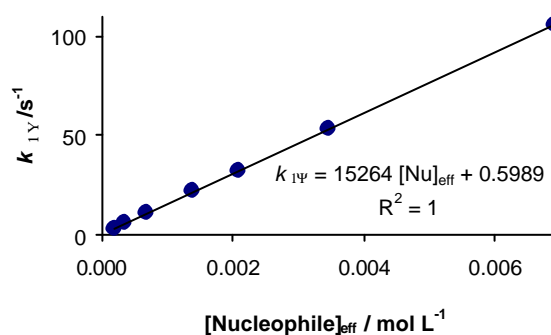
Equilibrium constants for the reaction of 4-methylpyridine (1c) with  $(\text{dpa})_2\text{CH}^+ \text{BF}_4^-$  in  $\text{CH}_2\text{Cl}_2$  (at 20 °C, in  $\text{CH}_2\text{Cl}_2$ , J&M, detection at 672 nm)

$[(\text{dpa})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / / mol L <sup>-1</sup>	$[(\text{dpa})_2\text{CH}^+]$ / mol L <sup>-1</sup>	$K$ / L mol <sup>-1</sup>
$3.55 \times 10^{-5}$	$6.49 \times 10^{-5}$	$2.00 \times 10^{-6}$	$5.34 \times 10^5$
$3.46 \times 10^{-5}$	$3.17 \times 10^{-5}$	$8.63 \times 10^{-6}$	$5.28 \times 10^5$
$3.29 \times 10^{-5}$	$1.81 \times 10^{-5}$	$1.69 \times 10^{-5}$	$4.60 \times 10^5$

$$K (20 \text{ °C}) = (5.07 \pm 0.34) \times 10^5 \text{ L mol}^{-1}$$

Reaction of 4-methylpyridine (1c) with  $(\text{mor})_2\text{CH}^+ \text{BF}_4^-$  in water (at 20 °C, cosolvent: 9 vol-%  $\text{CH}_3\text{CN}$ , stopped-flow, detection at 610 nm)

$[(\text{mor})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$1.72 \times 10^{-5}$	$6.91 \times 10^{-3}$	402	$1.06 \times 10^2$
$1.72 \times 10^{-5}$	$3.45 \times 10^{-3}$	201	53.5
$1.72 \times 10^{-5}$	$2.08 \times 10^{-3}$	121	32.3
$1.72 \times 10^{-5}$	$1.38 \times 10^{-3}$	80	21.8
$1.72 \times 10^{-5}$	$6.89 \times 10^{-4}$	40	10.9
$1.72 \times 10^{-5}$	$3.44 \times 10^{-4}$	20	5.95
$1.72 \times 10^{-5}$	$1.72 \times 10^{-4}$	10	3.08



$$k_{2,N} = 1.53 \times 10^4 \text{ M}^{-1}\text{s}^{-1}$$

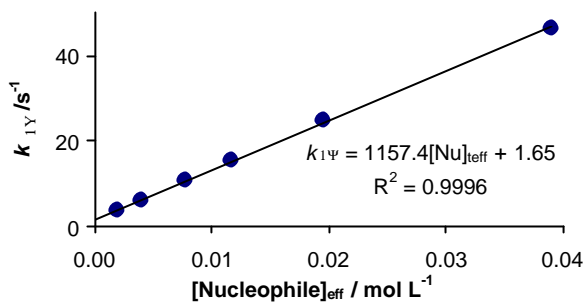
$$k_W = 0.673 \text{ s}^{-1[\text{S}2]}$$

$$k_{2,\text{OH}^-} = 1.06 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$pK_B = 7.94$$

Reaction of 4-methylpyridine (**1c**) with  $(\text{dma})_2\text{CH}^+ \text{BF}_4^-$  in water (at 20 °C, cosolvent: 0.5 vol-%  $\text{CH}_3\text{CN}$ , stopped-flow, detection at 610 nm)

$[(\text{dma})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$4.61 \times 10^{-5}$	$3.91 \times 10^{-2}$	848	46.6
$4.61 \times 10^{-5}$	$1.95 \times 10^{-2}$	423	24.7
$4.61 \times 10^{-5}$	$1.17 \times 10^{-2}$	254	15.4
$4.61 \times 10^{-5}$	$7.81 \times 10^{-3}$	169	10.7
$4.61 \times 10^{-5}$	$3.90 \times 10^{-3}$	85	5.97
$4.61 \times 10^{-5}$	$1.95 \times 10^{-3}$	42	3.66



$$k_{2,\text{OH}} = 1.31 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_{\text{B}} = 7.94$$

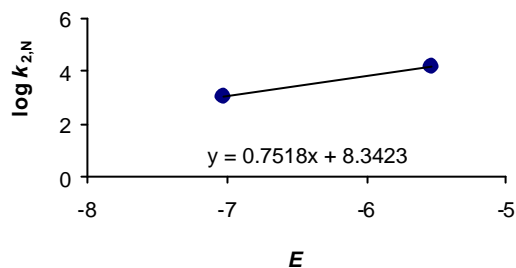
$$k_{2,\text{N}} = 1.16 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{-1} = 1.65 \text{ s}^{-1}$$

$$K = 7.03 \times 10^2 \text{ M}^{-1}$$

Reactivity parameters of 4-methylpyridine (**1c**) in water:  $N = 11.10$ ;  $s = 0.752$

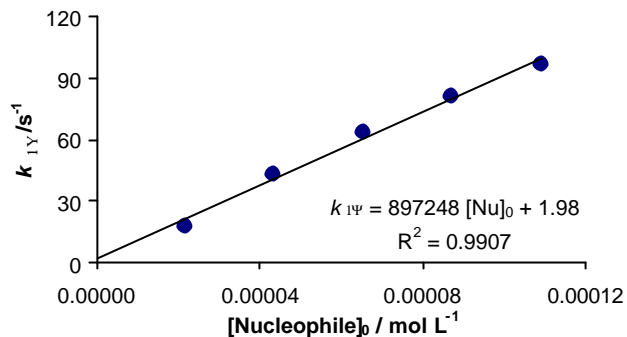
Reference electrophile	$E$ parameter	$k_{2,\text{N}}(20^\circ\text{C}) / \text{M}^{-1} \text{s}^{-1}$
$(\text{mor})_2\text{CH}^+$	-5.53	$1.53 \times 10^4$
$(\text{dma})_2\text{CH}^+$	-7.02	$1.16 \times 10^3$



## 4-Methoxypyridine (**1d**)

Reaction of 4-methoxypyridine (**1d**) with  $(\text{dpa})_2\text{CH}^+ \text{BF}_4^-$  in  $\text{CH}_2\text{Cl}_2$  (at 20 °C, stopped-flow, detection at 674 nm)

$[(\text{dpa})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$4.30 \times 10^{-6}$	$1.09 \times 10^{-4}$	25	97.0
$4.30 \times 10^{-6}$	$8.72 \times 10^{-5}$	20	81.1
$4.30 \times 10^{-6}$	$6.54 \times 10^{-5}$	15	63.7
$4.30 \times 10^{-6}$	$4.36 \times 10^{-5}$	10	43.5
$4.30 \times 10^{-6}$	$2.18 \times 10^{-5}$	5	18.0



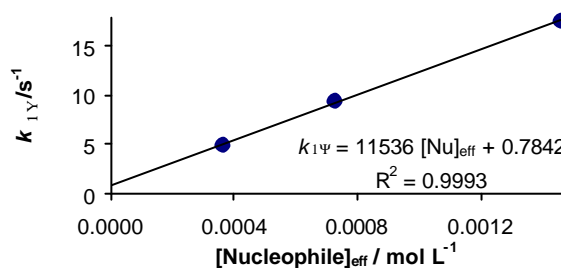
$$k_{2N} = 8.97 \times 10^5 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{-1} = 1.98 \text{ s}^{-1}$$

$$(K = 4.53 \times 10^5 \text{ M}^{-1})$$

Reaction of 4-methoxypyridine (**1d**) with  $(\text{mor})_2\text{CH}^+ \text{BF}_4^-$  in water (at 20 °C, cosolvent: 9 vol-%  $\text{CH}_3\text{CN}$ , stopped-flow, detection at 610 nm)

$[(\text{mor})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$1.72 \times 10^{-5}$	$1.46 \times 10^{-3}$	85	17.6
$1.72 \times 10^{-5}$	$7.29 \times 10^{-4}$	42	9.39
$1.72 \times 10^{-5}$	$3.63 \times 10^{-4}$	21	4.85



$$k_{2N} = 1.15 \times 10^4 \text{ M}^{-1}\text{s}^{-1}$$

$$k_W = 0.673 \text{ s}^{-1} [\text{S2}]$$

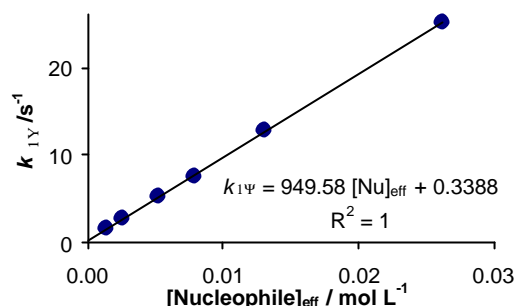
$$k_{2\text{OH}} = 1.06 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_B = 7.42$$

Reaction of 4-methoxypyridine (**1d**) with (dma)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in water (at 20 °C, cosolvent: 0.5 vol-% CH<sub>3</sub>CN, stopped-flow, detection at 610 nm)

[(dma) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> /[EI] <sub>0</sub>	k <sub>1Ψ</sub> / s <sup>-1</sup>
4.61 × 10 <sup>-5</sup>	2.62 × 10 <sup>-2</sup>	568	25.2
4.61 × 10 <sup>-5</sup>	1.31 × 10 <sup>-2</sup>	284	12.8
4.61 × 10 <sup>-5</sup>	7.83 × 10 <sup>-3</sup>	170	7.67
4.61 × 10 <sup>-5</sup>	5.23 × 10 <sup>-3</sup>	113	5.31
4.61 × 10 <sup>-5</sup>	2.60 × 10 <sup>-3</sup>	56	2.82
4.61 × 10 <sup>-5</sup>	1.30 × 10 <sup>-3</sup>	28	1.62

$$k_{2,N} = 9.50 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$



$$k_W = 0.0206 \text{ s}^{-1} \text{ [S2]}$$

$$k_{2,\text{OH}^-} = 1.31 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_B = 7.42$$

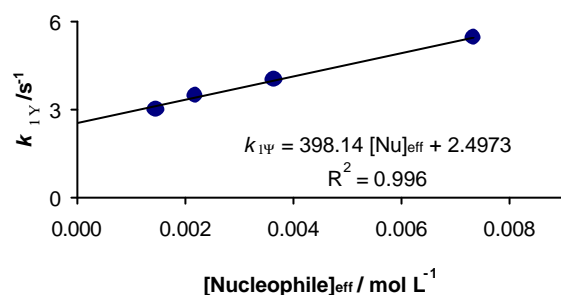
Reaction of 4-methoxypyridine (**1d**) with (pyr)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in water (at 20 °C, cosolvent: 0.5 vol-% CH<sub>3</sub>CN, stopped-flow, detection at 610 nm)

[(pyr) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> /[EI] <sub>0</sub>	k <sub>1Ψ</sub> / s <sup>-1</sup>
2.54 × 10 <sup>-5</sup>	7.35 × 10 <sup>-3</sup>	289	5.41
2.54 × 10 <sup>-5</sup>	3.66 × 10 <sup>-3</sup>	144	3.97
2.54 × 10 <sup>-5</sup>	2.19 × 10 <sup>-3</sup>	86	3.45
2.54 × 10 <sup>-5</sup>	1.46 × 10 <sup>-3</sup>	58	3.01

$$k_{2,N} = 3.98 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{-1} = 2.50 \text{ s}^{-1}$$

$$K = 1.59 \times 10^2 \text{ M}^{-1}$$

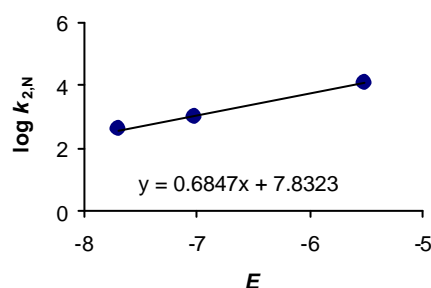


$$k_{2,\text{OH}^-} = 48.5 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_B = 7.42$$

Reactivity parameters of 4-methoxypyridine (**1d**) in water:  $N = 11.44$ ;  $s = 0.685$

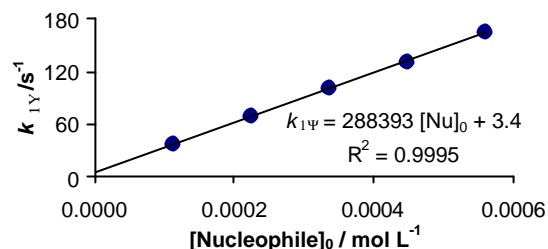
Reference electrophile	E parameter	k <sub>2,N</sub> (20°C) / M <sup>-1</sup> s <sup>-1</sup>
(mor) <sub>2</sub> CH <sup>+</sup>	-5.53	1.15 × 10 <sup>4</sup>
(dma) <sub>2</sub> CH <sup>+</sup>	-7.02	9.50 × 10 <sup>2</sup>
(pyr) <sub>2</sub> CH <sup>+</sup>	-7.69	3.98 × 10 <sup>2</sup>



## 4-Aminopyridine (1e)

Reaction of 4-aminopyridine (1e) with  $(\text{dma})_2\text{CH}^+ \text{BF}_4^-$  in  $\text{CH}_2\text{Cl}_2$  (at 20 °C, stopped-flow, detection at 612 nm)

$[(\text{dma})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$7.80 \times 10^{-6}$	$5.60 \times 10^{-4}$	72	$1.66 \times 10^2$
$7.80 \times 10^{-6}$	$4.48 \times 10^{-4}$	57	$1.31 \times 10^2$
$7.80 \times 10^{-6}$	$3.36 \times 10^{-4}$	43	$1.00 \times 10^2$
$7.80 \times 10^{-6}$	$2.24 \times 10^{-4}$	29	69.0
$7.80 \times 10^{-6}$	$1.12 \times 10^{-4}$	14	35.5



$$k_{2,\text{N}} = 2.88 \times 10^5 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{-1} = 3.40 \text{ s}^{-1}$$

$$K = 8.48 \times 10^4 \text{ M}^{-1}$$

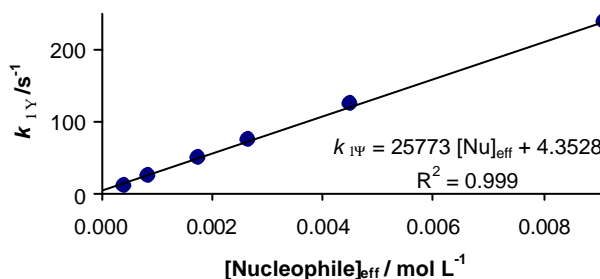
Reaction of 4-aminopyridine (1e) with  $(\text{mor})_2\text{CH}^+ \text{BF}_4^-$  in water (at 20 °C, cosolvent: 9 vol-%  $\text{CH}_3\text{CN}$ , stopped-flow, detection at 610 nm)

$[(\text{mor})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}$ / mol L <sup>-1</sup>	$[\text{OH}^-]$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}/[\text{EI}]_0$	$k_{\text{obs}}$ / s <sup>-1</sup>	$k_{1\Psi,\text{OH}}$ / s <sup>-1</sup>	$k_{1\Psi}$ / s <sup>-1</sup>
$3.65 \times 10^{-5}$	$9.45 \times 10^{-3}$	$9.10 \times 10^{-3}$	$3.46 \times 10^{-4}$	249	$2.37 \times 10^2$	$3.67 \times 10^{-1}$	$2.37 \times 10^2$
$3.65 \times 10^{-5}$	$4.73 \times 10^{-3}$	$4.49 \times 10^{-3}$	$2.43 \times 10^{-4}$	123	$1.24 \times 10^2$	$2.58 \times 10^{-1}$	$1.24 \times 10^2$
$3.65 \times 10^{-5}$	$2.84 \times 10^{-3}$	$2.65 \times 10^{-3}$	$1.87 \times 10^{-4}$	73	74.9	$1.98 \times 10^{-1}$	74.7
$3.65 \times 10^{-5}$	$1.89 \times 10^{-3}$	$1.74 \times 10^{-3}$	$1.51 \times 10^{-4}$	48	50.3	$1.60 \times 10^{-1}$	50.1
$3.65 \times 10^{-5}$	$9.45 \times 10^{-4}$	$8.40 \times 10^{-4}$	$1.05 \times 10^{-4}$	23	24.6	$1.12 \times 10^{-1}$	24.5
$3.65 \times 10^{-5}$	$4.73 \times 10^{-4}$	$4.00 \times 10^{-4}$	$7.26 \times 10^{-5}$	11	11.9	$7.70 \times 10^{-2}$	11.8

$$k_{2,\text{N}} = 2.58 \times 10^4 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{2,\text{OH}} = 1.06 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$pK_{\text{B}} = 4.88$$



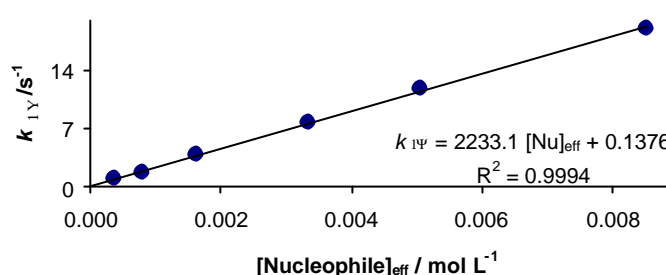
Reaction of 4-aminopyridine (**1e**) with  $(\text{dma})_2\text{CH}^+ \text{BF}_4^-$  in water (at 20 °C, cosolvent: 0.3 vol-%  $\text{CH}_3\text{CN}$ , stopped-flow, detection at 610 nm)

$[(\text{dma})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}$ / mol L <sup>-1</sup>	$[\text{OH}^-]$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}/[\text{E}]_0$	$k_{\text{obs}}$ / s <sup>-1</sup>	$k_{1\Psi, \text{OH}}$ / s <sup>-1</sup>	$k_{1\Psi}$ / s <sup>-1</sup>
$5.95 \times 10^{-5}$	$8.85 \times 10^{-3}$	$8.51 \times 10^{-3}$	$3.35 \times 10^{-4}$	143	19.0	$4.39 \times 10^{-2}$	19.0
$5.95 \times 10^{-5}$	$5.31 \times 10^{-3}$	$5.05 \times 10^{-3}$	$2.58 \times 10^{-4}$	85	11.7	$3.38 \times 10^{-2}$	11.7
$5.95 \times 10^{-5}$	$3.54 \times 10^{-3}$	$3.33 \times 10^{-3}$	$2.10 \times 10^{-4}$	56	7.75	$2.74 \times 10^{-2}$	7.72
$5.95 \times 10^{-5}$	$1.77 \times 10^{-3}$	$1.62 \times 10^{-3}$	$1.46 \times 10^{-4}$	27	3.82	$1.92 \times 10^{-2}$	3.80
$5.95 \times 10^{-5}$	$8.85 \times 10^{-4}$	$7.83 \times 10^{-4}$	$1.02 \times 10^{-4}$	13	1.80	$1.33 \times 10^{-2}$	1.79
$5.95 \times 10^{-5}$	$4.43 \times 10^{-4}$	$3.73 \times 10^{-4}$	$7.01 \times 10^{-5}$	6	$8.44 \times 10^{-1}$	$9.18 \times 10^{-3}$	$8.35 \times 10^{-1}$

$$k_{2, \text{N}} = 2.23 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{2, \text{OH}} = 1.31 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_{\text{B}} = 4.88$$



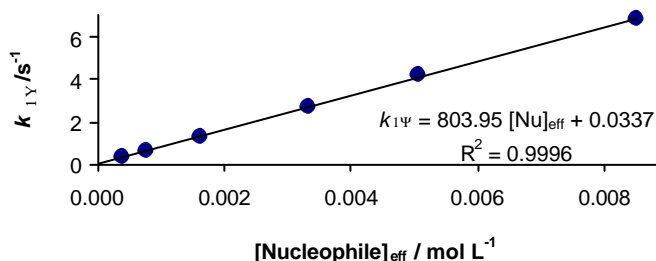
Reaction of 4-aminopyridine (**1e**) with  $(\text{pyr})_2\text{CH}^+ \text{BF}_4^-$  in water (at 20 °C, cosolvent: 0.2 vol-%  $\text{CH}_3\text{CN}$ , stopped-flow, detection at 608 nm)

$[(\text{pyr})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}$ / mol L <sup>-1</sup>	$[\text{OH}^-]$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}/[\text{E}]_0$	$k_{\text{obs}}$ / s <sup>-1</sup>	$k_{1\Psi, \text{OH}}$ / s <sup>-1</sup>	$k_{1\Psi}$ / s <sup>-1</sup>
$1.05 \times 10^{-5}$	$8.85 \times 10^{-3}$	$8.51 \times 10^{-3}$	$3.35 \times 10^{-4}$	811	6.84	$1.62 \times 10^{-2}$	6.82
$1.05 \times 10^{-5}$	$5.31 \times 10^{-3}$	$5.05 \times 10^{-3}$	$2.58 \times 10^{-4}$	481	4.19	$1.25 \times 10^{-2}$	4.18
$1.05 \times 10^{-5}$	$3.54 \times 10^{-3}$	$3.33 \times 10^{-3}$	$2.10 \times 10^{-4}$	317	2.76	$1.02 \times 10^{-2}$	2.75
$1.05 \times 10^{-5}$	$1.77 \times 10^{-3}$	$1.62 \times 10^{-3}$	$1.46 \times 10^{-4}$	155	1.31	$7.10 \times 10^{-3}$	1.30
$1.05 \times 10^{-5}$	$8.85 \times 10^{-4}$	$7.83 \times 10^{-4}$	$1.02 \times 10^{-4}$	75	$6.63 \times 10^{-1}$	$4.93 \times 10^{-3}$	$6.58 \times 10^{-1}$
$1.05 \times 10^{-5}$	$4.43 \times 10^{-4}$	$3.73 \times 10^{-4}$	$7.01 \times 10^{-5}$	36	$3.13 \times 10^{-1}$	$3.40 \times 10^{-3}$	$3.10 \times 10^{-1}$

$$k_{2, \text{N}} = 8.04 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{2, \text{OH}} = 48.5 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_{\text{B}} = 4.88$$



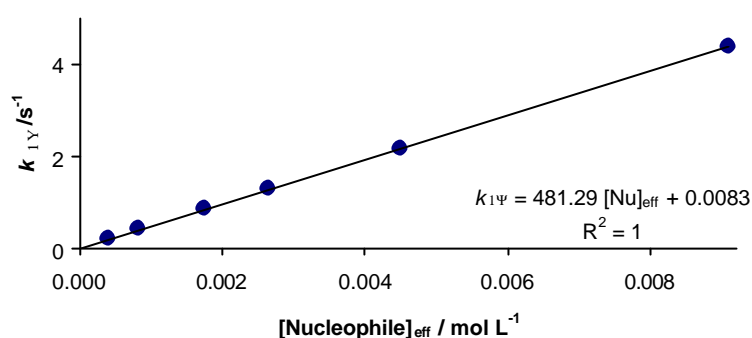
Reaction of 4-aminopyridine (**1e**) with (thq)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in water (at 20 °C, cosolvent: 0.9 vol-% CH<sub>3</sub>CN, stopped-flow, detection at 610 nm)

[(thq) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> / mol L <sup>-1</sup>	[OH <sup>-</sup> ] / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> /[EI] <sub>0</sub>	k <sub>obs</sub> / s <sup>-1</sup>	k <sub>1Ψ,OH</sub> / s <sup>-1</sup>	k <sub>1Ψ</sub> / s <sup>-1</sup>
3.01 × 10 <sup>-5</sup>	9.45 × 10 <sup>-3</sup>	9.10 × 10 <sup>-3</sup>	3.46 × 10 <sup>-4</sup>	302	4.39	8.18 × 10 <sup>-3</sup>	4.38
3.01 × 10 <sup>-5</sup>	4.73 × 10 <sup>-3</sup>	4.49 × 10 <sup>-3</sup>	2.43 × 10 <sup>-4</sup>	149	2.19	5.74 × 10 <sup>-3</sup>	2.18
3.01 × 10 <sup>-5</sup>	2.84 × 10 <sup>-3</sup>	2.65 × 10 <sup>-3</sup>	1.87 × 10 <sup>-4</sup>	88	1.29	4.41 × 10 <sup>-3</sup>	1.29
3.01 × 10 <sup>-5</sup>	1.89 × 10 <sup>-3</sup>	1.74 × 10 <sup>-3</sup>	1.51 × 10 <sup>-4</sup>	58	8.50 × 10 <sup>-1</sup>	3.57 × 10 <sup>-3</sup>	8.46 × 10 <sup>-1</sup>
3.01 × 10 <sup>-5</sup>	9.45 × 10 <sup>-4</sup>	8.40 × 10 <sup>-4</sup>	1.05 × 10 <sup>-4</sup>	28	4.10 × 10 <sup>-1</sup>	2.48 × 10 <sup>-3</sup>	4.08 × 10 <sup>-1</sup>
3.01 × 10 <sup>-5</sup>	4.73 × 10 <sup>-4</sup>	4.00 × 10 <sup>-4</sup>	7.26 × 10 <sup>-5</sup>	13	1.97 × 10 <sup>-1</sup>	1.71 × 10 <sup>-3</sup>	1.95 × 10 <sup>-1</sup>

$$k_{2,N} = 4.81 \times 10^2 \text{ M}^{-1} \text{ s}^{-1}$$

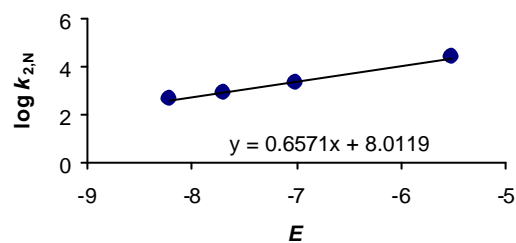
$$k_{2,OH} = 23.6 \text{ M}^{-1} \text{ s}^{-1} \text{ [S3]}$$

$$pK_B = 4.88$$



Reactivity parameters of 4-aminopyridine (**1e**) in water:  $N = 12.19$ ;  $s = 0.657$

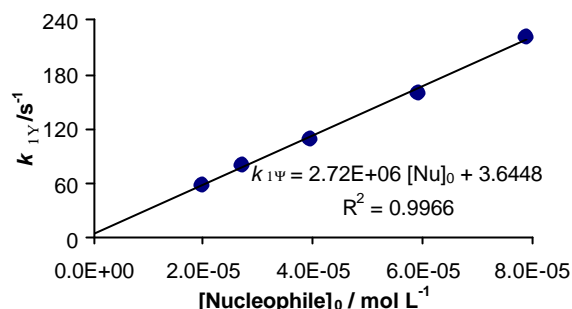
Reference electrophile	$E$ parameter	$k_{2,N}(20^\circ\text{C}) / \text{M}^{-1} \text{s}^{-1}$
(mor) <sub>2</sub> CH <sup>+</sup>	-5.53	$2.58 \times 10^4$
(dma) <sub>2</sub> CH <sup>+</sup>	-7.02	$2.23 \times 10^3$
(pyr) <sub>2</sub> CH <sup>+</sup>	-7.69	$8.04 \times 10^2$
(thq) <sub>2</sub> CH <sup>+</sup>	-8.22	$4.81 \times 10^2$



## 4-(Dimethylamino)pyridine (DMAP, **1f**) in CH<sub>2</sub>Cl<sub>2</sub>

Reaction of DMAP (**1f**) with (mpa)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>2</sub>Cl<sub>2</sub> (at 20 °C, stopped-flow, detection at 622 nm)

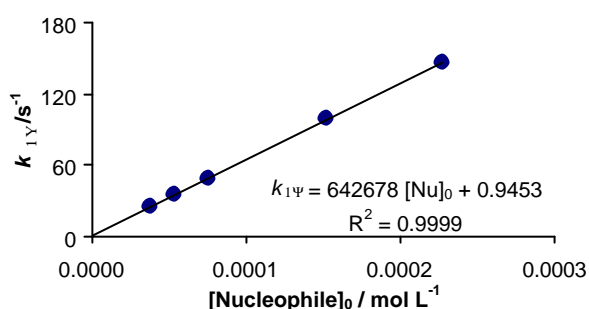
[(mpa) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> /[EI] <sub>0</sub>	k <sub>1Ψ</sub> / s <sup>-1</sup>
	7.89 × 10 <sup>-5</sup>		
2.76 × 10 <sup>-6</sup>	5	29	2.22 × 10 <sup>2</sup>
2.76 × 10 <sup>-6</sup>	5.92 × 10 <sup>-5</sup>	21	1.59 × 10 <sup>2</sup>
2.76 × 10 <sup>-6</sup>	3.95 × 10 <sup>-5</sup>	14	1.09 × 10 <sup>2</sup>
2.76 × 10 <sup>-6</sup>	2.70 × 10 <sup>-5</sup>	10	80.0
2.76 × 10 <sup>-6</sup>	1.97 × 10 <sup>-5</sup>	7	57.7



$$k_{2,N} = 2.72 \times 10^6 \text{ M}^{-1}\text{s}^{-1}$$

Reaction of DMAP (**1f**) with (dma)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>2</sub>Cl<sub>2</sub> (at 20 °C, stopped-flow, detection at 613 nm)

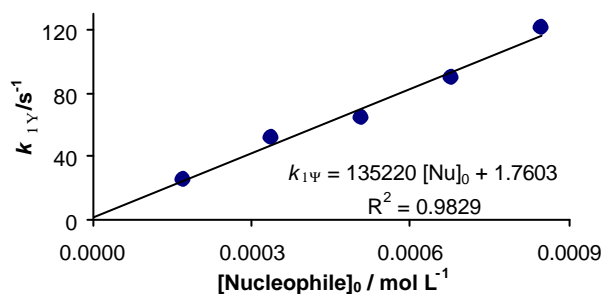
[(dma) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> /[EI] <sub>0</sub>	k <sub>1Ψ</sub> / s <sup>-1</sup>
7.20 × 10 <sup>-6</sup>	2.28 × 10 <sup>-4</sup>	32	1.47 × 10 <sup>2</sup>
7.20 × 10 <sup>-6</sup>	1.52 × 10 <sup>-4</sup>	21	99.5
7.20 × 10 <sup>-6</sup>	7.60 × 10 <sup>-5</sup>	11	49.7
7.20 × 10 <sup>-6</sup>	5.32 × 10 <sup>-5</sup>	7	34.8
7.20 × 10 <sup>-6</sup>	3.80 × 10 <sup>-5</sup>	5	25.4



$$k_{2,N} = 6.43 \times 10^5 \text{ M}^{-1}\text{s}^{-1}$$

Reaction of DMAP (**1f**) with (thq)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>2</sub>Cl<sub>2</sub> (at 20 °C, stopped-flow, detection at 628 nm)

[(thq) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> /[EI] <sub>0</sub>	k <sub>1Ψ</sub> / s <sup>-1</sup>
7.59 × 10 <sup>-6</sup>	8.48 × 10 <sup>-4</sup>	112	1.21 × 10 <sup>2</sup>
7.59 × 10 <sup>-6</sup>	6.78 × 10 <sup>-4</sup>	89	89.8
7.59 × 10 <sup>-6</sup>	5.09 × 10 <sup>-4</sup>	67	64.6
7.59 × 10 <sup>-6</sup>	3.39 × 10 <sup>-4</sup>	45	52.2
7.59 × 10 <sup>-6</sup>	1.70 × 10 <sup>-4</sup>	22	25.2



$$k_{2,N} = 1.35 \times 10^5 \text{ M}^{-1}\text{s}^{-1}$$



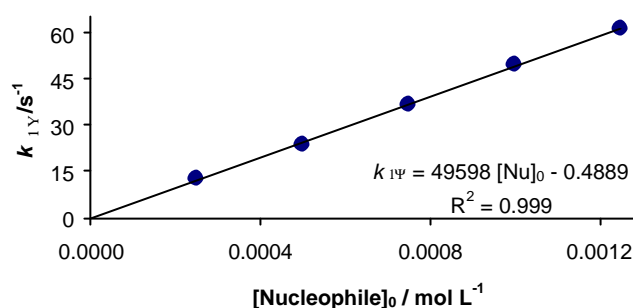
Equilibrium constants for the reaction of DMAP (**1f**) with (thq)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>2</sub>Cl<sub>2</sub> (at 20 °C, J&M, detection at 628 nm)

[(thq) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[(thq) <sub>2</sub> CH <sup>+</sup> ] / mol L <sup>-1</sup>	K/ L mol <sup>-1</sup>
1.69 × 10 <sup>-5</sup>	7.04 × 10 <sup>-5</sup>	1.17 × 10 <sup>-6</sup>	2.45 × 10 <sup>5</sup>
2.01 × 10 <sup>-5</sup>	8.38 × 10 <sup>-5</sup>	1.68 × 10 <sup>-6</sup>	1.68 × 10 <sup>5</sup>
2.65 × 10 <sup>-5</sup>	7.35 × 10 <sup>-5</sup>	1.83 × 10 <sup>-6</sup>	2.76 × 10 <sup>5</sup>
2.65 × 10 <sup>-5</sup>	3.68 × 10 <sup>-5</sup>	4.56 × 10 <sup>-6</sup>	3.23 × 10 <sup>5</sup>

$$K(20\text{ °C}) = (2.81 \pm 0.32) \times 10^5 \text{ L mol}^{-1}$$

Reaction of DMAP (**1f**) with (ind)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>2</sub>Cl<sub>2</sub> (at 20 °C, stopped-flow, detection at 625 nm)

[(ind) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> /[EI] <sub>0</sub>	k <sub>1Ψ</sub> / s <sup>-1</sup>
7.48 × 10 <sup>-6</sup>	1.25 × 10 <sup>-3</sup>	167	61.3
7.48 × 10 <sup>-6</sup>	9.97 × 10 <sup>-4</sup>	133	49.6
7.48 × 10 <sup>-6</sup>	7.48 × 10 <sup>-4</sup>	100	36.5
7.48 × 10 <sup>-6</sup>	4.99 × 10 <sup>-4</sup>	67	23.4
7.48 × 10 <sup>-6</sup>	2.49 × 10 <sup>-4</sup>	33	12.4



$$k_{2,N} = 4.96 \times 10^4 \text{ M}^{-1}\text{s}^{-1}$$

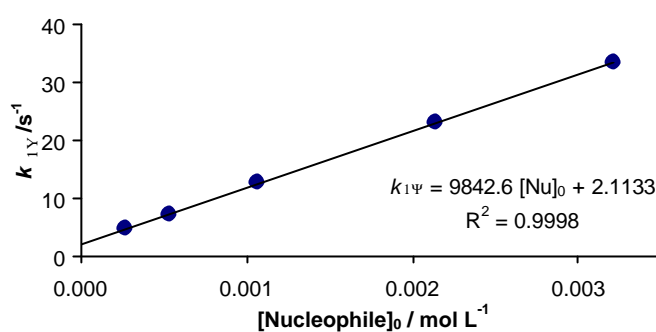
Equilibrium constants for the reaction of DMAP (**1f**) with (ind)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>2</sub>Cl<sub>2</sub> (at 20 °C, J&M, detection at 625 nm)

[(ind) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[(ind) <sub>2</sub> CH <sup>+</sup> ] / mol L <sup>-1</sup>	K / L mol <sup>-1</sup>
2.08 × 10 <sup>-5</sup>	1.55 × 10 <sup>-4</sup>	7.27 × 10 <sup>-7</sup>	2.06 × 10 <sup>5</sup>
2.08 × 10 <sup>-5</sup>	7.76 × 10 <sup>-5</sup>	1.93 × 10 <sup>-6</sup>	1.67 × 10 <sup>5</sup>
2.36 × 10 <sup>-5</sup>	8.78 × 10 <sup>-5</sup>	1.97 × 10 <sup>-6</sup>	1.66 × 10 <sup>5</sup>
2.36 × 10 <sup>-5</sup>	4.40 × 10 <sup>-5</sup>	5.01 × 10 <sup>-6</sup>	1.46 × 10 <sup>5</sup>

$$K(20\text{ °C}) = (1.71 \pm 0.22) \times 10^5 \text{ L mol}^{-1}$$

Reaction of DMAP (1f) with (jul)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>2</sub>Cl<sub>2</sub> (at 20 °C, stopped-flow, detection at 642 nm)

[(jul) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> /[E] <sub>0</sub>	k <sub>1ψ</sub> / s <sup>-1</sup>
7.65 × 10 <sup>-6</sup>	3.21 × 10 <sup>-3</sup>	420	33.7
7.65 × 10 <sup>-6</sup>	2.14 × 10 <sup>-3</sup>	280	23.1
7.65 × 10 <sup>-6</sup>	1.07 × 10 <sup>-3</sup>	140	12.9
7.65 × 10 <sup>-6</sup>	5.35 × 10 <sup>-4</sup>	70	7.24
7.65 × 10 <sup>-6</sup>	2.68 × 10 <sup>-4</sup>	35	4.72



$$k_{2,N} = 9.84 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

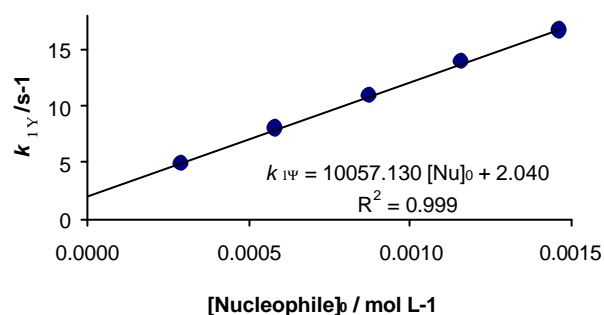
Equilibrium constants for the reaction of DMAP (1f) with (jul)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>2</sub>Cl<sub>2</sub> (at 20 °C, J&M, detection at 642 nm)

[(jul) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[(jul) <sub>2</sub> CH <sup>+</sup> ] / mol L <sup>-1</sup>	K / L mol <sup>-1</sup>
2.16 × 10 <sup>-5</sup>	4.78 × 10 <sup>-4</sup>	5.82 × 10 <sup>-6</sup>	5.87 × 10 <sup>3</sup>
2.16 × 10 <sup>-5</sup>	3.20 × 10 <sup>-4</sup>	7.75 × 10 <sup>-6</sup>	5.85 × 10 <sup>3</sup>
2.16 × 10 <sup>-5</sup>	1.60 × 10 <sup>-4</sup>	1.15 × 10 <sup>-5</sup>	5.83 × 10 <sup>3</sup>

$$K(20 \text{ °C}) = (5.85 \pm 0.02) \times 10^3 \text{ L mol}^{-1}$$

Reaction of DMAP (1f) with (jul)<sub>2</sub>CH<sup>+</sup> PF<sub>6</sub><sup>-</sup> in CH<sub>2</sub>Cl<sub>2</sub> (at 20 °C, stopped-flow, detection at 642 nm)

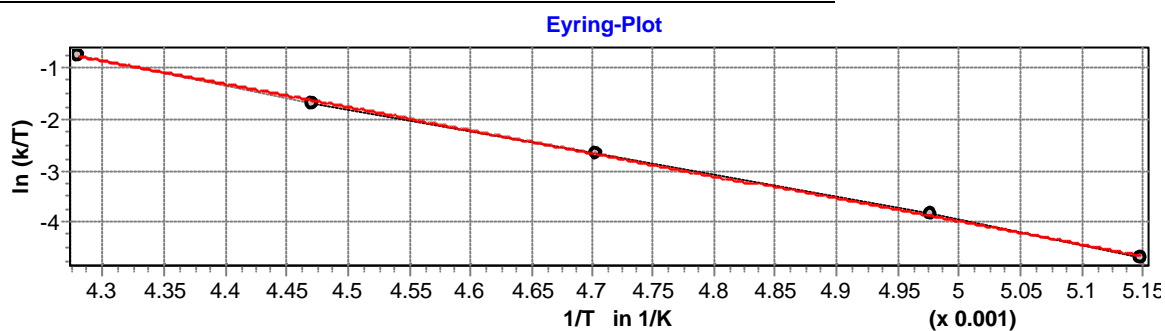
[(jul) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> /[E] <sub>0</sub>	k <sub>1ψ</sub> / s <sup>-1</sup>
8.40 × 10 <sup>-6</sup>	1.46 × 10 <sup>-3</sup>	174	16.6
8.40 × 10 <sup>-6</sup>	1.16 × 10 <sup>-3</sup>	138	13.8
8.40 × 10 <sup>-6</sup>	8.73 × 10 <sup>-4</sup>	104	10.9
8.40 × 10 <sup>-6</sup>	5.82 × 10 <sup>-4</sup>	69	7.95
8.40 × 10 <sup>-6</sup>	2.91 × 10 <sup>-4</sup>	38	4.86



$$k_{2,N} = 1.01 \times 10^4 \text{ M}^{-1}\text{s}^{-1}$$

Reaction of DMAP (**1f**) with  $(\text{lii})_2\text{CH}^+ \text{BF}_4^-$  in  $\text{CH}_2\text{Cl}_2$  (in  $\text{CH}_2\text{Cl}_2$ , Schöllly, detection at 640 nm)

$[(\text{lii})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0/[\text{EI}]_0$	Conv. / %	$T$ / °C	$k_{2,N}$ / L mol <sup>-1</sup> s <sup>-1</sup>
$3.00 \times 10^{-5}$	$3.27 \times 10^{-3}$	109	84	-78.9	1.77
$3.61 \times 10^{-5}$	$3.04 \times 10^{-3}$	84	87	-72.2	4.27
$2.69 \times 10^{-5}$	$2.35 \times 10^{-3}$	87	76	-60.5	$1.50 \times 10^1$
$3.13 \times 10^{-5}$	$1.71 \times 10^{-3}$	55	73	-49.5	$4.13 \times 10^1$
$2.40 \times 10^{-5}$	$1.05 \times 10^{-3}$	44	90	-39.5	$1.09 \times 10^2$



Eyring parameters:

$$\Delta H^\ddagger = 37.1 \pm 0.52 \text{ kJ mol}^{-1}$$

$$\Delta S^\ddagger = -45.3 \pm 2.5 \text{ J mol}^{-1} \text{ K}^{-1}$$

$$r^2 = 0.9994$$

$$k_{2,N} (20 \text{ °C}) = (6.45 \pm 0.53) \times 10^3 \text{ L mol}^{-1} \text{ s}^{-1}$$

Arrhenius parameters:

$$E_a = 38.9 \pm 0.52 \text{ kJ mol}^{-1}$$

$$\ln A = 24.7 \pm 0.30$$

$$r^2 = 0.9995$$

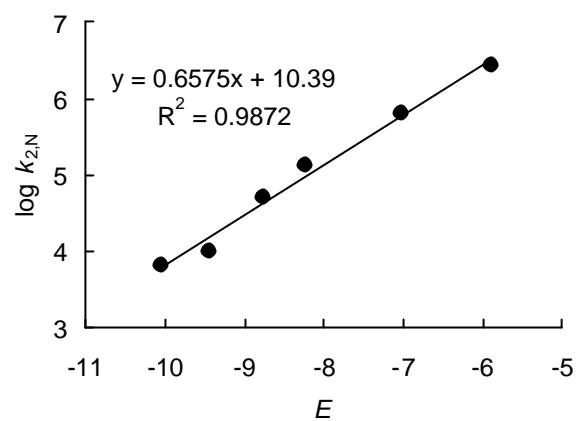
Equilibrium constants for the reaction of DMAP (**1f**) with  $(\text{lii})_2\text{CH}^+ \text{BF}_4^-$  in  $\text{CH}_2\text{Cl}_2$  (at 20 °C, J&M, detection at 639 nm)

$[(\text{lii})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[(\text{lii})_2\text{CH}^+]$ / mol L <sup>-1</sup>	$K$ / L mol <sup>-1</sup>
$1.69 \times 10^{-5}$	$7.23 \times 10^{-4}$	$3.33 \times 10^{-6}$	$5.76 \times 10^3$
$1.69 \times 10^{-5}$	$4.83 \times 10^{-4}$	$4.60 \times 10^{-6}$	$5.70 \times 10^3$
$1.69 \times 10^{-5}$	$2.42 \times 10^{-4}$	$7.34 \times 10^{-6}$	$5.63 \times 10^3$
$1.85 \times 10^{-5}$	$1.32 \times 10^{-4}$	$1.08 \times 10^{-5}$	$5.72 \times 10^3$

$$K (20 \text{ °C}) = (5.70 \pm 0.05) \times 10^3 \text{ L mol}^{-1}$$

Reactivity parameters of DMAP (**1f**) in CH<sub>2</sub>Cl<sub>2</sub>:  $N = 15.80$ ;  $s = 0.658$

Reference electrophile	$E$ parameter	$k_{2N}(20^\circ\text{C}) / \text{M}^{-1} \text{s}^{-1}$
(mpa) <sub>2</sub> CH <sup>+</sup>	-5.89	$2.72 \times 10^6$
(dma) <sub>2</sub> CH <sup>+</sup>	-7.02	$6.43 \times 10^5$
(thq) <sub>2</sub> CH <sup>+</sup>	-8.22	$1.35 \times 10^5$
(ind) <sub>2</sub> CH <sup>+</sup>	-8.76	$4.96 \times 10^4$
(jul) <sub>2</sub> CH <sup>+</sup>	-9.45	$9.84 \times 10^3$
(lil) <sub>2</sub> CH <sup>+</sup>	-10.04	$6.45 \times 10^3$

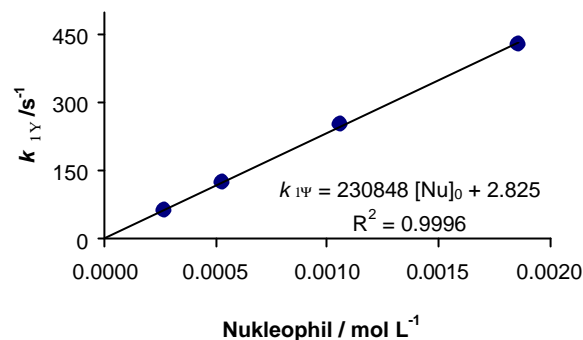


## 4-(Dimethylamino)pyridine (DMAP, **1f**) in Acetonitrile

Reaction of DMAP (**1f**) with  $(\text{dma})_2\text{CH}^+ \text{BF}_4^-$  in  $\text{CH}_3\text{CN}$  (at 20 °C, stopped-flow, detection at 610 nm)

$[(\text{dma})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$1.14 \times 10^{-5}$	$1.86 \times 10^{-3}$	163	$4.30 \times 10^2$
$1.14 \times 10^{-5}$	$1.06 \times 10^{-3}$	93	$2.52 \times 10^2$
$1.14 \times 10^{-5}$	$5.30 \times 10^{-4}$	46	$1.25 \times 10^2$
$1.14 \times 10^{-5}$	$2.65 \times 10^{-4}$	23	61.9

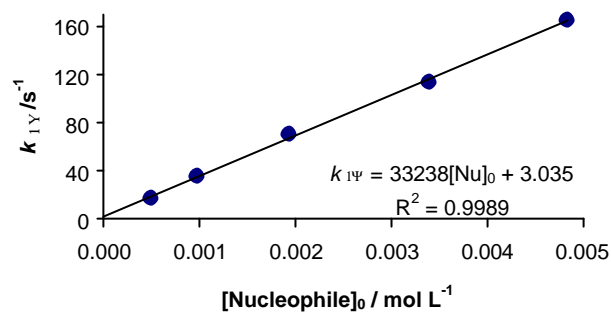
$$k_{2,N} = 2.31 \times 10^5 \text{ M}^{-1}\text{s}^{-1}$$



Reaction of DMAP (**1f**) with  $(\text{thq})_2\text{CH}^+ \text{BF}_4^-$  in  $\text{CH}_3\text{CN}$  (at 20 °C, stopped-flow, detection at 622 nm)

$[(\text{thq})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$9.64 \times 10^{-6}$	$4.84 \times 10^{-3}$	502	$1.64 \times 10^2$
$9.64 \times 10^{-6}$	$3.39 \times 10^{-3}$	352	$1.14 \times 10^2$
$9.64 \times 10^{-6}$	$1.93 \times 10^{-3}$	200	70.3
$9.64 \times 10^{-6}$	$9.67 \times 10^{-4}$	100	35.3
$9.64 \times 10^{-6}$	$4.84 \times 10^{-4}$	50	17.5

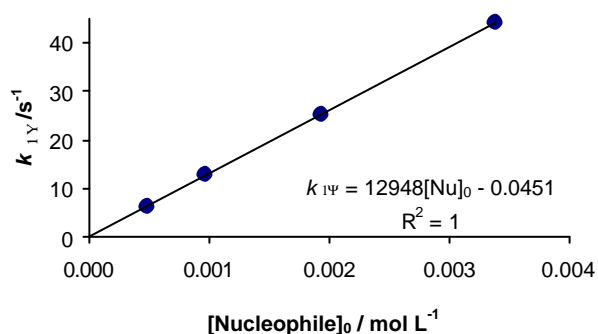
$$k_{2,N} = 3.32 \times 10^4 \text{ M}^{-1}\text{s}^{-1}$$



Reaction of DMAP (**1f**) with  $(\text{ind})_2\text{CH}^+ \text{BF}_4^-$  in  $\text{CH}_3\text{CN}$  (at 20 °C, stopped-flow, detection at 630 nm)

$[(\text{ind})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$7.74 \times 10^{-6}$	$3.39 \times 10^{-3}$	438	43.8
$7.74 \times 10^{-6}$	$1.93 \times 10^{-3}$	249	25.0
$7.74 \times 10^{-6}$	$9.67 \times 10^{-4}$	125	12.6
$7.74 \times 10^{-6}$	$4.84 \times 10^{-4}$	63	6.09

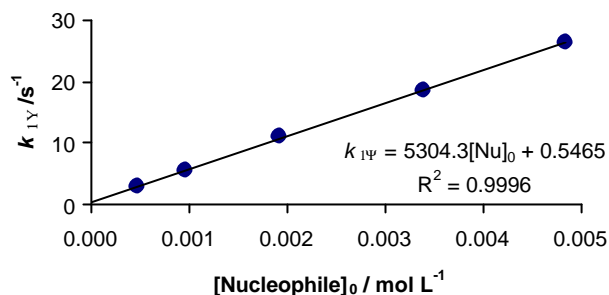
$$k_{2,N} = 1.29 \times 10^4 \text{ M}^{-1}\text{s}^{-1}$$



Reaction of DMAP (1f) with (jul)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>3</sub>CN (at 20 °C, stopped-flow, detection at 630 nm)

[(jul) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> /[EI] <sub>0</sub>	k <sub>1Ψ</sub> / s <sup>-1</sup>
2.86 × 10 <sup>-6</sup>	4.84 × 10 <sup>-3</sup>	1692	26.2
2.86 × 10 <sup>-6</sup>	3.39 × 10 <sup>-3</sup>	1185	18.4
2.86 × 10 <sup>-6</sup>	1.93 × 10 <sup>-3</sup>	675	11.1
2.86 × 10 <sup>-6</sup>	9.67 × 10 <sup>-4</sup>	338	5.67
2.86 × 10 <sup>-6</sup>	4.84 × 10 <sup>-4</sup>	169	2.95

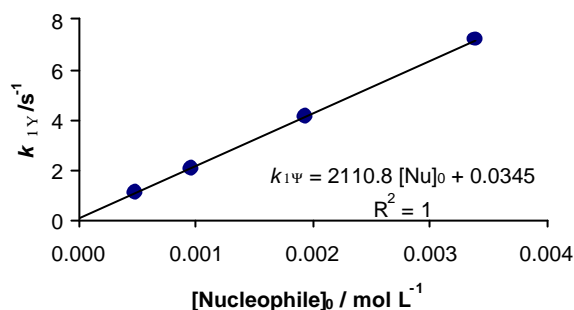
$$k_{2,N} = 5.30 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$



Reaction of DMAP (1f) with (lil)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in CH<sub>3</sub>CN (at 20 °C, stopped-flow, detection at 630 nm)

[(lil) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> /[EI] <sub>0</sub>	k <sub>1Ψ</sub> / s <sup>-1</sup>
9.18 × 10 <sup>-6</sup>	3.39 × 10 <sup>-3</sup>	369	7.20
9.18 × 10 <sup>-6</sup>	1.93 × 10 <sup>-3</sup>	210	4.09
9.18 × 10 <sup>-6</sup>	9.67 × 10 <sup>-4</sup>	105	2.07
9.18 × 10 <sup>-6</sup>	4.84 × 10 <sup>-4</sup>	53	1.07

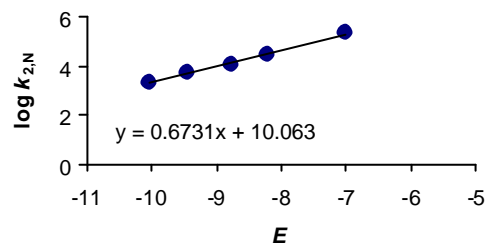
$$k_{2,N} = 2.11 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$



Reactivity parameters of DMAP (1f) in CH<sub>3</sub>CN:  $N = 14.95$ ;  $s = 0.673$

Reference electrophile  $E$  parameter  $k_{2,N}(20^\circ\text{C}) / \text{M}^{-1} \text{s}^{-1}$

(dma) <sub>2</sub> CH <sup>+</sup>	-7.02	2.31 × 10 <sup>5</sup>
(thq) <sub>2</sub> CH <sup>+</sup>	-8.22	3.32 × 10 <sup>4</sup>
(ind) <sub>2</sub> CH <sup>+</sup>	-8.76	1.29 × 10 <sup>4</sup>
(jul) <sub>2</sub> CH <sup>+</sup>	-9.45	5.30 × 10 <sup>3</sup>
(lil) <sub>2</sub> CH <sup>+</sup>	-10.04	2.11 × 10 <sup>3</sup>



## 4-(Dimethylamino)pyridine (DMAP, **1f**) in Water

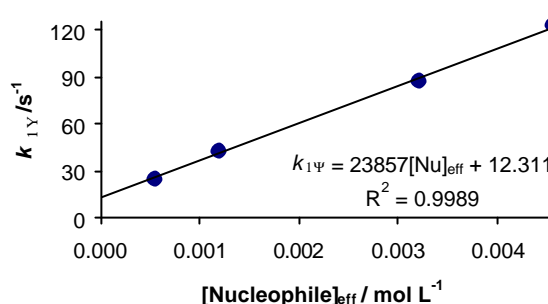
Reaction of DMAP (**1f**) with  $(\text{mor})_2\text{CH}^+ \text{BF}_4^-$  in water (at 20 °C, cosolvent: 9 vol-%  $\text{CH}_3\text{CN}$ , stopped-flow, detection at 607 nm)

$[(\text{mor})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}$ / mol L <sup>-1</sup>	$[\text{OH}^-]$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}/[\text{EI}]_0$	$k_{\text{obs}}$ / s <sup>-1</sup>	$k_{1\Psi,\text{OH}}$ / s <sup>-1</sup>	$k_{1\Psi}$ / s <sup>-1</sup>
$8.35 \times 10^{-5}$	$5.06 \times 10^{-3}$	$4.58 \times 10^{-3}$	$4.79 \times 10^{-4}$	55	$1.23 \times 10^2$	$5.08 \times 10^{-1}$	$1.22 \times 10^2$
$8.35 \times 10^{-5}$	$3.62 \times 10^{-3}$	$3.22 \times 10^{-3}$	$4.02 \times 10^{-4}$	39	87.8	$4.26 \times 10^{-1}$	87.4
$8.35 \times 10^{-5}$	$1.45 \times 10^{-3}$	$1.20 \times 10^{-3}$	$2.46 \times 10^{-4}$	14	42.8	$2.60 \times 10^{-1}$	42.5
$8.35 \times 10^{-5}$	$7.24 \times 10^{-4}$	$5.52 \times 10^{-4}$	$1.67 \times 10^{-4}$	7	25.1	$1.77 \times 10^{-1}$	24.9

$$k_{2,\text{N}} = 2.39 \times 10^4 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{2,\text{OH}} = 1.06 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_{\text{B}} = 4.30$$



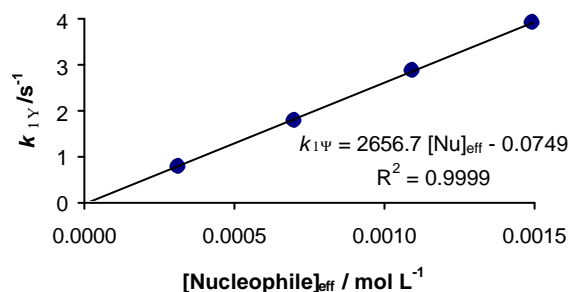
Reaction of DMAP (**1f**) with  $(\text{dma})_2\text{CH}^+ \text{BF}_4^-$  in water (at 20 °C, cosolvent: 0.4 vol-%  $\text{CH}_3\text{CN}$ , stopped-flow, detection at 604 nm)

$[(\text{dma})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}$ / mol L <sup>-1</sup>	$[\text{OH}^-]$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}/[\text{EI}]_0$	$k_{\text{obs}}$ / s <sup>-1</sup>	$k_{1\Psi,\text{OH}}$ / s <sup>-1</sup>	$k_{1\Psi}$ / s <sup>-1</sup>
$8.32 \times 10^{-5}$	$1.78 \times 10^{-3}$	$1.51 \times 10^{-3}$	$2.75 \times 10^{-4}$	18	3.95	$3.60 \times 10^{-2}$	3.91
$8.32 \times 10^{-5}$	$1.34 \times 10^{-3}$	$1.10 \times 10^{-3}$	$2.35 \times 10^{-4}$	13	2.91	$3.08 \times 10^{-2}$	2.88
$8.32 \times 10^{-5}$	$8.91 \times 10^{-4}$	$7.03 \times 10^{-4}$	$1.88 \times 10^{-4}$	8	1.81	$2.46 \times 10^{-2}$	1.79
$8.32 \times 10^{-5}$	$4.45 \times 10^{-4}$	$3.19 \times 10^{-4}$	$1.26 \times 10^{-4}$	4	$7.87 \times 10^{-1}$	$1.66 \times 10^{-2}$	$7.70 \times 10^{-1}$

$$k_{2,\text{N}} = 2.66 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{2,\text{OH}} = 1.31 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_{\text{B}} = 4.30$$



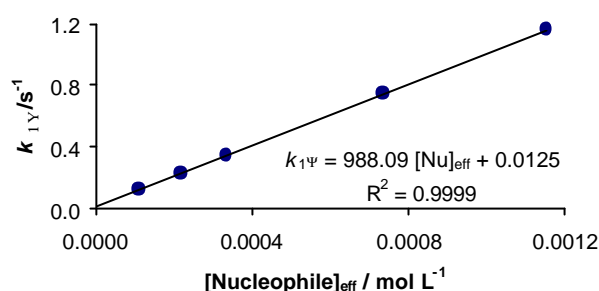
Reaction of DMAP (1f) with (pyr)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in water (at 20 °C, cosolvent: 0.2 vol-% CH<sub>3</sub>CN, stopped-flow, detection at 611 nm)

[(pyr) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> / mol L <sup>-1</sup>	[OH <sup>-</sup> ] / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> /[EI] <sub>0</sub>	k <sub>obs</sub> / s <sup>-1</sup>	k <sub>1Ψ,OH</sub> / s <sup>-1</sup>	k <sub>1Ψ</sub> / s <sup>-1</sup>
1.26 × 10 <sup>-5</sup>	1.40 × 10 <sup>-3</sup>	1.16 × 10 <sup>-3</sup>	2.41 × 10 <sup>-4</sup>	92	1.17	1.17 × 10 <sup>-2</sup>	1.16
1.26 × 10 <sup>-5</sup>	9.35 × 10 <sup>-4</sup>	7.42 × 10 <sup>-4</sup>	1.93 × 10 <sup>-4</sup>	59	7.53 × 10 <sup>-1</sup>	9.35 × 10 <sup>-3</sup>	7.44 × 10 <sup>-1</sup>
1.26 × 10 <sup>-5</sup>	4.67 × 10 <sup>-4</sup>	3.37 × 10 <sup>-4</sup>	1.30 × 10 <sup>-4</sup>	27	3.53 × 10 <sup>-1</sup>	6.30 × 10 <sup>-3</sup>	3.47 × 10 <sup>-1</sup>
1.26 × 10 <sup>-5</sup>	3.27 × 10 <sup>-4</sup>	2.22 × 10 <sup>-4</sup>	1.05 × 10 <sup>-4</sup>	18	2.38 × 10 <sup>-1</sup>	5.11 × 10 <sup>-3</sup>	2.33 × 10 <sup>-1</sup>
1.26 × 10 <sup>-5</sup>	1.87 × 10 <sup>-4</sup>	1.12 × 10 <sup>-4</sup>	7.49 × 10 <sup>-5</sup>	9	1.31 × 10 <sup>-1</sup>	3.63 × 10 <sup>-3</sup>	1.27 × 10 <sup>-1</sup>

$$k_{2,N} = 9.88 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{2,OH} = 48.5 \text{ M}^{-1}\text{s}^{-1}$$

$$pK_B = 4.30$$



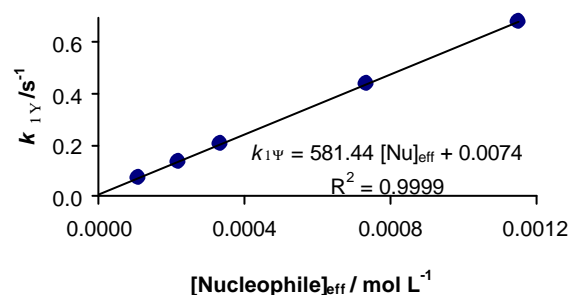
Reaction of DMAP (1f) with (thq)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in water (at 20 °C, cosolvent: 0.2 vol-% CH<sub>3</sub>CN, stopped-flow, detection at 611 nm)

[(thq) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> / mol L <sup>-1</sup>	[OH <sup>-</sup> ] / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> /[EI] <sub>0</sub>	k <sub>obs</sub> / s <sup>-1</sup>	k <sub>1Ψ,OH</sub> / s <sup>-1</sup>	k <sub>1Ψ</sub> / s <sup>-1</sup>
1.58 × 10 <sup>-5</sup>	1.40 × 10 <sup>-3</sup>	1.16 × 10 <sup>-3</sup>	2.41 × 10 <sup>-4</sup>	73	6.85 × 10 <sup>-1</sup>	5.69 × 10 <sup>-3</sup>	6.79 × 10 <sup>-1</sup>
1.58 × 10 <sup>-5</sup>	9.35 × 10 <sup>-4</sup>	7.42 × 10 <sup>-4</sup>	1.93 × 10 <sup>-4</sup>	47	4.45 × 10 <sup>-1</sup>	4.55 × 10 <sup>-3</sup>	4.40 × 10 <sup>-1</sup>
1.58 × 10 <sup>-5</sup>	4.67 × 10 <sup>-4</sup>	3.37 × 10 <sup>-4</sup>	1.30 × 10 <sup>-4</sup>	21	2.10 × 10 <sup>-1</sup>	3.07 × 10 <sup>-3</sup>	2.07 × 10 <sup>-1</sup>
1.58 × 10 <sup>-5</sup>	3.27 × 10 <sup>-4</sup>	2.22 × 10 <sup>-4</sup>	1.05 × 10 <sup>-4</sup>	14	1.40 × 10 <sup>-1</sup>	2.49 × 10 <sup>-3</sup>	1.38 × 10 <sup>-1</sup>
1.58 × 10 <sup>-5</sup>	1.87 × 10 <sup>-4</sup>	1.12 × 10 <sup>-4</sup>	7.49 × 10 <sup>-5</sup>	7	7.40 × 10 <sup>-2</sup>	1.77 × 10 <sup>-3</sup>	7.22 × 10 <sup>-2</sup>

$$k_{2,N} = 5.81 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{2,OH} = 23.6 \text{ M}^{-1}\text{s}^{-1} \text{ [S3]}$$

$$pK_B = 4.30$$





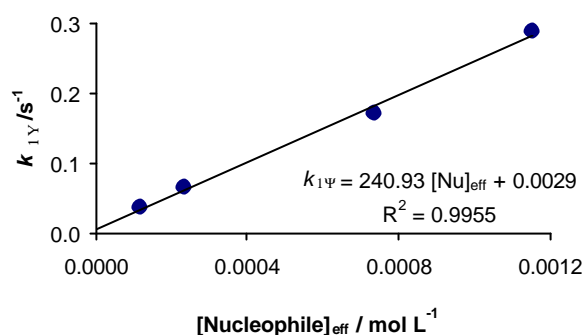
Reaction of DMAP (**1f**) with (ind)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in water (at 20 °C, cosolvent: 0.4 vol-% CH<sub>3</sub>CN, stopped-flow and J&M, detection at 614 nm)

[(ind) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub>	[Nu] <sub>0</sub>	[Nu] <sub>eff</sub>	[OH <sup>-</sup> ]	[Nu] <sub>eff</sub> /[EI] <sub>0</sub>	k <sub>obs</sub>	k <sub>1Ψ,OH</sub>	k <sub>1Ψ</sub>
/ mol L <sup>-1</sup>	/ mol L <sup>-1</sup>	/ mol L <sup>-1</sup>	/ mol L <sup>-1</sup>		/ s <sup>-1</sup>	/ s <sup>-1</sup>	/ s <sup>-1</sup>
1.48 × 10 <sup>-5</sup>	1.40 × 10 <sup>-3</sup>	1.16 × 10 <sup>-3</sup>	2.41 × 10 <sup>-4</sup>	78	2.91 × 10 <sup>-1</sup>	2.60 × 10 <sup>-3</sup>	2.88 × 10 <sup>-1</sup>
1.48 × 10 <sup>-5</sup>	9.35 × 10 <sup>-4</sup>	7.42 × 10 <sup>-4</sup>	1.93 × 10 <sup>-4</sup>	50	1.72 × 10 <sup>-1</sup>	2.08 × 10 <sup>-3</sup>	1.70 × 10 <sup>-1</sup>
1.47 × 10 <sup>-5</sup>	3.43 × 10 <sup>-4</sup>	2.35 × 10 <sup>-4</sup>	1.08 × 10 <sup>-4</sup>	16	6.63 × 10 <sup>-2</sup>	1.17 × 10 <sup>-3</sup>	6.51 × 10 <sup>-2</sup>
1.48 × 10 <sup>-5</sup>	1.97 × 10 <sup>-4</sup>	1.20 × 10 <sup>-4</sup>	7.74 × 10 <sup>-5</sup>	8	3.60 × 10 <sup>-2</sup>	8.36 × 10 <sup>-4</sup>	3.52 × 10 <sup>-2</sup>

$$k_{2,N} = 2.41 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{2,OH} = 10.8 \text{ M}^{-1}\text{s}^{-1}$$

$$pK_B = 4.30$$



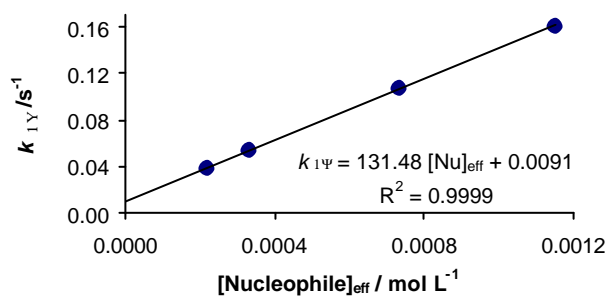
Reaction of DMAP (**1f**) with (jul)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in water (at 20 °C, cosolvent: 0.2 vol-% CH<sub>3</sub>CN, stopped-flow, detection at 630 nm)

[(jul) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub>	[Nu] <sub>0</sub>	[Nu] <sub>eff</sub>	[OH <sup>-</sup> ]	[Nu] <sub>eff</sub> /[EI] <sub>0</sub>	k <sub>obs</sub>	k <sub>1Ψ,OH</sub>	k <sub>1Ψ</sub>
/ mol L <sup>-1</sup>	/ mol L <sup>-1</sup>	/ mol L <sup>-1</sup>	/ mol L <sup>-1</sup>		/ s <sup>-1</sup>	/ s <sup>-1</sup>	/ s <sup>-1</sup>
1.26 × 10 <sup>-5</sup>	1.40 × 10 <sup>-3</sup>	1.16 × 10 <sup>-3</sup>	2.41 × 10 <sup>-4</sup>	92	1.62 × 10 <sup>-1</sup>	8.29 × 10 <sup>-4</sup>	1.61 × 10 <sup>-1</sup>
1.26 × 10 <sup>-5</sup>	9.35 × 10 <sup>-4</sup>	7.42 × 10 <sup>-4</sup>	1.93 × 10 <sup>-4</sup>	59	1.08 × 10 <sup>-1</sup>	6.63 × 10 <sup>-4</sup>	1.07 × 10 <sup>-1</sup>
1.26 × 10 <sup>-5</sup>	4.67 × 10 <sup>-4</sup>	3.37 × 10 <sup>-4</sup>	1.30 × 10 <sup>-4</sup>	27	5.34 × 10 <sup>-2</sup>	4.47 × 10 <sup>-4</sup>	5.30 × 10 <sup>-2</sup>
1.26 × 10 <sup>-5</sup>	3.27 × 10 <sup>-4</sup>	2.22 × 10 <sup>-4</sup>	1.05 × 10 <sup>-4</sup>	18	3.87 × 10 <sup>-2</sup>	3.63 × 10 <sup>-4</sup>	3.83 × 10 <sup>-2</sup>

$$k_{2,N} = 1.31 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{2,OH} = 3.44 \text{ M}^{-1}\text{s}^{-1}$$

$$pK_B = 4.30$$



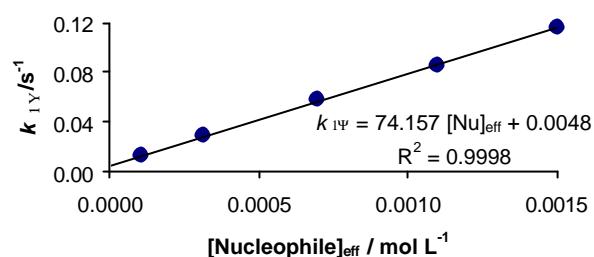
Reaction of DMAP (**1f**) with  $(\text{lii})_2\text{CH}^+ \text{BF}_4^-$  in water (at 20 °C, cosolvent: 0.4 vol-%  $\text{CH}_3\text{CN}$ , stopped-flow, detection at 630 nm)

$[(\text{lii})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}$ / mol L <sup>-1</sup>	$[\text{OH}^-]$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}/[\text{E}]_0$	$k_{\text{obs}}$ / s <sup>-1</sup>	$k_{1\Psi, \text{OH}^-}$ / s <sup>-1</sup>	$k_{1\Psi}$ / s <sup>-1</sup>
$1.81 \times 10^{-5}$	$1.78 \times 10^{-3}$	$1.51 \times 10^{-3}$	$2.75 \times 10^{-4}$	83	$1.17 \times 10^{-1}$	$5.93 \times 10^{-4}$	$1.16 \times 10^{-1}$
$1.81 \times 10^{-5}$	$1.34 \times 10^{-3}$	$1.10 \times 10^{-3}$	$2.35 \times 10^{-4}$	61	$8.66 \times 10^{-2}$	$5.08 \times 10^{-4}$	$8.61 \times 10^{-2}$
$1.81 \times 10^{-5}$	$8.91 \times 10^{-4}$	$7.03 \times 10^{-4}$	$1.88 \times 10^{-4}$	39	$5.82 \times 10^{-2}$	$4.06 \times 10^{-4}$	$5.78 \times 10^{-2}$
$1.81 \times 10^{-5}$	$4.45 \times 10^{-4}$	$3.19 \times 10^{-4}$	$1.26 \times 10^{-4}$	18	$2.89 \times 10^{-2}$	$2.73 \times 10^{-4}$	$2.86 \times 10^{-2}$
$1.81 \times 10^{-5}$	$1.78 \times 10^{-4}$	$1.05 \times 10^{-4}$	$7.27 \times 10^{-5}$	6	$1.22 \times 10^{-2}$	$1.57 \times 10^{-4}$	$1.20 \times 10^{-2}$

$$k_{2,\text{N}} = 74.2 \text{ M}^{-1}\text{s}^{-1}$$

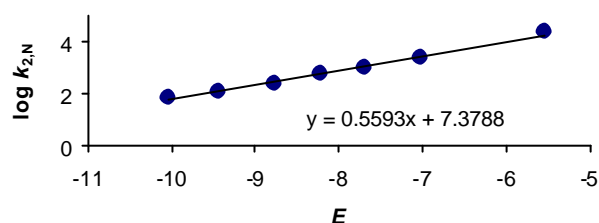
$$k_{2,\text{OH}} = 2.16 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_{\text{B}} = 4.30$$



Reactivity parameters of DMAP (**1f**) in water:  $N = 13.19$ ;  $s = 0.559$

Reference electrophile	$E$ parameter	$k_{2,\text{N}}(20^\circ\text{C}) / \text{M}^{-1} \text{s}^{-1}$
$(\text{mor})_2\text{CH}^+$	-5.53	$2.39 \times 10^4$
$(\text{dma})_2\text{CH}^+$	-7.02	$2.66 \times 10^3$
$(\text{pyr})_2\text{CH}^+$	-7.69	$9.88 \times 10^2$
$(\text{thq})_2\text{CH}^+$	-8.22	$5.81 \times 10^2$
$(\text{ind})_2\text{CH}^+$	-8.76	$2.41 \times 10^2$
$(\text{jul})_2\text{CH}^+$	-9.45	$1.31 \times 10^2$
$(\text{lii})_2\text{CH}^+$	-10.04	$7.42 \times 10^1$



## 4-(Dimethylamino)pyridine (DMAP, **1f**) in Methanol

In the reaction of the  $(\text{dma})_2\text{CH}^+ \text{BF}_4^-$  with **1f** in methanol, the competing reaction of the carbocation with the solvent methanol has to be considered. The observed pseudo-first-order rate constants  $k_{\text{obs}}$  reflect the sum of the reactions of the electrophile with **1f** ( $k_{1\Psi,\text{N}}$ ), methanol ( $k_{1\Psi,\text{MeOH}}$ ) and the reverse reaction ( $k_{-1}$ ) [Eq. (S9)].

$$k_{\text{obs}} = k_{1\Psi,\text{N}} + k_{1\Psi,\text{MeOH}} + k_{-1} \quad (\text{S9})$$

$$k_{1\Psi} = k_{\text{obs}} = k_{2,\text{N}}[\mathbf{1f}] + k_{1\Psi,\text{MeOH}} + k_{-1}$$

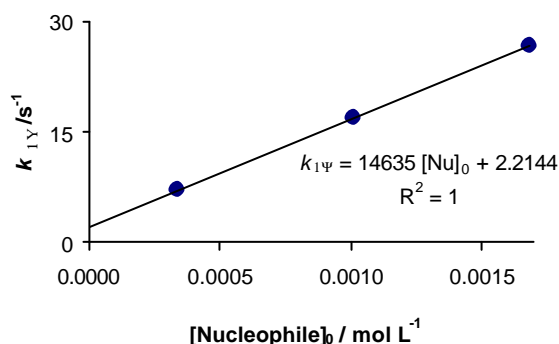
The slopes of the plots  $k_{1\Psi}$  versus **[1f]** corresponds to the second-order rate constant  $k_{2,\text{N}}$ . The intercept, which correspond to the reverse reactions and the reactions of  $(\text{dma})_2\text{CH}^+ \text{BF}_4^-$  with methanol is negligible. For technical reasons the experiment was performed in a solvent mixture containing 91 vol-%  $\text{CH}_3\text{OH}$  and 9 vol-%  $\text{CH}_3\text{CN}$  (corresponding to a 10/1 mixing ratio in the stopped-flow experiment).

Reaction of DMAP (**1f**) with  $(\text{dma})_2\text{CH}^+ \text{BF}_4^-$  in methanol (at 20 °C, cosolvent: 9 vol-%  $\text{CH}_3\text{CN}$ , stopped-flow, detection at 610 nm)

$[(\text{dma})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$2.02 \times 10^{-5}$	$1.68 \times 10^{-3}$	83	26.8
$2.02 \times 10^{-5}$	$1.01 \times 10^{-3}$	50	11.7
$2.02 \times 10^{-5}$	$3.36 \times 10^{-4}$	17	7.13

$$k_{2,\text{N}} = 1.46 \times 10^4 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{1,\text{MeOH}}(\text{CH}_3\text{OH}/\text{CH}_3\text{CN } 91/9 \text{ (v/v)}) = 2.31 \text{ s}^{-1} \text{ [S2]}$$

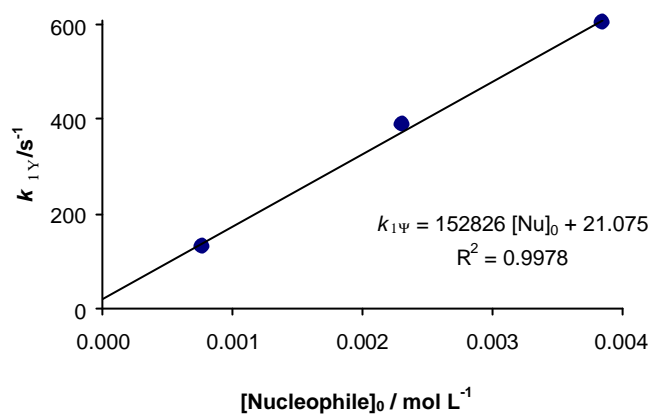


## 4-(Dimethylamino)pyridine (DMAP, **1f**) in DMSO

Reaction of DMAP (**1f**) with  $(\text{dma})_2\text{CH}^+ \text{BF}_4^-$  in DMSO (at 20 °C, stopped-flow, detection at 616 nm)

$[(\text{dma})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$9.27 \times 10^{-6}$	$3.85 \times 10^{-3}$	415	$6.03 \times 10^2$
$9.27 \times 10^{-6}$	$2.31 \times 10^{-3}$	249	$3.87 \times 10^2$
$9.27 \times 10^{-6}$	$7.68 \times 10^{-4}$	83	$1.32 \times 10^2$

$$k_{2,N} = 1.53 \times 10^5 \text{ M}^{-1}\text{s}^{-1}$$

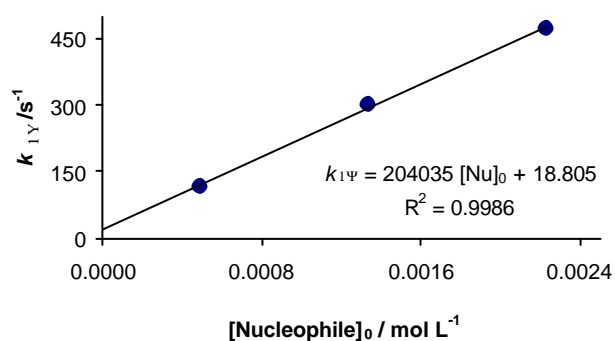


## 4-(Dimethylamino)pyridine (DMAP, **1f**) in DMF

Reaction of DMAP (**1f**) with  $(\text{dma})_2\text{CH}^+ \text{BF}_4^-$  in DMF (at 20 °C, stopped-flow, detection at 616 nm)

$[(\text{dma})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$9.27 \times 10^{-6}$	$2.23 \times 10^{-3}$	241	$4.70 \times 10^2$
$9.27 \times 10^{-6}$	$1.34 \times 10^{-3}$	145	$3.00 \times 10^2$
$9.27 \times 10^{-6}$	$4.91 \times 10^{-4}$	53	$1.15 \times 10^2$

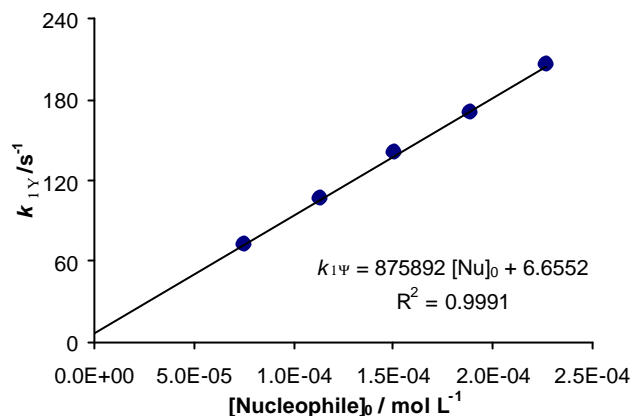
$$k_{2,N} = 2.04 \times 10^5 \text{ M}^{-1}\text{s}^{-1}$$



## 4-(Pyrrolidino)pyridine (**1g**)

Reaction of 4-pyrrolidinopyridine (**1g**) with  $(\text{dma})_2\text{CH}^+ \text{BF}_4^-$  in  $\text{CH}_2\text{Cl}_2$  (at 20 °C, stopped-flow, detection at 612 nm)

$[(\text{dma})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0/[\text{EI}]_0$	$k_{1\Psi}$ / s <sup>-1</sup>
$8.61 \times 10^{-6}$	$2.27 \times 10^{-4}$	26	$2.06 \times 10^2$
$8.61 \times 10^{-6}$	$1.89 \times 10^{-4}$	22	$1.70 \times 10^2$
$8.61 \times 10^{-6}$	$1.51 \times 10^{-4}$	18	$1.41 \times 10^2$
$8.61 \times 10^{-6}$	$1.13 \times 10^{-4}$	13	$1.06 \times 10^2$
$8.61 \times 10^{-6}$	$7.56 \times 10^{-5}$	9	72.1



$$k_{2,N} = 8.76 \times 10^5 \text{ M}^{-1}\text{s}^{-1}$$

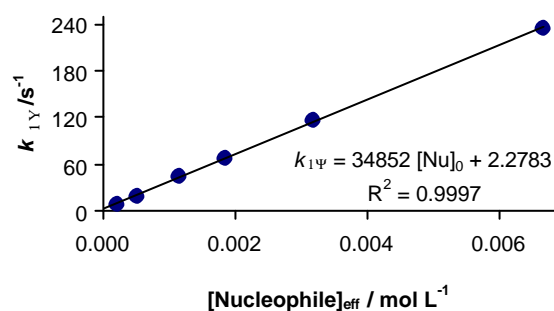
Reaction of 4pyrrolidinopyridine (**1g**) with  $(\text{mor})_2\text{CH}^+ \text{BF}_4^-$  in water (at 20 °C, cosolvent: 9 vol-%  $\text{CH}_3\text{CN}$ , stopped-flow, detection at 610 nm)

$[(\text{mor})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}$ / mol L <sup>-1</sup>	$[\text{OH}^-]$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}/[\text{EI}]_0$	$k_{\text{obs}}$ / s <sup>-1</sup>	$k_{1\Psi,\text{OH}}$ / s <sup>-1</sup>	$k_{1\Psi}$ / s <sup>-1</sup>
$3.65 \times 10^{-5}$	$7.42 \times 10^{-3}$	$6.69 \times 10^{-3}$	$7.29 \times 10^{-4}$	183	$2.35 \times 10^2$	$7.73 \times 10^{-1}$	$2.34 \times 10^2$
$3.65 \times 10^{-5}$	$3.71 \times 10^{-3}$	$3.21 \times 10^{-3}$	$5.05 \times 10^{-4}$	88	$1.17 \times 10^2$	$5.35 \times 10^{-1}$	$1.16 \times 10^2$
$3.65 \times 10^{-5}$	$2.23 \times 10^{-3}$	$1.85 \times 10^{-3}$	$3.83 \times 10^{-4}$	51	67.4	$4.06 \times 10^{-1}$	67.0
$3.65 \times 10^{-5}$	$1.48 \times 10^{-3}$	$1.17 \times 10^{-3}$	$3.05 \times 10^{-4}$	32	43.9	$3.24 \times 10^{-1}$	43.6
$3.65 \times 10^{-5}$	$7.42 \times 10^{-4}$	$5.36 \times 10^{-4}$	$2.06 \times 10^{-4}$	15	20.5	$2.19 \times 10^{-1}$	20.3
$3.65 \times 10^{-5}$	$3.71 \times 10^{-4}$	$2.35 \times 10^{-4}$	$1.36 \times 10^{-4}$	6	9.33	$1.45 \times 10^{-1}$	9.19

$$k_{2,N} = 3.49 \times 10^4 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{2,\text{OH}} = 1.06 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$\text{p}K_{\text{B}} = 4.1$$



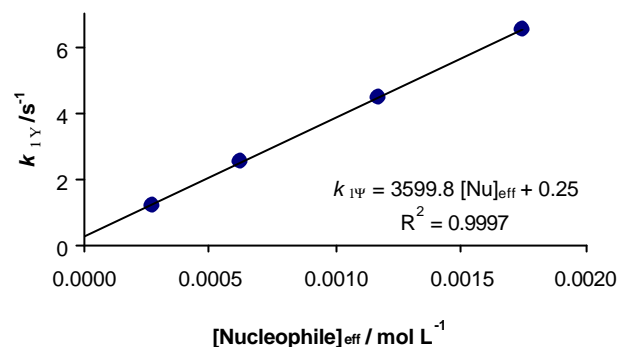
Reaction of 4-pyrrolidinopyridine (**1g**) with (dma)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in water (at 20 °C, cosolvent: 0.4 vol-% CH<sub>3</sub>CN, stopped-flow, detection at 610 nm)

[(dma) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> / mol L <sup>-1</sup>	[OH <sup>-</sup> ] / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> /[EI] <sub>0</sub>	k <sub>obs</sub> / s <sup>-1</sup>	k <sub>1Ψ,OH</sub> / s <sup>-1</sup>	k <sub>1Ψ</sub> / s <sup>-1</sup>
1.22 × 10 <sup>-5</sup>	2.12 × 10 <sup>-3</sup>	1.75 × 10 <sup>-3</sup>	3.73 × 10 <sup>-4</sup>	143	6.58	4.88 × 10 <sup>-2</sup>	6.53
1.22 × 10 <sup>-5</sup>	1.48 × 10 <sup>-3</sup>	1.17 × 10 <sup>-3</sup>	3.05 × 10 <sup>-4</sup>	96	4.51	4.00 × 10 <sup>-2</sup>	4.47
1.22 × 10 <sup>-5</sup>	8.48 × 10 <sup>-4</sup>	6.25 × 10 <sup>-4</sup>	2.23 × 10 <sup>-4</sup>	51	2.59	2.92 × 10 <sup>-2</sup>	2.56
1.22 × 10 <sup>-5</sup>	4.24 × 10 <sup>-4</sup>	2.76 × 10 <sup>-4</sup>	1.48 × 10 <sup>-4</sup>	23	1.22	1.94 × 10 <sup>-2</sup>	1.20

$$k_{2,N} = 3.60 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{2,OH} = 1.31 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

$$pK_B = 4.1$$



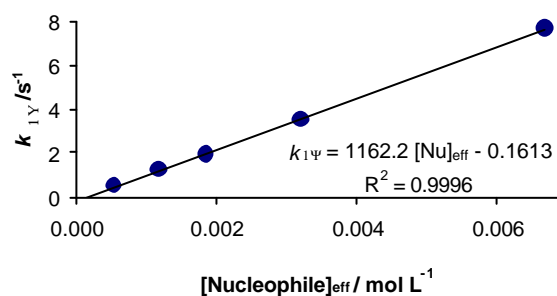
Reaction of 4-pyrrolidinopyridine (**1g**) with (pyr)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in water (at 20 °C, cosolvent: 0.9 vol-% CH<sub>3</sub>CN, stopped-flow, detection at 610 nm)

[(pyr) <sub>2</sub> CH <sup>+</sup> ] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>0</sub> / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> / mol L <sup>-1</sup>	[OH <sup>-</sup> ] / mol L <sup>-1</sup>	[Nu] <sub>eff</sub> /[EI] <sub>0</sub>	k <sub>obs</sub> / s <sup>-1</sup>	k <sub>1Ψ,OH</sub> / s <sup>-1</sup>	k <sub>1Ψ</sub> / s <sup>-1</sup>
2.78 × 10 <sup>-5</sup>	7.42 × 10 <sup>-3</sup>	6.69 × 10 <sup>-3</sup>	7.29 × 10 <sup>-4</sup>	241	7.69	3.54 × 10 <sup>-2</sup>	7.65
2.78 × 10 <sup>-5</sup>	3.71 × 10 <sup>-3</sup>	3.21 × 10 <sup>-3</sup>	5.05 × 10 <sup>-4</sup>	115	3.52	2.45 × 10 <sup>-2</sup>	3.50
2.78 × 10 <sup>-5</sup>	2.23 × 10 <sup>-3</sup>	1.85 × 10 <sup>-3</sup>	3.83 × 10 <sup>-4</sup>	66	1.95	1.86 × 10 <sup>-2</sup>	1.93
2.78 × 10 <sup>-5</sup>	1.48 × 10 <sup>-3</sup>	1.17 × 10 <sup>-3</sup>	3.05 × 10 <sup>-4</sup>	42	1.23	1.48 × 10 <sup>-2</sup>	1.22
2.78 × 10 <sup>-5</sup>	7.42 × 10 <sup>-4</sup>	5.36 × 10 <sup>-4</sup>	2.06 × 10 <sup>-4</sup>	19	5.42 × 10 <sup>-1</sup>	1.00 × 10 <sup>-2</sup>	5.32 × 10 <sup>-1</sup>

$$k_{2,N} = 1.16 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$$

$$k_{2,OH} = 48.5 \text{ M}^{-1}\text{s}^{-1}$$

$$pK_B = 4.1$$



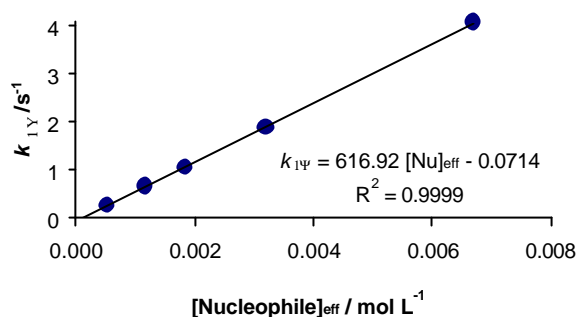
Reaction of 4-pyrrolidinopyridine (**1g**) with (thq)<sub>2</sub>CH<sup>+</sup> BF<sub>4</sub><sup>-</sup> in water (at 20 °C, cosolvent: 0.9 vol-% CH<sub>3</sub>CN, stopped-flow, detection at 610 nm)

$[(\text{thq})_2\text{CH}^+]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_0$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}$ / mol L <sup>-1</sup>	$[\text{OH}^-]$ / mol L <sup>-1</sup>	$[\text{Nu}]_{\text{eff}}/[\text{E}]_0$	$k_{\text{obs}}$ / s <sup>-1</sup>	$k_{1\Psi,\text{OH}}$ / s <sup>-1</sup>	$k_{1\Psi}$ / s <sup>-1</sup>
$3.01 \times 10^{-5}$	$7.42 \times 10^{-3}$	$6.69 \times 10^{-3}$	$7.29 \times 10^{-4}$	222	4.08	$1.72 \times 10^{-2}$	4.06
$3.01 \times 10^{-5}$	$3.71 \times 10^{-3}$	$3.21 \times 10^{-3}$	$5.05 \times 10^{-4}$	106	1.91	$1.19 \times 10^{-2}$	1.90
$3.01 \times 10^{-5}$	$2.23 \times 10^{-3}$	$1.85 \times 10^{-3}$	$3.83 \times 10^{-4}$	61	1.06	$9.04 \times 10^{-3}$	1.05
$3.01 \times 10^{-5}$	$1.48 \times 10^{-3}$	$1.17 \times 10^{-3}$	$3.05 \times 10^{-4}$	39	$6.67 \times 10^{-1}$	$7.21 \times 10^{-3}$	$6.60 \times 10^{-1}$
$3.01 \times 10^{-5}$	$7.42 \times 10^{-4}$	$5.36 \times 10^{-4}$	$2.06 \times 10^{-4}$	18	$2.76 \times 10^{-1}$	$4.87 \times 10^{-3}$	$2.71 \times 10^{-1}$

$$k_{2,\text{N}} = 6.17 \times 10^2 \text{ M}^{-1}\text{s}^{-1}$$

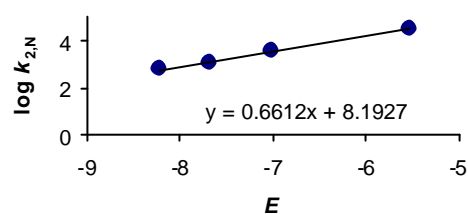
$$k_{2,\text{OH}} = 23.6 \text{ M}^{-1} \text{ s}^{-1} [\text{S}^1]$$

$$\text{p}K_{\text{B}} = 4.1$$



Reactivity parameters of 4-pyrrolidinopyridine (**1g**) in water:  $N = 12.39$ ;  $s = 0.661$

Reference electrophile	$E$ parameter	$k_2(20^\circ\text{C}) / \text{M}^{-1} \text{ s}^{-1}$
(mor) <sub>2</sub> CH <sup>+</sup>	-5.53	$3.49 \times 10^4$
(dma) <sub>2</sub> CH <sup>+</sup>	-7.02	$3.60 \times 10^3$
(pyr) <sub>2</sub> CH <sup>+</sup>	-7.69	$1.16 \times 10^3$
(thq) <sub>2</sub> CH <sup>+</sup>	-8.22	$6.17 \times 10^2$



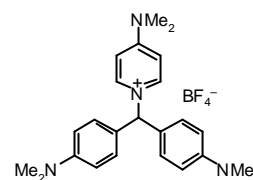
## Products from Reactions of DMAP (1f) with Benzhydrylium Salts

$^1\text{H}$  (300 MHz) and  $^{13}\text{C}$  NMR spectra (75.5 MHz) were recorded with a Bruker ARX 300. Chemical shifts were reported on the  $\delta$  scale relative to tetramethylsilane ( $^1\text{H}$ ) and  $\text{CDCl}_3$  ( $^{13}\text{C}$ ) as internal standards.

**General Procedure.** In a carefully dried, nitrogen-flushed Schlenk-flask a solution of 4-(dimethylamino)pyridine (DMAP, **1f**) in absolute  $\text{CH}_2\text{Cl}_2$  (5 mL) was added dropwise to a solution of the benzhydrylium tetrafluoroborate in absolute  $\text{CH}_2\text{Cl}_2$  (50 mL). After stirring at ambient temperature for 1 h, the solvent was evaporated in the vacuum to yield the crude product, which was washed with absolute  $\text{Et}_2\text{O}$  (20 mL) and dried several hours in the vacuum ( $10^{-2}$  mbar).

### 1-(Bis(4-dimethylaminophenyl)methyl)-4-(dimethylamino)pyridinium

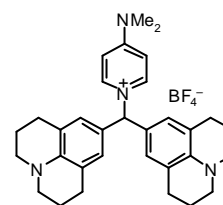
**tetrafluoroborate (2a)** was obtained from **1f** (61.8 mg, 0.506 mmol) and bis(4-dimethylaminophenyl)methylium tetrafluoroborat ( $\text{dma})_2\text{CH}^+ \text{BF}_4^-$  (172 mg, 0.506 mmol) as a bright blue solid (169 mg, 72 %).  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 2.88 (s, 12 H), 3.13 (s, 6 H), 6.52 (s, 1 H), 6.59 (d,  $J$  = 8.8 Hz, 4 H), 6.78 (d,  $J$  = 7.9 Hz, 2 H), 6.91 (d,  $J$  = 8.7 Hz, 4 H), 7.83 (d,  $J$  = 7.9 Hz, 2 H);  $^{13}\text{C}$  NMR (75.5 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 39.9 (q), 40.0 (q), 73.8 (d), 107.5 (d), 112.2 (d), 123.4 (s), 129.1 (d), 140.7 (d), 150.4 (s), 156.2 (s).



**2a**

### 1-(Bis(julolidin-9-yl)methyl)-4-(dimethylamino)pyridinium

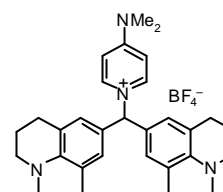
**tetrafluoroborate (2b)** was obtained from **1f** (20 mg, 0.16 mmol) and bis(julolidin-9-yl)methylium tetrafluoroborat ( $\text{jul})_2\text{CH}^+ \text{BF}_4^-$  (73 mg, 0.16 mmol) as a bright blue solid (63 mg, 68 %).  $^1\text{H}$  NMR (300 MHz,  $\text{CD}_3\text{CN}$ ):  $\delta$  = 1.83–1.89 (m, 8 H), 2.60–2.64 (m, 8 H), 3.13–3.14 (m, 8 H), 3.14 (s, superimposed, 6 H), 6.29 (s, 1 H), 6.50 (s, 4 H), 6.77 (d,  $J$  = 7.8 Hz, 2 H), 7.85 (d,  $J$  = 7.9 Hz, 2 H);  $^{13}\text{C}$  NMR (75.5 MHz,  $\text{CD}_3\text{CN}$ ):  $\delta$  = 22.5 (t), 28.4 (t), 40.6 (q), 50.4 (t), 75.0 (d), 108.4 (d), 122.6 (d), 127.8 (d), 141.8 (d), 144.1 (s), 148.6 (d), 157.6 (s).



**2b**

### 1-(Bis(lilolidin-8-yl)methyl)-4-(dimethylamino)pyridinium

**tetrafluoroborate (2c)** was obtained from **1f** (20 mg, 0.16 mmol) and bis(lilolidin-8-yl)methylium tetrafluoroborat ( $\text{lil})_2\text{CH}^+ \text{BF}_4^-$  (68 mg, 0.13 mmol) as a bright blue solid (51 mg, 58 %).  $^1\text{H}$  NMR (300 MHz,  $\text{CD}_3\text{CN}$ ):  $\delta$  = 1.98–2.02 (m, 4 H), 2.55–2.59 (m, 4 H), 2.79–2.85 (m, 4 H), 2.94–2.98 (m, 4 H), 3.15 (s, 6 H), 3.23–3.29 (m, 4 H), 6.45 (s, 1 H), 6.57 (s, 2 H), 6.67 (s, 2 H), 6.78 (d,  $J$  = 7.9 Hz, 2 H), 7.86 (d,  $J$  = 7.9 Hz, 2 H);  $^{13}\text{C}$  NMR (75.5 MHz,  $\text{CD}_3\text{CN}$ ):  $\delta$  = 23.6 (t), 24.6 (t), 29.1 (t), 40.6 (q), 47.4 (t), 55.6 (t), 75.6 (d), 108.4 (d), 120.2 (s), 123.1 (d), 127.6 (d), 130.4 (s), 141.8 (d), 151.6 (s), 157.6 (s).



**2c**



## Computational Details

**General.** All calculations were performed with Gaussian 03, Rev. B.01.<sup>[18]</sup> Structures were optimized at the MP2 (frozen core) level using the 6-31G(d) basis set. The MP2 frequencies (unscaled) were used to calculate the zero-point energies (*ZPE*). For the reaction products of the pyridines **1c**, **1d** and **1g** with the benzhydrylium ion two conformational isomers exist. The corresponding energies are shown in the supplemental material. Values for  $\Delta H_{298}$  and  $\Delta G_{298}$  in Table 4 were obtained by conformational averaging for these adducts **3c**, **3d** and **3g**.

Single point energies have subsequently been performed at the MP2(fc)/6-311++G(d,p) level of theory. Combination of the single point energies with thermochemical corrections calculated at the MP2(fc)/6-31G(d) level yields the " $E_0$ ", " $H_{298}$ " and " $G_{298}$ " values which can be found in the supplemental material.

### Reaction Energies for all Stationary Points as Optimized at the MP2(fc)/6-31G(d) Level of Theory

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}}$ / a.u.	$E_0$ / a.u.	$H_{298}$ / a.u.	$G_{298}$ / a.u.
benzhydryl cation	-500.0844505	-499.884452	-499.873233	-499.920254

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}}$ / a.u.	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-500.3260228	-500.126024	-500.114806	-500.161827

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}}$ / a.u.	$E_0$ / a.u.	$H_{298}$ / a.u.	$G_{298}$ / a.u.
<b>1a</b>	-706.5146861	-706.434802	-706.428440	-706.463903

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}}$ / a.u.	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-706.6691680	-706.589284	-706.582922	-706.618385

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}} / \text{a.u.}$	$E_0 / \text{a.u.}$	$H_{298} / \text{a.u.}$	$G_{298} / \text{a.u.}$
<b>1b</b>	-247.4825325	-247.393089	-247.387802	-247.419876

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}} / \text{a.u.}$	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-247.6092285	-247.519785	-247.514498	-247.546573

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}} / \text{a.u.}$	$E_0 / \text{a.u.}$	$H_{298} / \text{a.u.}$	$G_{298} / \text{a.u.}$
<b>1c</b>	-286.6543269	-286.536514	-286.529459	-286.566813

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}} / \text{a.u.}$	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-286.8107137	-286.692901	-286.685846	-286.723200

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}} / \text{a.u.}$	$E_0 / \text{a.u.}$	$H_{298} / \text{a.u.}$	$G_{298} / \text{a.u.}$
<b>1d</b>	-361.6730054	-361.549815	-361.542086	-361.580849

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}} / \text{a.u.}$	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-361.8711469	-361.747956	-361.740227	-361.778991

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}} / \text{a.u.}$	$E_0 / \text{a.u.}$	$H_{298} / \text{a.u.}$	$G_{298} / \text{a.u.}$
<b>1e</b>	-302.6786366	-302.572093	-302.565461	-302.601203

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}} / \text{a.u.}$	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-302.8444452	-302.737901	-302.731269	-302.767011

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}} / \text{a.u.}$	$E_0 / \text{a.u.}$	$H_{298} / \text{a.u.}$	$G_{298} / \text{a.u.}$
<b>1f</b>	-380.9952652	-380.830420	-380.821016	-380.863686

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}} / \text{a.u.}$	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-381.2156504	-381.050805	-381.041401	-381.084071

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}} / \text{a.u.}$	$E_0 / \text{a.u.}$	$H_{298} / \text{a.u.}$	$G_{298} / \text{a.u.}$
<b>1g</b>	-458.1656993	-457.963644	-457.953342	-457.999965

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}} / \text{a.u.}$	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-458.4267811	-458.224726	-458.214424	-458.261047

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}} / \text{a.u.}$	$E_0 / \text{a.u.}$	$H_{298} / \text{a.u.}$	$G_{298} / \text{a.u.}$
<b>3a</b>	-1206.6685233	-1206.384846	-1206.367481	-1206.431088

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}} / \text{a.u.}$	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-1207.0724461	-1206.788769	-1206.771404	-1206.835011

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}} / \text{a.u.}$	$E_0 / \text{a.u.}$	$H_{298} / \text{a.u.}$	$G_{298} / \text{a.u.}$
<b>3b</b>	-747.6426477	-747.349083	-747.332928	-747.393163

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}} / \text{a.u.}$	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-748.0165883	-747.723023	-747.706868	-747.767104

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}} / \text{a.u.}$	$E_0 / \text{a.u.}$	$H_{298} / \text{a.u.}$	$G_{298} / \text{a.u.}$
<b>3c</b> conformation 1	-786.8189340	-786.497199	-786.479170	-786.544094
<b>3c</b> conformation 2	-768.8189070	-786.497194	-786.479147	-786.544225

$\langle H_{298} \rangle = -786.479159 \text{ a.u.}$ ,  $\langle G_{298} \rangle = -786.544164 \text{ a.u.}$

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}} / \text{a.u.}$	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-787.2226685	-786.900934	-786.882905	-786.947829
-787.2226311	-786.900918	-786.882871	-786.947949

$\langle "H_{298}" \rangle = -786.882888 \text{ a.u.}$ ,  $\langle "G_{298}" \rangle = -786.947893 \text{ a.u.}$

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}} / \text{a.u.}$	$E_0 / \text{a.u.}$	$H_{298} / \text{a.u.}$	$G_{298} / \text{a.u.}$
<b>3d</b> conformation 1	-861.8410158	-861.513908	-861.495124	-861.561422
<b>3d</b> conformation 2	-861.8407311	-861.513651	-861.494851	-861.561258

$\langle H_{298} \rangle = -861.495007 \text{ a.u.}$ ,  $\langle G_{298} \rangle = -861.561347 \text{ a.u.}$

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}} / \text{a.u.}$	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-862.2862137	-861.959106	-861.940322	-862.006620
-862.2862417	-861.959162	-861.940362	-862.006769

$\langle "H_{298}" \rangle = -861.940342 \text{ a.u.}$ ,  $\langle "G_{298}" \rangle = -862.006700 \text{ a.u.}$

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}} / \text{a.u.}$	$E_0 / \text{a.u.}$	$H_{298} / \text{a.u.}$	$G_{298} / \text{a.u.}$
<b>3e</b>	-802.8531099	-802.543134	-802.525251	-802.589180

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}} / \text{a.u.}$	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-803.2643414	-802.954365	-802.936482	-803.000411

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}}$ / a.u.	$E_0$ / a.u.	$H_{298}$ / a.u.	$G_{298}$ / a.u.
<b>3f</b>	-881.1763403	-880.807821	-880.787051	-880.858252

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}}$ / a.u.	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-881.6428677	-881.274348	-881.253579	-881.324780

stationary point	MP2(FC)/6-31G(d)// MP2(FC)/6-31G(d)			
	$E_{\text{tot}}$ / a.u.	$E_0$ / a.u.	$H_{298}$ / a.u.	$G_{298}$ / a.u.
<b>3g</b> conformation 1	-958.3503958	-957.943976	-957.922795	-957.995212
<b>3g</b> conformation 2	-958.3503816	-957.943965	-957.922782	-957.995204

$\langle H_{298} \rangle = -957.922789$  a.u.,  $\langle G_{298} \rangle = -957.995208$  a.u.

MP2(FC)/6-311++G(d,p)// MP2(FC)/6-31G(d)			
$E_{\text{tot}}$ / a.u.	" $E_0$ " / a.u.	" $H_{298}$ " / a.u.	" $G_{298}$ " / a.u.
-958.8581181	-958.451698	-958.430517	-958.502934
-958.8577907	-958.451375	-958.430192	-958.502614

$\langle "H_{298}" \rangle = -958.430382$  a.u.,  $\langle "G_{298}" \rangle = -958.502801$  a.u.

## Structures of all Stationary Points

### benzhydryl cation ( $\text{Ph}_2\text{CH}^+$ )

```
1\1\GINC-LXSRV64\SP\RMP2-FC\6-311++G(d,p)\C13H11(1+)\UI22103\08-Oct-20
03\0\#\ MP2(FC)/6-311++G** SCF=TIGHT GEOM=CHECK GUESS=READ MAXDISK=163
75MB\SP Benzhydryl-Kation C2-symm.\1,1\H,0,0.,0.,2.1741320674\C,0,0.,
,0.,1.0804800674\C,0,0.,1.273046,0.4681010674\C,0,0.,-1.273046,0.46810
10674\C,0,0.420774,2.37777,1.2610170674\C,0,-0.420774,-2.37777,1.26101
70674\C,0,0.512659,3.640234,0.6963970674\C,0,-0.512659,-3.640234,0.696
3970674\C,0,0.125093,3.835002,-0.6372269326\C,0,-0.125093,-3.835002,-0
.6372269326\C,0,-0.357628,2.767371,-1.4103619326\C,0,0.357628,-2.76737
1,-1.4103619326\C,0,-0.419854,1.490389,-0.8747329326\C,0,0.419854,-1.4
90389,-0.8747329326\H,0,0.712828,2.21301,2.2962920674\H,0,-0.712828,-2
.21301,2.2962920674\H,0,0.861651,4.480154,1.2899140674\H,0,-0.861651,-
4.480154,1.2899140674\H,0,0.171297,4.831711,-1.0694079326\H,0,-0.17129
7,-4.831711,-1.0694079326\H,0,-0.702917,2.947923,-2.4243339326\H,0,0.7
02917,-2.947923,-2.4243339326\H,0,-0.871082,0.680509,-1.4401309326\H,0
,0.871082,-0.680509,-1.4401309326\Version=x86-Linux-G03RevB.01\State=
1-A\HF=-498.5635824\MP2=-500.3260228\RMSD=3.232e-09\PG=C02 [C2(C1H1),X
(C12H10)]\@\
```

### 1a

```
1\1\GINC-LXSRV107\SP\RMP2-FC\6-311++G(d,p)\C5H4C11N1\UI22102\05-Sep-20
05\0\#\ MP2(FC)/6-311++G** SCF=TIGHT GEOM=CHECK GUESS=READ MAXDISK=163
75MB\SP 4-Chlorpyridin C2V-symm.\0,1\N,0,0.,0.,-2.2959558103\C,0,0.,
1.143613,-1.5886068103\C,0,0.,1.203155,-0.1945828103\C,0,0.,0.,0.50673
41897\C,0,0.,-1.203155,-0.1945828103\C,0,0.,-1.143613,-1.5886068103\H,
0,0.,2.063062,-2.1709268103\H,0,0.,2.155634,0.3262151897\H,0,0.,-2.155
634,0.3262151897\H,0,0.,-2.063062,-2.1709268103\C1,0,0.,0.,2.242293189
7\Version=x86-Linux-G03RevB.01\State=1-A1\HF=-705.6742093\MP2=-706.66
9168\RMSD=6.610e-09\PG=C02V [C2(N1C1C11),SGV(C4H4)]\@\
```

### 1b

```
1\1\GINC-LXSRV88\SP\RMP2-FC\6-311++G(d,p)\C5H5N1\UI22103\30-Sep-2003\0
\#\ MP2(FC)/6-311++G** SCF=TIGHT GEOM=CHECK GUESS=READ\SP Pyridin C2V
-symm.\0,1\N,0,-1.2354982503,0.,-0.7133152472\C,0,-1.1971091093,0.,0.
6314514583\C,0,-0.01533855,0.,1.3740447612\C,0,1.2019678176,0.,0.69395
```

64429\C,0,1.1822883939,0.,-0.7003059549\C,0,-0.0517015509,0.,-1.352452  
6289\H,0,-2.1641977893,0.,1.1314127745\H,0,-0.0529914578,0.,2.45986912  
03\H,0,2.143488262,0.,1.2375435248\H,0,2.1038134189,0.,-1.2758265093\H  
,0,-0.1022666904,0.,-2.4399566516\\Version=x86-Linux-G03RevB.01\State=  
1-A1\HF=-246.7509496\MP2=-247.6092285\RMSD=2.253e-09\PG=C02V [C2(H1C1N  
1),SGV(C4H4)]\\@

### 1c

1\1\GINC-LXSRV7\SP\RMP2-FC\6-311++G(d,p)\C6H7N1\UI22102\06-Sep-2005\0\  
\# MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ MAXDISK=16375MB\  
\SP 4-Methylpyridin CS-symm.\0,1\N,0,0.00584114,-1.93081684,0.\C,0,0.  
00644514,-1.22010084,1.142072\C,0,0.00644514,0.17358816,1.190499\C,0,0  
.00882914,0.90684616,0.\C,0,0.00644514,0.17358816,-1.190499\C,0,0.0064  
4514,-1.22010084,-1.142072\H,0,0.00893914,-1.80265284,2.062042\H,0,0.0  
1176814,0.68145016,2.152762\H,0,0.01176814,0.68145016,-2.152762\H,0,0.  
00893914,-1.80265284,-2.062042\C,0,-0.02767786,2.41116316,0.\H,0,0.468  
30614,2.81626816,-0.886295\H,0,-1.06050586,2.77568316,0.\H,0,0.4683061  
4,2.81626816,0.886295\\Version=x86-Linux-G03RevB.01\State=1-A'\HF=-285  
.7990407\MP2=-286.8107137\RMSD=1.945e-09\PG=CS [SG(C2H1N1),X(C4H6)]\\@

### 1d

1\1\GINC-LXSRV85\SP\RMP2-FC\6-311++G(d,p)\C6H7N1O1\UI22102\05-Sep-2005  
\0\\# MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ MAXDISK=16375  
MB\\SP 4-Methoxy pyridin CS-symm.\0,1\N,0,0.9052040345,-2.145158,0.\C,  
0,1.7605900345,-1.101883,0.\C,0,1.3677480345,0.229948,0.\C,0,0.0000000  
345,0.527541,0.\C,0,-0.9096529655,-0.532688,0.\C,0,-0.3994009655,-1.83  
4083,0.\H,0,2.8187920345,-1.356907,0.\H,0,2.0919680345,1.038845,0.\H,0  
,-1.9829979655,-0.380473,0.\H,0,-1.0879709655,-2.677838,0.\O,0,-0.3254  
049655,1.851253,0.\C,0,-1.7169999655,2.171139,0.\H,0,-2.2123859655,1.7  
81694,0.89529\H,0,-1.7619129655,3.259223,0.\H,0,-2.2123859655,1.781694  
,-0.89529\\Version=x86-Linux-G03RevB.01\State=1-A'\HF=-360.66555\MP2=-  
361.8711469\RMSD=8.811e-09\PG=CS [SG(C6H5N1O1),X(H2)]\\@

### 1e

1\1\GINC-LXSRV77\SP\RMP2-FC\6-311++G(d,p)\C5H6N2\UI22102\05-Sep-2005\0\  
\# MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ MAXDISK=16375MB  
\SP 4-Aminopyridin CS-symm.\0,1\N,0,-0.00533388,-1.90689116,0.\C,0,-

0.00588788,-1.19079116,1.139214\C,0,-0.00588788,0.19952484,1.19522\C,0  
,0.00564112,0.93058884,0.\C,0,-0.00588788,0.19952484,-1.19522\C,0,-0.0  
0588788,-1.19079116,-1.139214\H,0,-0.00738088,-1.76889616,2.062096\H,0  
, -0.00899588,0.70846384,2.15677\H,0,-0.00899588,0.70846384,-2.15677\H,  
0,-0.00738088,-1.76889616,-2.062096\N,0,-0.06216788,2.32261884,0.\H,0,  
0.30636412,2.76121684,0.836707\H,0,0.30636412,2.76121684,-0.836707\\Ve  
rsion=x86-Linux-G03RevB.01\State=1-A'\HF=-301.8049239\MP2=-302.8444452  
\RMSD=8.752e-09\PG=CS [SG(C1N2),X(C4H6)]\@

## 1f

1\1\GINC-LXSRV105\SP\RMP2-FC\6-311++G(d,p)\C7H10N2\UI22103\09-Jan-2004  
\0\# MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ\SP Dimethyla  
minopyridin CS-symm.\0,1\N,0,-0.1075100758,-2.6722419091,0.\C,0,-0.05  
61730758,-1.9517349091,1.13448\C,0,0.0443269242,-0.5654519091,1.193801  
\C,0,0.1028229242,0.1810690909,0.\C,0,0.0443269242,-0.5654519091,-1.19  
3801\C,0,-0.0561730758,-1.9517349091,-1.13448\H,0,-0.1028100758,-2.523  
2989091,2.060712\H,0,0.0677199242,-0.0887309091,2.166669\H,0,0.0677199  
242,-0.0887309091,-2.166669\H,0,-0.1028100758,-2.5232989091,-2.060712\  
N,0,0.2506609242,1.5621500909,0.\C,0,-0.0561730758,2.2731450909,-1.232  
578\H,0,-1.0969160758,2.1327840909,-1.560631\H,0,0.1202559242,3.337594  
0909,-1.071119\H,0,0.6103689242,1.9480160909,-2.034388\C,0,-0.05617307  
58,2.2731450909,1.232578\H,0,0.6103689242,1.9480160909,2.034388\H,0,0.  
1202559242,3.3375940909,1.071119\H,0,-1.0969160758,2.1327840909,1.5606  
31\\Version=x86-Linux-G03RevB.01\State=1-A'\HF=-379.8688486\MP2=-381.2  
156504\RMSD=8.072e-09\PG=CS [SG(C1N2),X(C6H10)]\@

## 1g

1\1\GINC-LXSRV73\SP\RMP2-FC\6-311++G(d,p)\C9H12N2\UI22102\09-Sep-2005\  
0\# MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ MAXDISK=16375M  
B\SP 4-Pyrrolidinopyridin C2-symm.\0,1\N,0,0.,0.,3.3818480375\C,0,0.  
000812,1.13649,2.6604920375\C,0,0.,1.197834,1.2715570375\C,0,0.,0.,0.5  
279890375\C,0,0.,-1.197834,1.2715570375\C,0,-0.000812,-1.13649,2.66049  
20375\H,0,-0.001239,2.061353,3.2365040375\H,0,-0.016755,2.165513,0.780  
4200375\H,0,0.016755,-2.165513,0.7804200375\H,0,0.001239,-2.061353,3.2  
365040375\N,0,0.,0.,-0.8411469625\C,0,-0.126017,-1.209471,-1.646836962  
5\C,0,0.126017,1.209471,-1.6468369625\C,0,-0.38712,-0.659185,-3.048675  
9625\H,0,0.794859,-1.812264,-1.6155699625\C,0,0.38712,0.659185,-3.0486



759625\H,0,-0.794859,1.812264,-1.6155699625\H,0,-0.070316,-1.349161,-3.8357119625\H,0,0.070316,1.349161,-3.8357119625\H,0,-0.949173,-1.836808,-1.2843339625\H,0,0.949173,1.836808,-1.2843339625\H,0,-1.457327,-0.46538,-3.1769459625\H,0,1.457327,0.46538,-3.1769459625\\Version=x86-Linux-G03RevB.01\State=1-A\HF=-456.7988852\MP2=-458.4267811\RMDS=4.642e-09\PG=C02 [C2(N1C1N1),X(C8H12)]\ \@

### 3a

1\1\GINC-LX64I51\SP\RMP2-FC\6-311++G(d,p)\C18H15Cl1N1(1+)\UI22102\12-Sep-2005\0\#\ MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ\SP Benzhydryl-Kation + 4-Chloropyridin Cl-symm.\1,1\H,0,0.7861159864,0.0012018844,-1.9384170476\C,0,0.7627219864,0.0022118844,-0.8410270476\N,0,-0.7248780136,-0.0910031156,-0.5171600476\C,0,1.5000179864,-1.2142971156,-0.3493870476\C,0,1.2724449864,1.3269588844,-0.3328300476\C,0,1.4813489864,-2.3734611156,-1.1408500476\C,0,0.8648819864,1.8497438844,0.9025459524\C,0,2.1235799864,-3.5312921156,-0.7034790476\C,0,1.4061299864,3.0513988844,1.3621719524\C,0,2.7802399864,-3.5405521156,0.5305799524\C,0,2.3563669864,3.7317468844,0.5963679524\C,0,2.8033199864,-2.3869101156,1.3186939524\C,0,2.7629969864,3.2112168844,-0.6346280476\C,0,2.1634919864,-1.2238601156,0.8858579524\C,0,2.2211779864,2.0129448844,-1.1010550476\H,0,0.9808579864,-2.3639401156,-2.1087800476\H,0,0.1242099864,1.3310228844,1.5097909524\H,0,2.1187789864,-4.4217831156,-1.3264070476\H,0,1.0862809864,3.4559218844,2.3189359524\H,0,3.3286349864,-2.3890521156,2.2701389524\H,0,3.4976679864,3.7406658844,-1.2352980476\H,0,2.2115699864,-0.3206691156,1.4883789524\H,0,2.5426579864,1.6072418844,-2.0591440476\H,0,3.2873999864,-4.4400921156,0.8690589524\H,0,2.7751569864,4.6674798844,0.9563809524\C,0,-1.2152300136,-1.0071871156,0.3462639524\C,0,-2.5699010136,-1.0365531156,0.6438949524\C,0,-3.4277350136,-0.1151661156,0.0395029524\C,0,-2.8970020136,0.8247948844,-0.8521080476\C,0,-1.5381490136,0.8195528844,-1.1070640476\H,0,-0.4999840136,-1.6995551156,0.7750879524\H,0,-2.9473820136,-1.7774381156,1.3417739524\H,0,-3.5337060136,1.5557138844,-1.3407470476\H,0,-1.0585630136,1.5337848844,-1.7690320476\Cl,0,-5.1017510136,-0.1324251156,0.3816289524\\Version=IA64-Linux-G03RevB.01\State=1-A\HF=-1204.2740106\MP2=-1207.0724461\RMDS=6.006e-09\PG=C01 [X(C18H15Cl1N1)]\ \@

### 3b

```
1\1\GINC-LXSRV166\SP\RMP2-FC\6-311++G(d,p)\C18H16N1(1+)\UI22103\11-Oct-2003\0\#\# MP2(FC)/6-311++G** SCF=TIGHT GEOM=CHECK GUESS=READ\SP Benzhydryl-Kation + Pyridin C1-symm.\1,1\H,0,-0.0231710229,0.0065659542,-1.8946959924\C,0,-0.0293960229,-0.0612670458,-0.7993289924\N,0,0.1334959771,1.3886649542,-0.3537179924\C,0,-1.3596280229,-0.6171120458,-0.3663989924\C,0,1.1819369771,-0.8335970458,-0.3414959924\C,0,-2.4909740229,-0.3288390458,-1.1455679924\C,0,1.7573479771,-0.6252350458,0.919696076\C,0,-3.7459800229,-0.7984450458,-0.7600679924\C,0,2.8417999771,-1.4039090458,1.3269810076\C,0,-3.8794350229,-1.5513760458,0.4101640076\C,0,3.3516239771,-2.3939000458,0.4831080076\C,0,-2.7542600229,-1.8429160458,1.1856350076\C,0,2.7783229771,-2.6023670458,-0.7736639924\C,0,-1.4945890229,-1.3768050458,0.8042940076\C,0,1.6974019771,-1.8231730458,-1.1875779924\H,0,-2.3866430229,0.2473389542,-2.0645789924\H,0,1.3716319771,0.1439419542,1.5874410076\H,0,-4.6166410229,-0.5846660458,-1.3743129924\H,0,3.2879549771,-1.2380580458,2.3041150076\H,0,-2.8554150229,-2.4427320458,2.0863070076\H,0,3.1758199771,-3.3677040458,-1.4348219924\H,0,-0.6186680229,-1.6327550458,1.3945190076\H,0,1.2496949771,-1.9902710458,-2.1660999924\H,0,-4.8559010229,-1.9241890458,0.7076290076\H,0,4.1964759771,-2.9978320458,0.8026270076\C,0,-0.6809070229,1.9470609542,0.5678790076\C,0,-0.4721290229,3.2611849542,0.9725770076\C,0,0.5725059771,4.0003179542,0.4213960076\C,0,1.3977249771,3.3951529542,-0.5292739924\C,0,1.1640909771,2.0799959542,-0.8988339924\H,0,-1.4789330229,1.3206419542,0.9486200076\H,0,-1.1361840229,3.6901099542,1.7161530076\H,0,0.7426639771,5.0291199542,0.7252710076\H,0,2.2230409771,3.9316709542,-0.9862309924\H,0,1.7770559771,1.5395359542,-1.6130359924\Version=IA64-Linux-G03RevB.01\State=1-A\HF=-745.3572877\MP2=-748.0165883\RMSD=8.481e-09\PG=C01 [X(C18H16N1)]\@
```

### 3c conformation 1

```
1\1\GINC-LX64I52\SP\RMP2-FC\6-311++G(d,p)\C19H18N1(1+)\UI22102\25-Sep-2005\0\#\# MP2(FC)/6-311++G** SCF=TIGHT GEOM=CHECK GUESS=READ\SP Benzhydryl-Kation + 4-Methylpyridin C1-symm. Var1\1,1\H,0,0.4349279928,0.022440144,-1.9275719712\C,0,0.4384689928,0.0055840144,-0.8300209712\N,0,-1.0332080072,-0.1297889856,-0.4676249712\C,0,1.2261829928,-1.1870119856,-0.3557139712\C,0,0.9279419928,1.3449650144,-0.3391559712\C,0,1.2141419928,-2.3520029856,-1.1382179712\C,0,0.5275859928,1.8697440144,0.
```

8973620288\C,0,1.9044449928,-3.4875529856,-0.7160619712\C,0,1.05201499  
28,3.0851400144,1.3403800288\C,0,2.6037259928,-3.4687369856,0.49421402  
88\C,0,1.9789529928,3.7772540144,0.5570070288\C,0,2.6201629928,-2.3091  
689856,1.2735480288\C,0,2.3788349928,3.2545490144,-0.6752959712\C,0,1.  
9320959928,-1.1684339856,0.8554100288\C,0,1.8539939928,2.0423470144,-1  
.1245259712\H,0,0.6793759928,-2.3642289856,-2.0875729712\H,0,-0.195799  
0072,1.3421670144,1.5174530288\H,0,1.9037249928,-4.3825559856,-1.33255  
99712\H,0,0.7373489928,3.4910780144,2.2983030288\H,0,3.1775369928,-2.2  
890309856,2.2063920288\H,0,3.0953949928,3.7928070144,-1.2899039712\H,0  
,1.9731679928,-0.2599169856,1.4504900288\H,0,2.1702309928,1.6349790144  
, -2.0836589712\H,0,3.1482259928,-4.3507579856,0.8206340288\H,0,2.38460  
59928,4.7236390144,0.9040570288\C,0,-1.4699700072,-1.0311829856,0.4378  
070288\C,0,-2.8175040072,-1.0880169856,0.7684850288\C,0,-3.7431870072,  
-0.2231089856,0.1752390288\C,0,-3.2434370072,0.6956200144,-0.760070971  
2\C,0,-1.8935010072,0.7334730144,-1.0604889712\H,0,-0.7206650072,-1.68  
52579856,0.8681430288\H,0,-3.1357870072,-1.8208579856,1.5047400288\H,0  
, -3.9057790072,1.3983240144,-1.2582569712\H,0,-1.4561890072,1.44168501  
44,-1.7572169712\C,0,-5.2070850072,-0.2954319856,0.4911450288\H,0,-5.7  
237090072,-0.9053309856,-0.2578999712\H,0,-5.3777010072,-0.7476939856,  
1.4703550288\H,0,-5.6596450072,0.6990680144,0.4811820288\\Version=IA64  
-Linux-G03RevB.01\State=1-A\HF=-784.4111504\MP2=-787.2226685\RMSE=8.37  
2e-09\PG=C01 [X(C19H18N1)]\@

### 3c conformation 2

1\1\GINC-LX64I50\SP\RMP2-FC\6-311++G(d,p)\C19H18N1(1+)\UI22102\30-Sep-  
2005\0\#\ MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ\SP Benzhy  
dryl-Kation + 4-Methylpyridin C1-symm. Var2\1,1\H,0,0.4383180504,0.0  
039719568,-1.92764\C,0,0.4399020504,0.0074879568,-0.830104\N,0,-1.0314  
549496,-0.1387960432,-0.469773\C,0,1.2363230504,-1.1787010432,-0.35438  
8\C,0,0.9178460504,1.3507219568,-0.338509\C,0,1.2378950504,-2.34249004  
32,-1.138795\C,0,0.5135020504,1.8707339568,0.898768\C,0,1.9365210504,-  
3.4725770432,-0.715634\C,0,1.0271240504,3.0904109568,1.342608\C,0,2.63  
06240504,-3.4495410432,0.497531\C,0,1.9471330504,3.7917269568,0.559214  
\C,0,2.6335500504,-2.2911320432,1.278774\C,0,2.3508910504,3.2738809568  
, -0.673863\C,0,1.9371280504,-1.1558690432,0.859635\C,0,1.8367980504,2.  
0573649568,-1.12392\H,0,0.7070790504,-2.3579530432,-2.090314\H,0,-0.20  
44969496,1.3357569568,1.518791\H,0,1.9462720504,-4.3665220432,-1.33358

7\H,0,0.7095260504,3.4925379568,2.301173\H,0,3.1869350504,-2.267557043  
2,2.21391\H,0,3.0620500504,3.8192649568,-1.288462\H,0,1.9678840504,-0.  
2480130432,1.45632\H,0,2.1560670504,1.6538199568,-2.083663\H,0,3.18159  
50504,-4.3272220432,0.82479\H,0,2.3444070504,4.7414259568,0.906902\C,0  
, -1.4646709496, -1.0506480432, 0.42668\C,0, -2.8131449496, -1.1208340432, 0  
.751411\C,0, -3.7441589496, -0.2630840432, 0.156411\C,0, -3.2474399496, 0.6  
691609568, -0.767234\C,0, -1.8966619496, 0.7205889568, -1.061369\H,0, -0.71  
22569496, -1.7031380432, 0.853936\H,0, -3.1293929496, -1.8665490432, 1.4754  
75\H,0, -3.9146639496, 1.3644589568, -1.269336\H,0, -1.4627109496, 1.435313  
9568, -1.753522\C,0, -5.2004639496, -0.3078910432, 0.510494\H,0, -5.4318239  
496, 0.4650679568, 1.251354\H,0, -5.8224089496, -0.1229780432, -0.368817\H,  
0, -5.4743759496, -1.2759750432, 0.934841\\Version=IA64-Linux-G03RevB.01\  
State=1-A\HF=-784.4111285\MP2=-787.2226311\RMSE=6.288e-09\PG=C01 [X(C1  
9H18N1)]\@

### 3d conformation 1

1\l\GINC-LX64I49\SP\RMP2-FC\6-311++G(d,p)\C19H18N1O1(1+)\UI22102\26-Se  
p-2005\0\#\ MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ\SP Ben  
zhydryl-Kation + 4-Methoxypyridin C1-symm. Var1\1,1\H,0,-0.863671034,  
0.0049479252,-1.9515070136\C,0,-0.789837034,0.0128339252,-0.8562550136  
\N,0,0.703433966,-0.0390740748,-0.5933630136\C,0,-1.384956034,1.302501  
9252,-0.3536550136\C,0,-1.419677034,-1.2437110748,-0.3087150136\C,0,-1  
.285966034,2.4464629252,-1.1604210136\C,0,-0.999515034,-1.8138100748,0  
.9007379864\C,0,-1.795482034,3.6656049252,-0.7147650136\C,0,-1.6500310  
34,-2.9427530748,1.4017819864\C,0,-2.398673034,3.7512719252,0.54332998  
64\C,0,-2.722197034,-3.5032700748,0.7032599864\C,0,-2.500859034,2.6133  
749252,1.3476979864\C,0,-3.141857034,-2.9355610748,-0.5023290136\C,0,-  
1.994229034,1.3895359252,0.9059449864\C,0,-2.492047034,-1.8097010748,-  
1.0089860136\H,0,-0.826499034,2.3783509252,-2.1461020136\H,0,-0.163016  
034,-1.3891040748,1.4537279864\H,0,-1.729586034,4.5450849252,-1.349868  
0136\H,0,-1.319519034,-3.3845080748,2.3383149864\H,0,-2.984778034,2.67  
59019252,2.3189179864\H,0,-3.971645034,-3.3722670748,-1.0516150136\H,0  
, -2.103451034, 0.5011229252, 1.5224959864\H,0, -2.823273034, -1.3669180748  
, -1.9471640136\H,0, -2.803133034, 4.6990829252, 0.8884599864\H,0, -3.22557  
1034, -4.3830210748, 1.0948389864\C,0, 1.321006966, 0.8149799252, 0.2492669  
864\C,0, 2.679981966, 0.7173319252, 0.5010409864\C,0, 3.428941966, -0.28569  
90748, -0.1343340136\C,0, 2.758681966, -1.1655630748, -1.0079450136\C,0, 1.

406568966,-1.0263430748,-1.2122860136\H,0,0.697156966,1.5741379252,0.7  
066489864\H,0,3.125511966,1.4284029252,1.1864739864\H,0,3.315073966,-1  
.9495440748,-1.5114520136\H,0,0.836288966,-1.6868000748,-1.8575190136\  
C,0,5.479093966,0.3804949252,0.8903769864\H,0,5.425113966,1.4123619252  
,0.5357529864\H,0,5.102350966,0.2967729252,1.9123769864\H,0,6.50286396  
6,0.0191289252,0.8376709864\O,0,4.736505966,-0.4921880748,0.0088259864  
\Version=IA64-Linux-G03RevB.01\State=1-A\HF=-859.2813681\MP2=-862.286  
2137\RMSD=9.251e-09\PG=C01 [X(C19H18N1O1)]\@\

### 3d conformation 2

1\GINC-LX64I48\SP\RMP2-FC\6-311++G(d,p)\C19H18N1O1(1+)\UI22102\28-Se  
p-2005\0\#\ MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ\SP Ben  
zhydryl-Kation + 4-Methoxy pyridin Cl-symm. Var2\1,1\H,0,0.7400329796,  
0.0169829796,-1.9286399592\C,0,0.7460849796,0.0212709796,-0.8310059592  
\N,0,-0.7068970204,-0.2108900204,-0.4583349592\C,0,1.6128489796,-1.118  
1440204,-0.3613879592\C,0,1.1572779796,1.3881269796,-0.3432129592\C,0,  
1.6687739796,-2.2835770204,-1.1410329592\C,0,0.7201899796,1.8952339796  
,0.8881150408\C,0,2.4324259796,-3.3722170204,-0.7215319592\C,0,1.17425  
19796,3.1392659796,1.3297220408\C,0,3.1379459796,-3.3062160204,0.48343  
30408\C,0,2.0670729796,3.8784089796,0.5498210408\C,0,3.0865489796,-2.1  
459370204,1.2600930408\C,0,2.5035449796,3.3735529796,-0.6775489592\C,0  
,2.3247779796,-1.0520520204,0.8443890408\C,0,2.0495809796,2.1323409796  
,-1.1246529592\H,0,1.1280339796,-2.3322720204,-2.0857849592\H,0,0.0230  
229796,1.3305779796,1.5054070408\H,0,2.4832589796,-4.2673310204,-1.335  
8069592\H,0,0.8316939796,3.5305869796,2.2842050408\H,0,3.6476959796,-2  
.0883480204,2.1891320408\H,0,3.1938699796,3.9478989796,-1.2895789592\H  
,0,2.3113989796,-0.1416010204,1.4378520408\H,0,2.3947559796,1.73894097  
96,-2.0796739592\H,0,3.7390739796,-4.1514140204,0.8079330408\H,0,2.417  
9339796,4.8469019796,0.8956800408\C,0,-1.0818680204,-1.1355720204,0.46  
06360408\C,0,-2.4085880204,-1.2827350204,0.8053450408\C,0,-3.389421020  
4,-0.4749680204,0.2030550408\C,0,-2.9746880204,0.4777139796,-0.7458969  
592\C,0,-1.6303660204,0.5863509796,-1.0445619592\H,0,-0.2901610204,-1.  
7380700204,0.8896870408\H,0,-2.7037120204,-2.0235400204,1.5414470408\H  
,0,-3.6703930204,1.1360609796,-1.2524539592\H,0,-1.2492120204,1.316976  
9796,-1.7513859592\C,0,-5.6959880204,0.1111569796,0.0027550408\H,0,-5.  
5428050204,1.1681139796,0.2325400408\H,0,-5.7389510204,-0.0538190204,-  
1.0762829592\H,0,-6.6090820204,-0.2454200204,0.4724460408\O,0,-4.64507

70204,-0.6883780204,0.5915540408\\Version=IA64-Linux-G03RevB.01\\State=1-A\\HF=-859.2810981\\MP2=-862.2862417\\RMSD=6.643e-09\\PG=C01 [X(C19H18N1O1)]\\@

### 3e

1\\1\\GINC-LX64I62\\SP\\RMP2-FC\\6-311++G(d,p)\\C18H17N2(1+)\\UI22102\\15-Sep-2005\\0\\# MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ\\SP Benzhydryl-Kation + 4-Aminopyridin C1-symm.\\1,1\\H,0,0.4229091007,0.0042029712,-1.932172\\C,0,0.4176391007,0.0077109712,-0.834283\\N,0,-1.0439238993,-0.1343090288,-0.472567\\C,0,1.2106241007,-1.1830980288,-0.358954\\C,0,0.9166371007,1.3444379712,-0.343332\\C,0,1.2048651007,-2.3488840288,-1.139725\\C,0,0.4908921007,1.8892809712,0.875539\\C,0,1.8986111007,-3.4815570288,-0.715144\\C,0,1.0244141007,3.0996469712,1.322191\\C,0,2.5949981007,-3.4594230288,0.496677\\C,0,1.9853841007,3.7676979712,0.559545\\C,0,2.6043591007,-2.2993320288,1.275208\\C,0,2.4106641007,3.2253519712,-0.655776\\C,0,1.9129471007,-1.1616710288,0.853927\\C,0,1.8777691007,2.0176799712,-1.107316\\H,0,0.6710541007,-2.3638210288,-2.089493\\H,0,-0.2599848993,1.3804099712,1.477913\\H,0,1.9025551007,-4.3771200288,-1.330933\\H,0,0.6899301007,3.5205829712,2.26693\\H,0,3.1586921007,-2.2763390288,2.209873\\H,0,3.1539091007,3.7447319712,-1.25486\\H,0,1.9470041007,-0.2527540288,1.449033\\H,0,2.2138801007,1.5948869712,-2.052992\\H,0,3.1422361007,-4.3390770288,0.825038\\H,0,2.3977331007,4.7103949712,0.908894\\C,0,-1.4887988993,-1.0313360288,0.43998\\C,0,-2.8219018993,-1.0994190288,0.78038\\C,0,-3.7599658993,-0.2312120288,0.17991\\C,0,-3.2601308993,0.6956769712,-0.765698\\C,0,-1.9185178993,0.7237919712,-1.059891\\H,0,-0.7424748993,-1.6857700288,0.874608\\H,0,-3.1345458993,-1.8344610288,1.516381\\H,0,-3.9226338993,1.3952029712,-1.267209\\H,0,-1.4918968993,1.4313109712,-1.763924\\N,0,-5.0681448993,-0.2793670288,0.49104\\H,0,-5.4276298993,-0.9469930288,1.160662\\H,0,-5.7391878993,0.3382829712,0.053512\\Version=IA64-Linux-G03RevB.01\\State=1-A\\HF=-800.4278436\\MP2=-803.2643414\\RMSD=3.869e-09\\PG=C01 [X(C18H17N2)]\\@

### 3f

1\\1\\GINC-IBMSMP\\SP\\RMP2-FC\\6-311++G(d,p)\\C20H21N2(1+)\\UI22102\\02-Sep-2004\\0\\# MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ\\SP Benzhydryl-Kation + DMAP C1-symm.\\1,1\\H,0,1.1293880581,0.0611140065,-1.9543310323\\C,0,1.0670570581,0.0305380065,-0.8584460323\\N,0,-0.4096529419,-

0.0733769935,-0.5732540323\C,0,1.7976600581,-1.2007349935,-0.3838650323\C,0,1.5891860581,1.3325300065,-0.3009200323\C,0,1.7848020581,-2.3447579935,-1.1957970323\C,0,1.1085890581,1.8664280065,0.9020299677\C,0,2.4205480581,-3.5120899935,-0.7743710323\C,0,1.6629720581,3.0410610065,1.4142609677\C,0,3.0657570581,-3.5465179935,0.4651819677\C,0,2.6992510581,3.6841140065,0.7333209677\C,0,3.0818210581,-2.4083369935,1.2751459677\C,0,3.1794580581,3.1525480065,-0.4662680323\C,0,2.4483050581,-1.2362449935,0.8569839677\C,0,2.6264700581,1.9805590065,-0.9827210323\H,0,1.2904750581,-2.3158619935,-2.1663570323\H,0,0.2984690581,1.3770630065,1.4401879677\H,0,2.4194090581,-4.3909389935,-1.4138640323\H,0,1.2857370581,3.4536550065,2.3465519677\H,0,3.5962320581,-2.4295209935,2.2324479677\H,0,3.9817770581,3.6526420065,-1.0023140323\H,0,2.4869750581,-0.3451109935,1.4782279677\H,0,3.0052960581,1.5658070065,-1.9157530323\H,0,3.5680350581,-4.4533889935,0.7914279677\H,0,3.1276380581,4.5991030065,1.1334449677\C,0,-0.9278939419,-0.9599559935,0.3118449677\C,0,-2.2749589419,-0.9902779935,0.5920769677\C,0,-3.1731819419,-0.0882319935,-0.0404710323\C,0,-2.5866619419,0.8290650065,-0.9583680323\C,0,-1.2336499419,0.8146860065,-1.1897010323\H,0,-0.2245079419,-1.6414839935,0.7757839677\H,0,-2.6190989419,-1.7258879935,1.3084099677\H,0,-3.1813089419,1.5591290065,-1.4933500323\H,0,-0.7555619419,1.5138330065,-1.8689530323\N,0,-4.4950499419,-0.0996649935,0.2094319677\C,0,-5.3811709419,0.8491600065,-0.4678160323\H,0,-5.3494429419,0.7076380065,-1.5523460323\H,0,-6.4001859419,0.6752250065,-0.1283520323\H,0,-5.1077269419,1.8805460065,-0.2255300323\C,0,-5.0533549419,-1.0627789935,1.1603259677\H,0,-4.8632909419,-2.0892429935,0.8323879677\H,0,-4.6317789419,-0.9147069935,2.1591899677\H,0,-6.1296329419,-0.9128839935,1.2152799677\Version=IBM64-RS6000-G03RevB.01\State=1-A\HF=-878.4978706\MP2=-881.6428677\RMSD=5.115e-09\PG=C01 [X(C20H21N2)]\@

### 3g conformation 1

1\1\GINC-LX64I67\SP\RMP2-FC\6-311++G(d,p)\C22H23N2(1+)\UI22102\26-Sep-2005\0\#\ MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ\SP Benzhydryl-Kation + 4-Pyrrolidinopyridin C1-symm. Var1\1,1\H,0,1.7714870651,0.0551239704,-1.953185\C,0,1.6702230651,0.0258919704,-0.860094\N,0,0.1853940651,-0.0676280296,-0.626028\C,0,2.3759060651,-1.2101920296,-0.36005\C,0,2.1845260651,1.3239659704,-0.285894\C,0,2.377750651,-2.3567750296,-1.168344\C,0,1.6613140651,1.8683279704,0.894249\C,0,2.99085906

51,-3.5279040296,-0.724531\C,0,2.2082710651,3.0383289704,1.424891\C,0,  
3.5985770651,-3.5635820296,0.533799\C,0,3.2793830651,3.6663809704,0.78  
5209\C,0,3.5998320651,-2.4229220296,1.340354\C,0,3.8021800651,3.124350  
9704,-0.391712\C,0,2.9888680651,-1.2470530296,0.899722\C,0,3.256959065  
1,1.9569039704,-0.926268\H,0,1.9124350651,-2.3270370296,-2.153091\H,0,  
0.8236250651,1.3908609704,1.399786\H,0,3.0012890651,-4.4089070296,-1.3  
60996\H,0,1.7978480651,3.4590559704,2.339378\H,0,4.0849830651,-2.44518  
70296,2.312803\H,0,4.6318940651,3.6127369704,-0.89594\H,0,3.0156770651  
, -0.3542930296,1.51929\H,0,3.6688920651,1.5338999704,-1.841413\H,0,4.0  
832270651,-4.4735180296,0.877727\H,0,3.7018570651,4.5778389704,1.19946  
8\C,0,-0.3675859349,-0.9409630296,0.253139\C,0,-1.7226549349,-0.960248  
0296,0.488316\C,0,-2.5906299349,-0.0564590296,-0.183102\C,0,-1.9704969  
349,0.8482419704,-1.091481\C,0,-0.6116969349,0.8201779704,-1.27965\H,0  
,0.3169010651,-1.6181950296,0.750456\H,0,-2.1014769349,-1.6726560296,1  
.211833\H,0,-2.5513119349,1.5636149704,-1.661976\H,0,-0.1061959349,1.5  
037889704,-1.954916\N,0,-3.9135829349,-0.0557450296,0.022689\C,0,-4.84  
19079349,0.9167849704,-0.582883\H,0,-4.9682159349,0.7017419704,-1.6518  
92\H,0,-4.4536389349,1.9340049704,-0.471899\C,0,-4.6120549349,-1.03055  
50296,0.880281\H,0,-4.2720689349,-2.0463800296,0.655702\H,0,-4.4084269  
349,-0.8161460296,1.937449\C,0,-6.0804039349,-0.8047670296,0.527199\H,  
0,-6.7472479349,-1.0906850296,1.343858\H,0,-6.3475249349,-1.3978630296  
, -0.353019\C,0,-6.1333999349,0.6871919704,0.198871\H,0,-6.1225659349,1  
.2799519704,1.118893\H,0,-7.0171739349,0.9710949704,-0.377069\\Version  
=IA64-Linux-G03RevB.01\State=1-A\HF=-955.4307506\MP2=-958.8581181\RMSE  
=9.973e-09\PG=C01 [X(C22H23N2)]\@

### 3g conformation 2

1\1\GINC-LX64I67\SP\RMP2-FC\6-311++G(d,p)\C22H23N2(1+)\UI22102\23-Sep-  
2005\0\#\ MP2(FC)/6-311++G\*\* SCF=TIGHT GEOM=CHECK GUESS=READ\SP Benzh  
ydryl-Kation + 4-Pyrrolidinopyridin C1-symm. Var2\1,1\H,0,1.769195964  
5,0.0615791302,-1.9547600118\C,0,1.6691659645,0.0288511302,-0.86170301  
18\N,0,0.1845669645,-0.0648858698,-0.6259580118\C,0,2.3752609645,-1.20  
89538698,-0.3665510118\C,0,2.1846039645,1.3248631302,-0.2837740118\C,0  
,2.3776289645,-2.3522938698,-1.1794280118\C,0,1.6632999645,1.865318130  
2,0.8990129882\C,0,2.9910689645,-3.5249868698,-0.7402380118\C,0,2.2113  
429645,3.0333671302,1.4328169882\C,0,3.5986759645,-3.5654728698,0.5179  
969882\C,0,3.2816859645,3.6633121302,0.7937049882\C,0,3.5994359645,-2.



4280468698,1.3291159882\C,0,3.8025869645,3.1251771302,-0.3858360118\C,  
0,2.9880759645,-1.2506618698,0.8931499882\C,0,3.2562449645,1.959713130  
2,-0.9235800118\H,0,1.9123259645,-2.3187958698,-2.1640700118\H,0,0.826  
1969645,1.3863691302,1.4041129882\H,0,3.0018589645,-4.4034328698,-1.38  
02190118\H,0,1.8023539645,3.4511121302,2.3493109882\H,0,4.0844979645,-  
2.4540378698,2.3015169882\H,0,4.6316729645,3.6150731302,-0.8896320118\  
H,0,3.0144999645,-0.3603858698,1.5162939882\H,0,3.6666909645,1.5397401  
302,-1.8407850118\H,0,4.0836139645,-4.4766068698,0.8583309882\H,0,3.70  
50049645,4.5732471302,1.2104429882\C,0,-0.3678340355,-0.9419678698,0.2  
498669882\C,0,-1.7221820355,-0.9592158698,0.4892889882\C,0,-2.59109703  
55,-0.0549888698,-0.1803660118\C,0,-1.9719750355,0.8514721302,-1.08765  
60118\C,0,-0.6129560355,0.8261981302,-1.2746180118\H,0,0.3156559645,-1  
.6283028698,0.7359339882\H,0,-2.1042550355,-1.6954958698,1.1867149882\  
H,0,-2.5494150355,1.5902681302,-1.6310980118\H,0,-0.1066990355,1.51986  
01302,-1.9389680118\N,0,-3.9137520355,-0.0544608698,0.0272919882\C,0,-  
4.8644810355,0.7971361302,-0.7110710118\C,0,-4.5895820355,-0.910874869  
8,1.0189549882\C,0,-5.9958880355,-0.3188758698,1.0789779882\C,0,-6.218  
2830355,0.1936971302,-0.3438110118\H,0,-4.7771360355,1.8385541302,-0.3  
752250118\H,0,-4.6584160355,0.7551291302,-1.7850940118\H,0,-4.06768603  
55,-0.8647468698,1.9799009882\H,0,-4.5989760355,-1.9524738698,0.672468  
9882\H,0,-6.0205470355,0.5120061302,1.7909829882\H,0,-6.7381880355,-1.  
0563848698,1.3927089882\H,0,-7.0266250355,0.9247971302,-0.4175950118\H  
,0,-6.4501250355,-0.6382398698,-1.0161400118\\Version=IA64-Linux-G03Re  
vB.01\State=1-A\HF=-955.4307029\MP2=-958.8577907\RMSD=3.553e-09\PG=C01  
[X(C22H23N2)]\@

## References

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- [S2] S. Minegishi, S. Kobayashi, H. Mayr, *J. Am. Chem. Soc.* **2004**, *126*, 5174–5181.
- [S3] Calculated from the reactivity parameters  $E$ ,  $N$ , and  $s$  from ref. [S1].